ACCESS DB# 129047

## SEARCH REQUEST FORM

### Scientific and Technical Information Center

	imber $30 2 - 18$	Examiner #: 6933 J. I. Serial Number: 10 G. Its Format Preferred (circle)	49 873
If more than one search is submit			
Please provide a detailed statement of the se Include the elected species or structures, key utility of the invention. Define any terms th known. Please attach a copy of the cover she	arch topic, and describe a words, synonyms, acrony at may have a special mea	s specifically as possible the subjections, and registry numbers, and containing. Give examples or relevant cabstract.	et matter to be searched.  The searched or the concept or the search of
Title of Invention:		Si	CIENTIFIC REFERENCE BR
Inventors (please provide full names):			Sci & Tech. Injo. Cntr
inventors (piease provide fun names).			AUG ZI
Earliest Priority Filing Date:			Pat. & T.M. Office
*For Sequence Searches Only* Please include	all pertinent information (p	arent, child, divisional, or issued pate	nt numbers) along with the
appropriate serial number.  To pula of claim	7. Ehad	l Ka i	
-			
			,
•			
		•	
STAFF USE ONLY	Type of Search	Vendors and cost when	e applicable
Searcher: A. Tulker	NA Sequence (#)	STN	
Searcher Phone #:	AA Sequence (#)	Dialog	
Searcher Location:	Structure (#)	Questel/Orbit	
Date Searcher Picked Up:	Bibliographic	Dr.Link	· · · · · · · · · · · · · · · · · · ·
Date Completed: 8/10/64	Litigation	Lexis/Nexis	<del> </del>
Searcher Prep & Review Time:	Fulltext	Sequence Systems	
Clerical Prep Time:	Patent Family	WWW/Internet	
Online Time: 4	Other	Other (specify)	



# STIC Search Report

# STIC Database Tracking Number: 129047

TO: Duc Truong

Location: REM6D41

Art Unit: 1711 August 10, 2004 10 DM

Case Serial Number: 10/642873

From: Kathleen Fuller Location: EIC 1700

REMSEN 4B28

Phone: 571/272-2505

Kathleen.Fuller@uspto.gov

Search Notes		
	•	





		A CONTRACT	HH-2017	200
525	1 7 . V	E _ 2	' - Y - '	Y _ 1
-		ET I	FIE.	Æ
<b>~</b>		P.S. B.		
11 ]	L A	<b>Æ.</b>		

Questions about the scope or the results of the search? Contact the EIC searcher or contact:

Kathleen Fuller, EIC 1700 Team Leader 571/272-2505 REMSEN 4B28

Voluntary Results Feedback Form
<ul> <li>I am an examiner in Workgroup: Example: 1713</li> <li>Relevant prior art found, search results used as follows:</li> </ul>
102 rejection
103 rejection
Cited as being of interest.
Helped examiner better understand the invention.
Helped examiner better understand the state of the art in their technology.
Types of relevant prior art found:
Foreign Patent(s)
<ul> <li>☐ Non-Patent Literature         (journal articles, conference proceedings, new product announcements etc.)</li> </ul>
<ul> <li>Results verified the lack of relevant prior art (helped determine patentability).</li> <li>Results were not useful in determining patentability or understanding the invention.</li> </ul>
Comments:

Drop off or send completed forms to EIC1700 REMSEN 4B28



#### TRUONG 10/642873 8/10/04 Page 1

=> FILE REG

FILE 'REGISTRY' ENTERED AT 17:38:25 ON 10 AUG 2004
USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT.
PLEASE SEE "HELP USAGETERMS" FOR DETAILS.
COPYRIGHT (C) 2004 American Chemical Society (ACS)

Property values tagged with IC are from the ZIC/VINITI data file provided by InfoChem.

STRUCTURE FILE UPDATES: 9 AUG 2004 HIGHEST RN 724701-07-9 DICTIONARY FILE UPDATES: 9 AUG 2004 HIGHEST RN 724701-07-9

TSCA INFORMATION NOW CURRENT THROUGH MAY 21, 2004

Please note that search-term pricing does apply when conducting SmartSELECT searches.

Crossover limits have been increased. See HELP CROSSOVER for details.

Experimental and calculated property data are now available. For more information enter HELP PROP at an arrow prompt in the file or refer to the file summary sheet on the web at: http://www.cas.org/ONLINE/DBSS/registryss.html

#### => FILE HCAPLU

FILE 'HCAPLUS' ENTERED AT 17:38:30 ON 10 AUG 2004
USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT.
PLEASE SEE "HELP USAGETERMS" FOR DETAILS.
COPYRIGHT (C) 2004 AMERICAN CHEMICAL SOCIETY (ACS)

Copyright of the articles to which records in this database refer is held by the publishers listed in the PUBLISHER (PB) field (available for records published or updated in Chemical Abstracts after December 26, 1996), unless otherwise indicated in the original publications. The CA Lexicon is the copyrighted intellectual property of the the American Chemical Society and is provided to assist you in searching databases on STN. Any dissemination, distribution, copying, or storing of this information, without the prior written consent of CAS, is strictly prohibited.

FILE COVERS 1907 - 10 Aug 2004 VOL 141 ISS 7 FILE LAST UPDATED: 9 Aug 2004 (20040809/ED)

This file contains CAS Registry Numbers for easy and accurate substance identification.

=> D	QUE	
L50	134516	SEA FILE=REGISTRY ABB=ON 46.492.16/RID
L51	16503	SEA FILE=REGISTRY ABB=ON L50 AND PMS/CI
L52	174	SEA FILE=REGISTRY ABB=ON CARDANOL
L53	3	SEA FILE=REGISTRY ABB=ON LAURIC ACID/CN OR MYRISTIC ACID/CN
		OR PALMITIC ACID/CN
L54	7	SEA FILE=REGISTRY ABB=ON STEARIC ACID/CN OR ARACHIDIC ACID/CN
		OR PALMITOLEIC ACID/CN OR OLEIC ACID/CN OR RICINOLEIC ACID/CN
		OR LINOLEIC ACID/CN OR ARACHIDONIC ACID/CN
L55	4	SEA FILE=REGISTRY ABB=ON DODECYL MERCAPTAN/CN OR PHENYL
		MERCAPTAN/CN OR LAURYL THIOGLYCOLATE/CN OR OCTYL THIOGLYCOLATE/

```
CN
          27181 SEA FILE=HCAPLUS ABB=ON L51
L56
L57
         127435 SEA FILE=HCAPLUS ABB=ON L52 OR L53 OR L54 OR L55
L60
           7482 SEA FILE=HCAPLUS ABB=ON L56(L)COPOLYMER?
L62
             10 SEA FILE=HCAPLUS ABB=ON
                                        L56(L)L57
             41 SEA FILE=HCAPLUS ABB=ON
                                        L56(L) FATTY ACID#(L) COPOLYMER?
L65
            836 SEA FILE=HCAPLUS ABB=ON
                                        L57 (L) COPOLYMER?
L66
            10 SEA FILE=HCAPLUS ABB=ON
                                        L60 AND L66
L67
             56 SEA FILE=HCAPLUS ABB=ON L62 OR L65 OR L67
L68
            16 SEA FILE=HCAPLUS ABB=ON L68 AND (PREP OR IMF OR SPN OR
L69
                POF)/RL
           7564 SEA FILE=REGISTRY ABB=ON 554-63-8/CRN OR 143-07-7/CRN OR
L70
                57-10-3/CRN OR 506-32-1/CRN OR 506-30-9/CRN OR 373-49-9/CRN OR
                141-22-0/CRN OR 112-80-1/CRN
           5141 SEA FILE=REGISTRY ABB=ON 60-33-3/CRN OR 57-11-4/CRN
L71
             35 SEA FILE=REGISTRY ABB=ON L51 AND (L70 OR L71)
L72
             1 SEA FILE=REGISTRY ABB=ON L51 AND L52
L73
L74
            22 SEA FILE=HCAPLUS ABB=ON L72 OR L73
L75
            12 SEA FILE=HCAPLUS ABB=ON L74(L) (POF OR PREP OR IMF OR SPN)/RL
L76
            27 SEA FILE=HCAPLUS ABB=ON L69 OR L75
=> D L76 ALL 1-27 HITSTR
L76 ANSWER 1 OF 27 HCAPLUS COPYRIGHT 2004 ACS on STN
     2004:307542 HCAPLUS
AN
DN
     140:329578
ED
     Entered STN: 15 Apr 2004
    Multicolor pressure-sensitive thermal recording material containing fatty
TI
     acid monoamide
     Sasaki, Takashi; Date, Takashi; Kubota, Yukio; Suzuki, Minoru
IN
PΑ
     Nippon Paper Industries, Co., Ltd., Japan; Pentax Corporation
```

- Jpn. Kokai Tokkyo Koho, 16 pp. SO CODEN: JKXXAF
- DT Patent
- Japanese T.A
- IC ICM B41M005-26
  - ICS B41M005-155; B41M005-165; B41M005-28; B41M005-30
- 74-6 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)

#### FAN.CNT 1

PATEN	NT NO.	KIND DA	TE	APPLICATION	NO.	DATE
	004114349 002-277579		040415 020924	JP 2002-2775	79	20020924
PATENT NO	CLASS	PATENT FAM	ILY CLASSI	FICATION COD	ES	
JP 200411	14349 ICM ICS	B41M005-26 B41M005-15		-165; B41M00	5-28; B41M0	005-30
JP 200411	14349 FTERM	2H026/BB33 2H026/DD53	; 2H026/BB ; 2H026/FF	11; 2H026/BB 37; 2H026/DD 03; 2H026/FF 29; 2H085/BB	02; 2H026/E	DD43; AA07;

- OS MARPAT 140:329578
- AB The material has ≥1 recording layer containing (1) the colorless or pale colored electron donative dye precursor in which  $\geq 1$  dye is microencapsulated with a shell fractured by applying higher temperature and

2H085/DD02; 2H085/DD43

Ţ

pressure than decided ones, (2) the electron attractive developer containing (poly) 4-hydroxy benzoic acid H[OCO(p-C6H4)]mOCO(p-C6H4)OH (I; m = 0-2) ora condensation reaction product of (a) a carboxylic acid component containing I and/or its derivative and (b) a polyvalent alc. component containing  $\geq 3$ valent alc. (HOCH2)2R1CCH2[OCH2R2C(CH2OH)CH2]nOH as an essential ingredient, and (3) a saturated fatty acid monoamide ROCONH2 with average particle diameter  $\leq 1 \mu m$ . The material shows improved color discrimination and heat and humidity resistance of background. ST heat pressure sensitive multicolor printing material; fatty acid monoamide printing material; polyester hydroxybenzoic acid alc color developer printing; microcapsule dye precursor printing material IT Polyesters, preparation RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (color developer; multicolor pressure-sensitive thermal recording material containing fatty acid monoamide) TT Aminoplasts RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (microcapsule shell; multicolor pressure-sensitive thermal recording material containing fatty acid monoamide) TΤ Printing (impact) Thermal printing materials (multicolor pressure-sensitive thermal recording material containing fatty acid monoamide) 8063-22-7 TΤ RL: MOA (Modifier or additive use); TEM (Technical or engineered material use); USES (Uses) (G 270; multicolor pressure-sensitive thermal recording material containing fatty acid monoamide) 247038-95-5P, Hexapentaerythritol-4-hydroxybenzoic acid-terephthalic acid ΤТ copolymer RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (color developer; multicolor pressure-sensitive thermal recording material containing fatty acid monoamide) IT 99-96-7, 4-Hydroxybenzoic acid, uses 104880-73-1 RL: TEM (Technical or engineered material use); USES (Uses) (color developer; multicolor pressure-sensitive thermal recording material containing fatty acid monoamide) ΙT 9003-08-1P, Formaldehyde-melamine copolymer RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (microcapsule shell; multicolor pressure-sensitive thermal recording material containing fatty acid monoamide) ΙT 124-26-5, Stearic acid amide 629-54-9, Palmitic acid amide RL: MOA (Modifier or additive use); TEM (Technical or engineered material use); USES (Uses) (multicolor pressure-sensitive thermal recording material containing fatty acid monoamide) ΙT 40677-31-4 RL: TEM (Technical or engineered material use); USES (Uses) (multicolor pressure-sensitive thermal recording material containing fatty acid monoamide) IT 9003-08-1P, Formaldehyde-melamine copolymer RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(microcapsule shell; multicolor pressure-sensitive thermal recording

material containing fatty acid monoamide)

RN 9003-08-1 HCAPLUS

CN 1,3,5-Triazine-2,4,6-triamine, polymer with formaldehyde (9CI) (CA INDEX NAME)

CM 1

CRN 108-78-1 CMF C3 H6 N6

2 CM

CRN 50-00-0 CMF C H2 O

 $H_2C = 0$ 

L76 ANSWER 2 OF 27 HCAPLUS COPYRIGHT 2004 ACS on STN

2004:305328 HCAPLUS AN

DN 140:329568

Entered STN: 15 Apr 2004 ED

Thermal or pressure-sensitive printing material containing fatty acid TΙ monoamide

IN Sasaki, Takashi; Date, Takashi

PΑ Nippon Paper Industries, Co., Ltd., Japan

Jpn. Kokai Tokkyo Koho, 12 pp. SO

CODEN: JKXXAF

DT Patent

LA Japanese

ICM B41M005-26 IC ICS B41M005-28

CC 74-6 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)

FAN.CNT 1

APPLICATION NO. PATENT NO. KIND DATE DATE \_\_\_\_\_\_ -----\_\_\_\_ 20040415 JP 2002-277574 20020924 JP 2004114348 A2 PΙ PRAI JP 2002-277574 20020924 CLASS

CLASS PATENT FAMILY CLASSIFICATION CODES PATENT NO. JP 2004114348 ICM B41M005-26

ICS

JP 2004114348 FTERM 2H026/AA07; 2H026/DD12; 2H026/DD34; 2H026/DD43; 2H026/DD48; 2H026/FF05

MARPAT 140:329568 OS

B41M005-28

```
The material contains (A) an electron donative leuco dye microencapsulated
AB
     or forming complex fine particles with polyvalent isocyanate polymer, (B)
     an electron attractive developer, and (C) a saturated fatty acid monoamide
     R1CONH2 (R1 = C11-21 alkyl). It shows improved color development and
     abrasion resistance.
ST
     thermal printing material microencapsulated leuco dye; complex fine
     particle leuco dye isocyanate polymer printing; fatty acid monoamide
     thermal printing material
TΤ
     Aminoplasts
     RL: IMF (Industrial manufacture); TEM (Technical or engineered
     material use); PREP (Preparation); USES (Uses)
        (microcapsule shell; thermal printing material containing fatty acid
        monoamide)
IT
     Printing (impact)
        (pressure-sensitive printing material containing fatty acid monoamide)
ΙT
     Thermal printing materials
        (thermal printing material containing fatty acid monoamide)
IT
     8063-22-7
     RL: MOA (Modifier or additive use); TEM (Technical or engineered material
     use); USES (Uses)
        (G 270; printing material containing fatty acid monoamide)
     51852-81-4P, Trimethylolpropane-xylylene diisocyanate copolymer
IT
     148130-89-6P, Takenate D 110N homopolymer
     RL: IMF (Industrial manufacture); TEM (Technical or engineered
     material use); PREP (Preparation); USES (Uses)
        (microcapsule shell; printing material containing fatty acid monoamide)
TΤ
     9003-08-1P, Formaldehyde-melamine copolymer
     RL: IMF (Industrial manufacture); TEM (Technical or engineered
     material use); PREP (Preparation); USES (Uses)
        (microcapsule shell; thermal printing material containing fatty
        acid monoamide)
     124-26-5, Stearic acid amide
                                    629-54-9, Palmitic acid amide
TΤ
     RL: MOA (Modifier or additive use); TEM (Technical or engineered material
     use); USES (Uses)
        (printing material containing fatty acid monoamide)
ΙT
     70516-41-5, S 205
     RL: TEM (Technical or engineered material use); USES (Uses)
        (printing material containing fatty acid monoamide)
IT
     89331-94-2, ODB 2
     RL: TEM (Technical or engineered material use); USES (Uses)
        (thermal printing material containing fatty acid monoamide)
TΤ
     9003-08-1P, Formaldehyde-melamine copolymer
     RL: IMF (Industrial manufacture); TEM (Technical or engineered
     material use); PREP (Preparation); USES (Uses)
        (microcapsule shell; thermal printing material containing fatty
        acid monoamide)
RN
     9003-08-1 HCAPLUS
     1,3,5-Triazine-2,4,6-triamine, polymer with formaldehyde (9CI) (CA INDEX
CN
     NAME)
     CM
          1
```

KATHLEEN FULLER EIC 1700 REMSEN 4B28 571/272-2505

CRN 108-78-1 CMF C3 H6 N6

CM 2

CRN 50-00-0 CMF C H2 O

 $H_2C = 0$ 

```
L76 ANSWER 3 OF 27 HCAPLUS COPYRIGHT 2004 ACS on STN
```

AN 2003:892465 HCAPLUS

DN 139:366426

ED Entered STN: 14 Nov 2003

TI Adhesion promoting agent and coating compositions for polymeric substrates

IN Kondos, Constantine A.; Martz, Jonathan T.; Nakajima, Masayuki

PA USA

SO U.S. Pat. Appl. Publ., 17 pp., Cont. of U.S. Ser. No. 564,174.

CODEN: USXXCO

DT Patent

LA English

IC ICM C08F008-00

NCL 525191000

CC 42-10 (Coatings, Inks, and Related Products)

Section cross-reference(s): 38

FAN.CNT 1

PATENT NO.

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI US 2003212209	A1	20031113	US 2003-404289	20030401
PRAI US 2000-564174	A1	20000503	•	
CLASS				

CLASS PATENT FAMILY CLASSIFICATION CODES

US 2003212209 ICM C08F008-00

ICM C08F008-00 NCL 525191000

US 2003212209 ECLA G11B027/031; H04N007/16E3; H04N007/167D; H04N007/26Q; H04N007/30; H04N009/804B; H04N005/913

AB The present invention provides an adhesion promoting agent based on a graft copolymer formed by a halogenated polyolefin polymer with at least one reactive functionality and a substantially saturated hydrocarbon polymer having more than one reactive functionality. At least one reactive functionality of the saturated hydrocarbon polymer is reactive with the reactive functionality of the halogenated polyolefin to form the graft copolymer of the invention. A typical graft copolymer was manufactured by stirring 200 g Hardlen 13MLJ (chlorinated maleated polyolefin) with 120 g Polytail H (hydrogenated polybutadienediol) 6 h at 120°.

ST plastic substrate coating halogenated polyolefin adduct; chlorinated maleated polyolefin hydrogenated polybutadienediol adduct coating plastic

substrate

IT Alkyd resins

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(acrylic, thermosetting binder; grafted products of reactive hydrocarbon polymers and halogenated polyolefins for adhesion-promoting binders for coatings for plastics)

IT Polyolefins

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(chlorinated, maleated, reaction products, with hydrogenated polybutadienediols; grafted products of reactive hydrocarbon polymers and halogenated polyolefins for adhesion-promoting binders for coatings for plastics)

IT Epoxy resins, uses

Polyesters, uses

Polyurethanes, uses

RL: POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses)

(curable binder; grafted products of reactive hydrocarbon polymers and halogenated polyolefins for adhesion-promoting binders for coatings for plastics)

IT Adhesion promoters

(grafted products of reactive hydrocarbon polymers and halogenated polyolefins for adhesion-promoting binders for coatings for plastics)

IT Styrene-butadiene rubber, uses

RL: POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses)

(hydrogenated, block, diblock, Kraton G1726X, addnl. binder; grafted products of reactive hydrocarbon polymers and halogenated polyolefins for adhesion-promoting binders for coatings for plastics)

IT Polyolefins

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(maleated, chlorinated, reaction products, with hydrogenated polybutadienediols; grafted products of reactive hydrocarbon polymers and halogenated polyolefins for adhesion-promoting binders for coatings for plastics)

IT Plastics, miscellaneous

RL: MSC (Miscellaneous)

(substrate; grafted products of reactive hydrocarbon polymers and halogenated polyolefins for adhesion-promoting binders for coatings for plastics)

IT Polyolefins

RL: MSC (Miscellaneous)

(thermoplastic, substrate; grafted products of reactive hydrocarbon polymers and halogenated polyolefins for adhesion-promoting binders for coatings for plastics)

IT Coating materials

(thermosetting; grafted products of reactive hydrocarbon polymers and halogenated polyolefins for adhesion-promoting binders for coatings for plastics)

IT 68855-15-2P, Pentaerythritol-phthalic anhydride copolymer ester with Emersol 315 and crotonic acid

RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)

(binder precursor; grafted products of reactive hydrocarbon polymers and halogenated polyolefins for adhesion-promoting binders for coatings for plastics)

IT 620626-01-9P

RL: CPS (Chemical process); IMF (Industrial manufacture); PEP (Physical, engineering or chemical process); PREP (Preparation); PROC (Process) (curable binder precursor; grafted products of reactive hydrocarbon polymers and halogenated polyolefins for adhesion-promoting binders for coatings for plastics)

#### IT 620626-02-0P

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(cured binder; grafted products of reactive hydrocarbon polymers and halogenated polyolefins for adhesion-promoting binders for coatings for plastics)

IT 108-31-6DP, Maleic anhydride, reaction products with chlorinated polyolefins and hydrogenated polybutadienediols 9003-17-2DP, Polybutadiene, diols, hydrogenated, reaction products with chlorinated maleated polyolefins 87913-10-8DP, Polytail H, reaction products with chlorinated maleated polyolefins 88507-04-4DP, Polytail HA, reaction products with chlorinated maleated polyolefins 217818-36-5DP, Hardlen CY9122P, reaction products with hydrogenated polybutadienediols 620963-25-9DP, Hardlen 13MLJ, reaction products with hydrogenated polybutadienediols

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(grafted products of reactive hydrocarbon polymers and halogenated polyolefins for adhesion-promoting binders for coatings for plastics) 9003-55-8

RL: POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses)

(styrene-butadiene rubber, hydrogenated, block, diblock, Kraton G1726X, addnl. binder; grafted products of reactive hydrocarbon polymers and halogenated polyolefins for adhesion-promoting binders for coatings for plastics)

IT 201687-78-7, ETA 3183 221680-44-0, Dexflex 777 459124-89-1, Sequel 1440 620963-29-3, D 161B 620963-30-6, Dexflex 727 620963-31-7, CA 287 620963-44-2, Amtuff 3110

RL: MSC (Miscellaneous)

(substrate; grafted products of reactive hydrocarbon polymers and halogenated polyolefins for adhesion-promoting binders for coatings for plastics)

#### IT 620626-02-0P

IT

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(cured binder; grafted products of reactive hydrocarbon polymers and halogenated polyolefins for adhesion-promoting binders for coatings for plastics)

#### RN 620626-02-0 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, methyl ester, polymer with 2,2-bis(hydroxymethyl)-1,3-propanediol polymer with 1,3-isobenzofurandione 2-butenoate (9Z,12Z)-9,12-octadecadienoate, 2-ethylhexyl 2-propenoate, formaldehyde, 2-propenenitrile and 1,3,5-triazine-2,4,6-triamine (9CI) (CA INDEX NAME)

CM 1

CRN 108-78-1 CMF C3 H6 N6

CM 2

CRN 107-13-1 CMF C3 H3 N

$$H_2C = CH - C = N$$

CM 3

CRN 103-11-7 CMF C11 H20 O2

$$\begin{array}{c} \text{O} \\ \parallel \\ \text{CH}_2\text{--O-C-CH} \Longrightarrow \text{CH}_2 \\ \parallel \\ \text{Et-CH-Bu-n} \end{array}$$

CM 4

CRN 80-62-6 CMF C5 H8 O2

$$\begin{array}{c|c} ^{H_2C} & \text{O} \\ \parallel & \parallel \\ \text{Me-} \text{C-} \text{C-} \text{OMe} \end{array}$$

CM 5

CRN 50-00-0 CMF C H2 O

 $H_2C = 0$ 

CM 6

CRN 68855-15-2

• TRUONG 10/642873 8/10/04 Page 10

CMF C18 H32 O2 . x (C8 H4 O3 . C5 H12 O4)x . x C4 H6 O2

CM 7

CRN 3724-65-0 CMF C4 H6 O2

 $Me-CH-CO_2H$ 

,

CM 8

CRN 60-33-3 CMF C18 H32 O2

Double bond geometry as shown.

$$(CH_2)$$
 7  $Z$   $(CH_2)$  4  $Me$ 

CM 9

CRN 26659-15-4

CMF (C8 H4 O3 . C5 H12 O4)  $\times$ 

CCI PMS

CM 10

CRN 115-77-5 CMF C5 H12 O4

$$\begin{array}{c} \text{CH}_2-\text{OH} \\ | \\ \text{HO-CH}_2-\text{C-CH}_2-\text{OH} \\ | \\ \text{CH}_2-\text{OH} \end{array}$$

CM 11

CRN 85-44-9 CMF C8 H4 O3

```
L76
    ANSWER 4 OF 27 HCAPLUS COPYRIGHT 2004 ACS on STN
     2003:851431 HCAPLUS
AN
     140:425001
DN
     Entered STN: 31 Oct 2003
ED
ΤI
     Development of self-catalysis low-temperature-curable quick-drying amino
     paints
ΑU
     Wang, Xiaomu
     Northwest Yongxin Chemical Industry Co., Ltd., Lanzhou, 730020, Peop. Rep.
CS
     Xiandai Tuliao Yu Tuzhuang (2003), (5), 1-2
SO
     CODEN: XTYTAL; ISSN: 1007-9548
PB
     Xiandai Tuliao Yu Tuzhuang Bianjibu
DT
     Journal
LA
     Chinese
     42-8 (Coatings, Inks, and Related Products)
CC
AB
     An enamel contained an alkyd resin 48-65, an amino resin iso-Bu ether
     18-24, TiO2 20-22, a dispersant 0.5-1, solvents 1-3%, and a trace of
     viscosity stabilizer. An alkyd resin was prepared from oleic acid 14-18,
     phthalic anhydride 18-22, benzoic acid 1-3, trimellitic anhydride 1-2,
     2-methyl-1,3-propylene glycol 4-8, trimethylolpropane 8-14, xylene 2-4, an
     antioxidant 0.004, and solvents 44-47%.
     alkyd amino resin enamel
ST
TΤ
     Enamels (paints)
        (alkyd resins and amino resins for self-catalysis low-temperature-curable
        quick-drying enamels)
TΤ
     Alkyd resins
     Aminoplasts
     RL: POF (Polymer in formulation); SPN (Synthetic preparation); TEM
     (Technical or engineered material use); PREP (Preparation); USES (Uses)
        (alkyd resins and amino resins for self-catalysis low-temperature-curable
        quick-drying enamels)
TΨ
     691004-83-8P
     RL: POF (Polymer in formulation); PRP (Properties); SPN
     (Synthetic preparation); TEM (Technical or engineered material use);
     PREP (Preparation); USES (Uses)
        (alkyd resins and amino resins for self-catalysis low-temperature-curable
        quick-drying enamels)
     691004-82-7P, 2-Methyl-1,3-propylene glycol-oleic acid-phthalic
TΤ
     anhydride-trimellitic anhydride-trimethylolpropane copolymer
     RL: POF (Polymer in formulation); SPN (Synthetic preparation); TEM
     (Technical or engineered material use); PREP (Preparation); USES (Uses)
        (alkyd resins and amino resins for self-catalysis low-temperature-curable
        quick-drying enamels)
TΤ
     691004-83-8P
     RL: POF (Polymer in formulation); PRP (Properties); SPN
     (Synthetic preparation); TEM (Technical or engineered material use);
     PREP (Preparation); USES (Uses)
        (alkyd resins and amino resins for self-catalysis low-temperature-curable
        quick-drying enamels)
RN
     691004-83-8 HCAPLUS
     5-Isobenzofurancarboxylic acid, 1,3-dihydro-1,3-dioxo-, polymer with
CN
     2-ethyl-2-(hydroxymethyl)-1,3-propanediol, formaldehyde,
     1,3-isobenzofurandione, 2-methyl-1,3-propanediol, (9Z)-9-octadecenoic acid
     and 1,3,5-triazine-2,4,6-triamine (9CI) (CA INDEX NAME)
    CM
          1
    CRN 2163-42-0
```

CMF C4 H10 O2

$$\begin{array}{c} \text{Me} \\ | \\ \text{HO-CH}_2\text{--CH-CH}_2\text{--OH} \end{array}$$

CM 2

CRN 552-30-7 CMF C9 H4 O5

CM 3

CRN 112-80-1 CMF C18 H34 O2

Double bond geometry as shown.

$$HO_2C$$
 (CH<sub>2</sub>) 7  $Z$  (CH<sub>2</sub>) 7  $Me$ 

CM 4

CRN 108-78-1 CMF C3 H6 N6

CM 5

CRN 85-44-9 CMF C8 H4 O3

CM 6

CRN 77-99-6 CMF C6 H14 O3

$$\begin{array}{c} \text{CH}_2-\text{OH} \\ | \\ \text{HO-CH}_2-\text{C-Et} \\ | \\ \text{CH}_2-\text{OH} \end{array}$$

CM 7

CRN 50-00-0 CMF C H2 O

 $H_2C = 0$ 

```
ΑN
     2002:126241 HCAPLUS
DN
     136:185435
     Entered STN: 19 Feb 2002
ED
ΤI
     Process for coating of ferromagnetic articles
     Yono, Masayoshi; Watanabe, Mikio
IN
     Shin-Etsu Chemical Industry Co., Ltd., Japan
PΑ
     Jpn. Kokai Tokkyo Koho, 12 pp.
SO
     CODEN: JKXXAF
DT
     Patent
LA
     Japanese
     ICM B05D001-02
IC
     ICS B05D007-14; B05D007-24; C23F011-00; F16B033-06
CC
     42-2 (Coatings, Inks, and Related Products)
     Section cross-reference(s): 77
FAN.CNT 1
     PATENT NO.
                                            APPLICATION NO.
                         KIND
                                DATE
```

L76 ANSWER 5 OF 27 HCAPLUS COPYRIGHT 2004 ACS on STN

PI JP 20020523	62	A2	20020219	JP 2000-243158	20000810
PRAI JP 2000-243 CLASS	158		20000810		
	~~ ~ ~ ~			T. T	
PATENT NO.	CLASS	PATENT	FAMILY CLASS	IFICATION CODES	
JP 2002052362	ICM	B05D001	L-02		
	ICS	B05D007	7-14; B05D007	-24; C23F011-00; F16B0	33-06

DATE

```
Ferromagnetic articles especially having screw portions are coated uniformly by
     spraying coatings while the articles are fixed and rotated by a magnet.
     Thus, p-aminophenol was treated with HCl to give a condensate, which was
     further treated with Li stearate to give a product. A composition containing
the
     product 12.4, Epokey 863 (liquid epoxy resin solution) 100, U-Van 22R
     (butylated melamine resin) 50, and poly(vinyl butyral) 10 parts was
     sprayed on a rotating bolt to form a coating showing good corrosion
     resistance and no peeling or defects by screwing it into and out from a
ST
     anticorrosive polyaminophenol coating spraying ferromagnetic article; bolt
     coating spray aminophenol polymer
ΙT
     Coating materials
        (anticorrosive; spray coating of ferromagnetic articles having screw
        portions)
IT
     Polyamines
     RL: IMF (Industrial manufacture); PEP (Physical, engineering or
     chemical process); POF (Polymer in formulation); PRP
     (Properties); PYP (Physical process); TEM (Technical or engineered
     material use); PREP (Preparation); PROC (Process); USES (Uses)
        (aromatic; spray coating of ferromagnetic articles having screw portions)
IT
     Bolts
     Ferromagnetic materials
        (spray coating of ferromagnetic articles having screw portions)
IT
     Epoxy resins, uses
     Polyvinyl butyrals
     RL: PEP (Physical, engineering or chemical process); POF (Polymer in
     formulation); PRP (Properties); PYP (Physical process); TEM
     (Technical or engineered material use); PROC (Process); USES (Uses)
        (spray coating of ferromagnetic articles having screw portions)
ΙT
     Coating process
        (spray; spray coating of ferromagnetic articles having screw portions)
     100-44-7DP, Benzyl chloride, reaction products with aromatic amine
ΙT
     condensates 112-80-1DP, Oleic acid, reaction products with
     aniline-nitrobenzene copolymer 4485-12-5DP, Lithium stearate,
     reaction products with aromatic amine polymers 25668-00-2DP, p-Aminophenol
     homopolymer, reaction products with lithium stearate
                                                            25668-00-2P,
                                32036-19-4DP, Poly(imino-1,4-phenylene),
     p-Aminophenol homopolymer
     reaction products with lithium stearate
                                               32036-19-4P,
     Poly(imino-1,4-phenylene) 51774-88-0P, m-Phenylenediamine-resorcinol
                 63519-68-6P, Poly(imino-1,3-phenylene)
                                                         186376-45-4P,
     p-Benzoquinone-1,8-diaminonaphthalene-pyrogallol copolymer
     333788-70-8DP, Aniline-nitrobenzene copolymer, reaction products with
                  333788-70-8P, Aniline-nitrobenzene copolymer
     RL: IMF (Industrial manufacture); PEP (Physical, engineering or
     chemical process); POF (Polymer in formulation); PRP
     (Properties); PYP (Physical process); TEM (Technical or engineered
    material use); PREP (Preparation); PROC (Process); USES (Uses)
        (spray coating of ferromagnetic articles having screw portions)
ΤТ
     398488-46-5, Epokey 863-U-Van 22R copolymer
     RL: PEP (Physical, engineering or chemical process); POF (Polymer in
     formulation); PRP (Properties); PYP (Physical process); TEM
     (Technical or engineered material use); PROC (Process); USES (Uses)
        (spray coating of ferromagnetic articles having screw portions)
IT
     112-80-1DP, Oleic acid, reaction products with
     aniline-nitrobenzene copolymer
     RL: IMF (Industrial manufacture); PEP (Physical, engineering or
     chemical process); POF (Polymer in formulation); PRP
     (Properties); PYP (Physical process); TEM (Technical or engineered
```

TRUONG 10/642873 8/10/04 Page 15

material use); PREP (Preparation); PROC (Process); USES (Uses)
 (spray coating of ferromagnetic articles having screw portions)

RN 112-80-1 HCAPLUS

CN 9-Octadecenoic acid (9Z)- (9CI) (CA INDEX NAME)

Double bond geometry as shown.

IT 398488-46-5, Epokey 863-U-Van 22R copolymer

RL: PEP (Physical, engineering or chemical process); POF (Polymer in formulation); PRP (Properties); PYP (Physical process); TEM (Technical or engineered material use); PROC (Process); USES (Uses) (spray coating of ferromagnetic articles having screw portions)

RN 398488-46-5 HCAPLUS

CN Formaldehyde, polymer with Epokey 863 and 1,3,5-triazine-2,4,6-triamine (9CI) (CA INDEX NAME)

CM 1

CRN 211058-94-5 CMF Unspecified CCI PMS, MAN

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

CM 2

CRN 108-78-1 CMF C3 H6 N6

CM 3

CRN 50-00-0 CMF C H2 O

 $H_2C = 0$ 

L76 ANSWER 6 OF 27 HCAPLUS COPYRIGHT 2004 ACS on STN

AN 2001:816802 HCAPLUS

DN 135:359218

ED Entered STN: 09 Nov 2001

TI Adhesion promoting agent and coating compositions for polymeric substrates

```
Kondos, Constantine A.; Martz, Jonathan T.; Nakajima, Masayuki
IN
      PPG Industries Ohio, Inc., USA
PA
      PCT Int. Appl., 53 pp.
SO
      CODEN: PIXXD2
DT
      Patent
LA
      English
      ICM C09J
IC
CC
      42-10 (Coatings, Inks, and Related Products)
      Section cross-reference(s): 38
FAN.CNT 1
                                                 APPLICATION NO.
      PATENT NO.
                            KIND
                                     DATE
                                                                             DATE
      _____
                                     -----
                                                   _____
                            ----
                                                                              _____
     WO 2001083628
                                                  WO 2001-US10940
                             A2
                                     20011108
                                                                             20010404
PΙ
      WO 2001083628
                             А3
                                     20020627
          W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CR, CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM

RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG
                                                US 2000-564175
                                     20030715
                                                                               20000503
      US 6593423
                              В1
      AU 2001051297
                              A5
                                                   AU 2001-51297
                                                                               20010404
                                     20011112
PRAI US 2000-564175
                              Α
                                      20000503
     WO 2001-US10940
                              W
                                     20010404
CLASS
                 CLASS PATENT FAMILY CLASSIFICATION CODES
 PATENT NO.
 ______
 WO 2001083628 ICM
                            C09J
                  ECLA
                           C08J007/04L51; C09D123/28; C09D151/06
 US 6593423
     The present invention provides an adhesion promoting agent which includes
AB
      a graft copolymer formed by a halogenated polyolefin polymer with at least
      one reactive functionality and a substantially saturated hydrocarbon polymer
      having more than one reactive functionality. At least one reactive
      functionality of the saturated hydrocarbon polymer is reactive with the
      reactive functionality of the halogenated polyolefin to form the graft
      copolymer of the invention.
      adhesion promotor coating compn polymer substrate; polyolefin halogenated
ST
     hydrocarbon graft polymer adhesion promotor
IT
     Adhesion promoters
      Coating materials
     Crosslinking agents
         (adhesion promoting agent and coating compns. for polymeric substrates)
IT
     Acrylic polymers, uses
     Aminoplasts
      Polyesters, uses
      Polyurethanes, uses
     RL: POF (Polymer in formulation); TEM (Technical or engineered material
     use); USES (Uses)
         (adhesion promoting agent and coating compns. for polymeric substrates)
TΨ
     Epoxides
      Polyolefins
      RL: TEM (Technical or engineered material use); USES (Uses)
         (adhesion promoting agent and coating compns. for polymeric substrates)
TΤ
      Polymers, uses
     RL: IMF (Industrial manufacture); POF (Polymer in formulation); TEM
      (Technical or engineered material use); PREP (Preparation); USES (Uses)
```

(graft; adhesion promoting agent and coating compns. for polymeric substrates) IT Polyolefins RL: IMF (Industrial manufacture); POF (Polymer in formulation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (halogenated; adhesion promoting agent and coating compns. for polymeric substrates) ΙT 371968-38-6P, Hardlen MLJ 13-Polytail H graft copolymer 371968-39-7P, Hardlen MLJ 13-Polytail HA graft copolymer 371968-40-0P, acrylonitrile-2-ethylhexyl acrylate-crotonic acid-Emersol 315-Pentaerythritol-phthalic anhydride-methyl methacrylate-Resimene 717-styrene copolymer 372150-82-8P, Hardlen CY 9122P-Polytail H graft 372150-83-9P, eponex 1510-Hardlen CY 9122P-Polytail H graft copolymer copolymer RL: IMF (Industrial manufacture); POF (Polymer in formulation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (adhesion promoting agent and coating compns. for polymeric substrates) 371968-40-0P, acrylonitrile-2-ethylhexyl acrylate-crotonic IT acid-Emersol 315-Pentaerythritol-phthalic anhydride-methyl methacrylate-Resimene 717-styrene copolymer RL: IMF (Industrial manufacture); POF (Polymer in formulation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (adhesion promoting agent and coating compns. for polymeric substrates) RN 371968-40-0 HCAPLUS 9,12-Octadecadienoic acid (9Z,12Z)-, polymer with 2,2-bis(hydroxymethyl)-CN 1,3-propanediol, 2-butenoic acid, ethenylbenzene, 2-ethylhexyl 2-propenoate, formaldehyde, 1,3-isobenzofurandione, methyl 2-methyl-2-propenoate, 2-propenenitrile and 1,3,5-triazine-2,4,6-triamine (9CI) (CA INDEX NAME) CM 1 CRN 3724-65-0 CMF C4 H6 O2 Me-CH-CO2H 2 CM CRN 115-77-5 CMF C5 H12 O4 CH2-OH

CM 3

CRN 108-78-1

CMF C3 H6 N6

CM 4

CRN 107-13-1 CMF C3 H3 N

$$H_2C = CH - C = N$$

CM 5

CRN 103-11-7 CMF C11 H20 O2

$$CH_2 - O - C - CH = CH_2$$
 $CH_2 - O - C - CH = CH_2$ 
 $CH_2 - O - C - CH = CH_2$ 

CM 6

CRN 100-42-5 CMF C8 H8

$$H_2C = CH - Ph$$

CM 7

CRN 85-44-9 CMF C8 H4 O3

```
ÇM
          8
          80-62-6
          C5 H8 O2
     CM
 H<sub>2</sub>C
Me-C-C-OMe
     CM
     CRN
         60-33-3
     CMF C18 N32 02
Double bond geometry as shown.
HO<sub>2</sub>C (CH<sub>2</sub>) 7
                             (CH<sub>2</sub>)<sub>4</sub>
         10
     CM
     CRN 50-00-0
     CMF C H2 O
H_2C = 0
L76 ANSWER 7 OF 27 HCAPLUS COPYRIGHT 2004 ACS on STN
     2001:224477 HCAPLUS
AN
DN
     134:223470
     Entered STN: 30 Mar 2001
ED
     Epoxy resins exhibiting higher tenacity after hardening and their
ΤI
     manufacture
     Lunak, Stanislav; Dobas, Ivan; Sima, Milan
ΙN
PΑ
     Synpo A. S., Czech Rep.
     Czech Rep., 5 pp.
SO
     CODEN: CZXXED
DT
     Patent
LA
     Czech
IC
     ICM C08G059-16
CC
     37-6 (Plastics Manufacture and Processing)
FAN.CNT 1
     PATENT NO.
                                             APPLICATION NO.
                         KIND
                                DATE
                                                                    DATE
                                             -----
                         ----
    CZ 285979
                          В6
                                 19991215
                                            CZ 1989-1836
                                                                    19890324
PΙ
PRAI CS 1989-1836
                          Α
                                 19890324
CLASS
                CLASS PATENT FAMILY CLASSIFICATION CODES
 PATENT NO.
```

C08G059-16 CZ 285979 ICM The title resins, useful for coatings, sealants, adhesives, prepregs, AB laminates and molded articles, comprise 10-90% products obtained by reacting epoxy resins with polymerized fatty acids from unsatd. vegetable The reaction is carried out in 2 steps where, in the 1st step, 10-90% of the total amount of the resin reacts with polymerized fatty acids to obtain an intermediate product with acid number <10 mg KOH/g and, in the 2nd step, the remaining epoxy resin is added and the reaction completed. A typical title resin, curable with diaminodiphenylmethane, was prepared by heating 100 parts bisphenol A-based epoxy resin (0.4 epoxy equiv/100 g) with 20 parts polymerized rape oil fatty acids (90% dimers, 8% monomers, 2% tri- and higher oligomers) for 2 h at 150° in the presence of 0.1% benzyl(lauryl)trimethylammonium bromide. epoxy resin manuf fatty acid oligomer deriv high tenacity; fatty acid STunsatd dimer epoxy resin deriv high tenacity TΤ Fatty acids, preparation RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (castor-oil, polymerized, reaction products with epoxy resins, crosslinked; manufacture of epoxy resins with higher tenacity after hardening) TΨ Fatty acids, preparation RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (dimer acids, reaction products, with epoxy resins, crosslinked; manufacture of epoxy resins with higher tenacity after hardening) TΤ Fatty acids, preparation RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (linseed-oil, polymerized, reaction products with epoxy resins, crosslinked; manufacture of epoxy resins with higher tenacity after hardening) TΤ Fatty acids, preparation RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (rape-oil, polymerized, reaction products with epoxy resins, crosslinked; manufacture of epoxy resins with higher tenacity after hardening) TΤ Aminoplasts RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (reaction products with epoxy resin adducts with polymerized unsatd. fatty acids, crosslinked; manufacture of epoxy resins with higher tenacity after hardening) ፐጥ Epoxy resins, preparation RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (reaction products, with polymerized unsatd. fatty acids, crosslinked; manufacture of epoxy resins with higher tenacity after hardening) ΙT Fatty acids, preparation RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (soya, polymerized, reaction products with epoxy resins, crosslinked; manufacture of epoxy resins with higher tenacity after hardening) TΤ Fatty acids, preparation RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(sunflower-oil, polymerized, reaction products with epoxy resins, crosslinked; manufacture of epoxy resins with higher tenacity after

Fatty acids, preparation IT

hardening)

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(unsatd., dimers, polymers, with epoxy resins, crosslinked; manufacture of epoxy resins with higher tenacity after hardening)

9003-08-1DP, Formaldehyde-Melamine copolymer, reaction products with epoxy resin adducts with polymerized unsatd. fatty acids 25068-38-6DP, Bisphenol A-Epichlorohydrin copolymer, reaction products with polymerized unsatd. fatty acids and diaminodiphenylmethane 32144-31-3DP, Diglycidylaniline, polymers, reaction products with polymerized unsatd. fatty acids and hexahydrophthalic anhydride 58421-55-9DP, Bisphenol F-epichlorohydrin copolymer, reaction products with polymerized unsatd. fatty acids and melamine resin RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(crosslinked; manufacture of epoxy resins with higher tenacity after hardening)

9003-08-1DP, Formaldehyde-Melamine copolymer, reaction products with epoxy resin adducts with polymerized unsatd. fatty acids

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(crosslinked; manufacture of epoxy resins with higher tenacity after hardening)

RN 9003-08-1 HCAPLUS

CN 1,3,5-Triazine-2,4,6-triamine, polymer with formaldehyde (9CI) (CA INDEX NAME)

CM 1

ΙT

ΙT

CRN 108-78-1 CMF C3 H6 N6

CM 2

CRN 50-00-0 CMF C H2 O

H2C==O

L76 ANSWER 8 OF 27 HCAPLUS COPYRIGHT 2004 ACS on STN

AN 2001:137644 HCAPLUS

DN 134:326846

ED Entered STN: 25 Feb 2001

TI Hyperbranched poly(ether ketone) analogues with heterocyclic triazine moiety: synthesis and peripheral functionalization

AU Cho, Song Yun; Chang, Youngkyu; Kim, Jin Seok; Lee, Sang Cheon; Kim,

Chulhee

CS Department of Polymer Science and Engineering, Hyperstructured Organic Materials Research Center, Inha University, Inchon, 402-751, S. Korea

SO Macromolecular Chemistry and Physics (2001), 202(2), 263-269 CODEN: MCHPES; ISSN: 1022-1352

PB Wiley-VCH Verlag GmbH

DT Journal

LA English

CC 35-5 (Chemistry of Synthetic High Polymers)

AB Hyperbranched poly(ether ketone) with 1,3,5-s-triazine moiety was prepared by a one-pot polymerization of an AB2 type monomer,

2,4-bis(4-hydroxyphenyl)-6-(4-

(4-(4-fluorobenzoyl)phenoxy)phenyl)-1,3,5-s-triazine, which was synthesized from cyanuric chloride. The selective reactivity of three chlorine atoms on cyanuric chloride toward nucleophiles provides a very efficient route for the systematic synthesis of AB2 type triazine monomers and their hyperbranched polymers. The resulting polymers exhibited a glass transition at 264° without any indication of crystallinity. The modification of the peripheral hydroxyl groups on the hyperbranched polymers by methoxy, oligo oxyethylene, or stearyl moieties brought about remarkable changes in their solubility and glass transition temps. The amphiphilic nature of the 2-[2-(2-(2-methoxyethoxy)ethoxy)ethoxy]ethoxy-terminated poly(ether ketone) analog in an aqueous phase was investigated by using fluorescence techniques and dynamic light scattering. It was found that the analog forms a self-aggregation at a critical aggregation concentration of

12.6 mg/L. The mean diameter of the aggregates was 320 nm. The steady-state fluorescence anisotropy value (r) of 1,6-diphenyl-1,3,5-hexatriene (DPH) in the hydrophobic domain was 0.240.

ST hyperbranched polyether polyketone deriv contg triazine prepn; fluorescence aggregation glass transition dendrimer polyether polyketone

Polyketones RL: PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(polyether-; preparation and characterization of hyperbranched poly(ether ketone) analogs with heterocyclic triazine moiety)

IT Polyethers, preparation

RL: PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(polyketone-; preparation and characterization of hyperbranched poly(ether ketone) analogs with heterocyclic triazine moiety)

IT Fluorescence

ΙT

Glass transition temperature

Molecular association

(preparation and characterization of hyperbranched poly(ether ketone) analogs with heterocyclic triazine moiety)

IT Dendritic polymers

RL: PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(preparation and characterization of hyperbranched poly(ether ketone) analogs with heterocyclic triazine moiety)

IT 108-77-0, Cyanuric chloride 111-77-3, Diethylene glycol monomethyl ether 403-43-0, 4-Fluorobenzoyl chloride 9004-74-4, Polyethylene glycol monomethyl ether 13139-86-1, 4-Methoxyphenyl magnesium bromide 21473-02-9, 4-Phenoxyphenylmagnesium bromide

RL: RCT (Reactant); RACT (Reactant or reagent)

(in preparation of hyperbranched poly(ether ketone) analogs with heterocyclic triazine moiety)

IT 318481-72-0P, 2,4-Dichloro-6-(4-phenoxyphenyl)-1,3,5-s-triazine

```
319491-73-1P, 2,4-Bis(4-methoxyphenyl)-6-(4-phenoxyphenyl)-1,3,5-s-
     triazine 336611-79-1P
                               336611-80-4P
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (in preparation of hyperbranched poly(ether ketone) analogs with
        heterocyclic triazine moiety)
ፐጥ
     336627-21-5P
                    336627-22-6P
                                  336627-23-7P 336627-24-8P
     RL: PRP (Properties); SPN (Synthetic preparation); PREP
     (Preparation)
        (preparation and peripheral functionalization of hyperbranched poly(ether
        ketone) analogs with heterocyclic triazine moiety)
ΙT
     336611-81-5P
     RL: PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP
     (Preparation); RACT (Reactant or reagent)
        (preparation of hyperbranched poly(ether ketone) analogs with heterocyclic
        triazine moiety)
              THERE ARE 39 CITED REFERENCES AVAILABLE FOR THIS RECORD
RE.CNT
        39
RE
(1) Armarego, W; Purification of Laboratiry Chemical, 4th edition 1996
(2) Chu, F; Polym Bull 1993, V30, P265 HCAPLUS
(3) Fink, R; Chem Mater 1998, V10, P3620 HCAPLUS
(4) Fink, R; Macromolecules 1997, V30, P8177 HCAPLUS
(5) Flory, P; J Am Chem Soc 1952, V74, P2718 HCAPLUS
(6) Frechet, J; Science 1994, V263, P1710 HCAPLUS
(7) Garcia, M; Anal Chem 1999, V71, P256 HCAPLUS
(8) Harada, A; Macromolecules 1998, V31, P288 HCAPLUS
(9) Hawker, C; Macromolecules 1996, V29, P4370 HCAPLUS
(10) Ihre, H; Macromolecules 1998, V31, P4061 HCAPLUS
(11) Johansson, M; Trands Polym Sci 1996, V4, P398 HCAPLUS
(12) Kavanov, A; Macromolecules 1995, V28, P2303
(13) Kim, C; Macromolecules 1996, V29, P6353 HCAPLUS
(14) Kim, Y; J Am Chem Soc 1990, V112, P4592 HCAPLUS
(15) Kim, Y; J Am Chem Soc 1992, V114, P4947 HCAPLUS
(16) Kim, Y; J Polym Sci, Part A: Polym Chem 1998, V36, P1685 HCAPLUS
(17) Kim, Y; Macromolecules 1992, V25, P5561 HCAPLUS
(18) Kondo, S; J Macromol Sci Chem 1990, VA27, P1513
(19) Kwon, G; Langmuir 1993, V9, P945 HCAPLUS
(20) Lackowski, W; J Am Chem Soc 1999, V121, P1419 HCAPLUS
(21) Lee, S; Macromolecules 1999, V32, P1847 HCAPLUS
(22) Malmsrom, E; J Macromol Sci, Rev Macromol Chem Phys 1997, V37, P555
(23) McGlade, M; Macromolecules 1987, V20, P1782 HCAPLUS
(24) Morikawa, A; Macromolecules 1998, V31, P5999 HCAPLUS
(25) Mueller, A; Macromolecules 1998, V31, P776 HCAPLUS
(26) Nagasaki, Y; Macromolecules 1998, V31, P1473 HCAPLUS
(27) Newkome, G; Advances in dendritic macromolecules 1995, V1 & 2
(28) Newkome, G; Dendritic Molecules: Concepts, Syntheses, Perspectives 1996
(29) Ringsdorf, H; Macromolecules 1991, V24, P1678 HCAPLUS
(30) Shah, P; Eur Polym J 1984, V20, P519 HCAPLUS
(31) Shu, C; Macromolecules 1999, V32, P100 HCAPLUS
(32) Stutz, H; J Polym Sci, Part B: Polym Phys 1995, V33, P333 HCAPLUS
(33) Thurmond, K; J Am Chem Soc 1996, V118, P7239 HCAPLUS
(34) Uhrich, K; Macromolecules 1992, V25, P4583 HCAPLUS
(35) Voit, B; Acta Polym 1995, V46, P87 HCAPLUS
(36) Weimer, M; J Polym Sci, Part A: Polym Chem 1998, V36, P955 HCAPLUS
(37) Wilhelm, M; Macromolecules 1991, V24, P1033 HCAPLUS
(38) Wooley, K; Polymer J 1994, V26, P187 HCAPLUS
(39) Zeng, F; Chem Rev 1997, V97, P1681 HCAPLUS
ΙT
     336627-24-8P
     RL: PRP (Properties); SPN (Synthetic preparation); PREP
```

#### (Preparation)

(preparation and peripheral functionalization of hyperbranched poly(ether ketone) analogs with heterocyclic triazine moiety)

RN 336627-24-8 HCAPLUS

CN Methanone, [4-[4-[4,6-bis(4-hydroxyphenyl)-1,3,5-triazin-2-yl]phenoxy]phenyl](4-fluorophenyl)-, homopolymer, octadecanoate (ester) (9CI) (CA INDEX NAME)

CM 1

CRN 57-11-4 CMF C18 H36 O2 2

 ${\rm HO_2C^-}$  (CH<sub>2</sub>)<sub>16</sub>-Me

CM 2

CRN 336611-81-5

CMF (C34 H22 F N3 O4)x

CCI PMS

CM 3

CRN 336611-80-4 CMF C34 H22 F N3 O4

L76 ANSWER 9 OF 27 HCAPLUS COPYRIGHT 2004 ACS on STN

AN 2000:19402 HCAPLUS

DN 132:65420

ED Entered STN: 10 Jan 2000

TI Viscosity control agents for aqueous dispersions

IN Nakamura, Hiroshi; Date, Kazuyuki

PA Toyota Central Research and Development Laboratories, Inc., Japan

SO Jpn. Kokai Tokkyo Koho, 8 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM C09K003-00

ICS C09D201-00

CC 42-5 (Coatings, Inks, and Related Products)

FAN.CNT 1 APPLICATION NO. DATE PATENT NO. KIND DATE \_\_\_\_\_ \_\_\_\_ JP 2000001662 A2 20000107 JP 1998-167437 19980615 PRAI JP 1998-167437 19980615 CLASS PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES \_\_\_\_\_ \_\_\_\_\_\_ JP 2000001662 ICM C09K003-00 ICS C09D201-00 Viscosity control agents having hydrophobic and hydrophilic portions, useful for water-thinned coatings, are prepared by hydrophobization of hydrophilic polymers and/or hydrophilization of hydrophobic polymers. Thus, acrylic acid 96, 2-hydroxyethyl acrylate 173, and Me acrylate 116 parts were polymerized in ethylene glycol mono-Me ether in the presence of AIBN to a polymer, 100 parts of which was treated with 15 parts stearoyl chloride in DMF in the presence of Et3N to give a hydrophobized polymer (I). A 25% aqueous dispersion of dimethylethanolamine-neutralized I showed pseudoplastic flow. hydrophobized acrylic resin viscosity controller; water thinned coating viscosity controller Polyoxyalkylenes, uses ΙT RL: IMF (Industrial manufacture); MOA (Modifier or additive use); PRP (Properties); PREP (Preparation); USES (Uses) (acrylic, graft; amphiphilic viscosity control agents for aqueous dispersions) ΙT Amphiphiles (amphiphilic viscosity control agents for aqueous dispersions) ΙT Viscosity (control agent; amphiphilic viscosity control agents for aqueous dispersions) ΙT Coating materials (dispersion, water-thinned; amphiphilic viscosity control agents for aqueous dispersions) 124-30-1DP, Stearylamine, reaction products with acrylic acid-hydroxyethyl IT acrylate-Me acrylate copolymer 61386-05-8DP, Acrylic acid-2-hydroxyethyl acrylate-methyl acrylate copolymer, reaction products with stearylamine 253328-31-3P, Acryloyl chloride-butyl methacrylate-methyl methacrylate-polyethylene glycol graft copolymer 253328-32-4P, Acryloyl chloride-butyl methacrylate-methyl methacrylate-oxirane graft copolymer 253328-87-9P, Acrylic acid-2-hydroxyethyl acrylate-methyl acrylate 253328-88-0P, Acrylic acid-2-hydroxyethyl copolymer stearate acrylate-methyl acrylate copolymer ester with 1,2-epoxyoctadecane RL: IMF (Industrial manufacture); MOA (Modifier or additive use); PRP (Properties); PREP (Preparation); USES (Uses) (amphiphilic viscosity control agents for aqueous dispersions) IT 253328-34-6P, Acryloyl chloride-butyl methacrylate-Cymel 303-methyl methacrylate-polyethylene glycol copolymer dimethylethanolamine salt 253328-90-4P, Acrylic acid-2-hydroxyethyl acrylate-methyl acrylate copolymer stearate, polymer with Cymel 303, dimethylethanolamine salt 253328-92-6P, Acrylic acid-2-hydroxyethyl acrylate-methyl acrylate copolymer ester with 1,2-epoxyoctadecane, polymer with Cymel 303, dimethylethanolamine salt RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (coating; amphiphilic viscosity control agents for aqueous dispersions) IT 253328-90-4P, Acrylic acid-2-hydroxyethyl acrylate-methyl acrylate

copolymer stearate, polymer with Cymel 303, dimethylethanolamine salt

RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (coating; amphiphilic viscosity control agents for aqueous dispersions) 253328-90-4 HCAPLUS RN2-Propenoic acid, polymer with 2-hydroxyethyl 2-propenoate and methyl CN 2-propenoate, octadecanoate, polymer with formaldehyde and 1,3,5-triazine-2,4,6-triamine, compd. with 2-(dimethylamino)ethanol (9CI) (CA INDEX NAME) CM 1 CRN 108-01-0 CMF C4 H11 N O  $Me_2N-CH_2-CH_2-OH$ 2 CM CRN 253328-89-1 (C18 H36 O2 . x (C5 H8 O3 . C4 H6 O2 . C3 H4 O2)x . C3 H6 N6 . C H2 CMF 0)xCCI PMS CM 3 CRN 108-78-1 C3 H6 N6 CMF NH<sub>2</sub> NH2 H<sub>2</sub>N CM 4 CRN 50-00-0 CMF C H2 O  $H_2C = 0$ 5 CM 253328-87-9 CRN CMF C18 H36 O2 . x (C5 H8 O3 . C4 H6 O2 . C3 H4 O2)x CM 6

TRUONG 10/642873 8/10/04 Page 27

CRN 57-11-4 CMF C18 H36 O2

 ${\rm HO_2C^-}$  (CH<sub>2</sub>)<sub>16</sub>-Me

CM 7

CRN 61386-05-8

CMF (C5 H8 O3 . C4 H6 O2 . C3 H4 O2)x

CCI PMS

CM 8

CRN 818-61-1 CMF C5 H8 O3

CM 9

CRN 96-33-3 CMF C4 H6 O2

CM 10

CRN 79-10-7 CMF C3 H4 O2

L76 ANSWER 10 OF 27 HCAPLUS COPYRIGHT 2004 ACS on STN

AN 1999:407763 HCAPLUS

DN 131:33029

ED Entered STN: 02 Jul 1999

TI Crosslinkable copolymers of vinyl silane and vinyl esters of branched fatty acid for coating compositions

IN Huybrechts, Josef Theresu

PA E. I. Du Pont de Nemours & Co., USA

SO Can. Pat. Appl., 22 pp.

CODEN: CPXXEB DTPatent LA English ICM C08F230-08 IC ICS C09D004-00; C08F218-10 42-10 (Coatings, Inks, and Related Products) Section cross-reference(s): 37 FAN.CNT 1 PATENT NO. KIND DATE APPLICATION NO. DATE
PI CA 2230166 AA 19980825 CA 1998-2230166 19980223
US 5886125 A 19990323 US 1997-805546 19970225
PRAI US 1997-805546 19970225 CLASS CLASS PATENT FAMILY CLASSIFICATION CODES PATENT NO. \_\_\_\_\_ CA 2230166 ICM C08F230-08 ICS C09D004-00; C08F218-10 Composition comprises a copolymer having weight average mol. weight <40,000, OH value 20-160 and acid value <20 and derived from (A) 5-95 parts vinyl silane CH2:CHSi(Rlm)R23-m(R1 = aryl, C1-10 alkyl; R2 = hydrolyzable group), (B) 5-95 parts vinyl ester of a branched fatty acid CH2:CHOCOC(R3)(R4)R5 (R3, R4, R5 = H or C1-12 alkyl group, wherein the total of R3, R4 and R5contain  $\geq 3$  C), and (C) 0-90 parts olefinically unsatd. compound The compns. have good balance of solvent and chemical resistance, hardness, flexibility and adherence to a variety of substrates, and particularly, are useful in automotive top coating compns. Thus, 55.72 parts VeoVa 9 (Neononanoic acid, ethenyl ester) was reacted with Silquest A 171 (vinyltrimethoxysilane) 15.92 and 2-hydroxyethyl methacrylate 7.96 in Solvesso 100 and n-butanol to give a resin having solids 80.3%, acid value 3.6 mgKOH/g. Mn 3000 and Mw 24,300. vinyl silane copolymer curable coating automotive; fatty acid branched ST vinyl ester copolymer; polysiloxane polyester curable coating ITPolysiloxanes, uses Polysiloxanes, uses RL: IMF (Industrial manufacture); POF (Polymer in formulation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (aminoplast-, polyester-; crosslinkable copolymers of vinyl silane and vinyl esters of branched fatty acid for coating of) TΨ Polyesters, uses Polyesters, uses RL: IMF (Industrial manufacture); POF (Polymer in formulation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (aminoplast-, siloxanes; crosslinkable copolymers of vinyl silane and vinyl esters of branched fatty acid for coating of) ΙT Coating materials (crosslinkable copolymers of vinyl silane and vinyl esters of branched fatty acid for coating compns.) ΙT Automobiles Crosslinking agents (crosslinkable copolymers of vinyl silane and vinyl esters of branched fatty acid for coating of) IT Aminoplasts RL: MOA (Modifier or additive use); USES (Uses) (crosslinking agents; crosslinkable copolymers of vinyl silane and vinyl esters of branched fatty acid for coating compns.)

```
ΙT
     Polysiloxanes, uses
     Polysiloxanes, uses
     Polysiloxanes, uses
     RL: IMF (Industrial manufacture); POF (Polymer in
     formulation); TEM (Technical or engineered material use); PREP
     (Preparation); USES (Uses)
        (epoxy-polyester-; crosslinkable copolymers of vinyl silane and vinyl
        esters of branched fatty acid for coating compns.)
IT
     Polyesters, uses
     Polyesters, uses
     Polyesters, uses
     RL: IMF (Industrial manufacture); POF (Polymer in
     formulation); TEM (Technical or engineered material use); PREP
     (Preparation); USES (Uses)
        (epoxy-siloxane-; crosslinkable copolymers of vinyl silane and vinyl
        esters of branched fatty acid for coating compns.)
IT
     Aminoplasts
     Aminoplasts
     RL: IMF (Industrial manufacture); POF (Polymer in
     formulation); TEM (Technical or engineered material use); PREP
     (Preparation); USES (Uses)
        (polyester-, siloxanes; crosslinkable copolymers of vinyl silane and
        vinyl esters of branched fatty acid for coating of)
IT
     Polysiloxanes, uses
     Polysiloxanes, uses
     RL: IMF (Industrial manufacture); POF (Polymer in
     formulation); TEM (Technical or engineered material use); PREP
     (Preparation); USES (Uses)
        (polyester-; crosslinkable copolymers of vinyl silane and vinyl esters
        of branched fatty acid for coating compns.)
IT
     Polyurethanes, uses
     Polyurethanes, uses
     Polyurethanes, uses
     RL: IMF (Industrial manufacture); POF (Polymer in
     formulation); TEM (Technical or engineered material use); PREP
     (Preparation); USES (Uses)
        (polyester-polysiloxane-; crosslinkable copolymers of vinyl silane and
        vinyl esters of branched fatty acid for coating of)
IT
     Polysiloxanes, uses
     Polysiloxanes, uses
     Polysiloxanes, uses
     RL: IMF (Industrial manufacture); POF (Polymer in
    formulation); TEM (Technical or engineered material use); PREP
     (Preparation); USES (Uses)
        (polyester-polyurethane-; crosslinkable copolymers of vinyl silane and
        vinyl esters of branched fatty acid for coating of)
TΤ
    Epoxy resins, uses
    Epoxy resins, uses
    Epoxy resins, uses
    RL: IMF (Industrial manufacture); POF (Polymer in
    formulation); TEM (Technical or engineered material use); PREP
     (Preparation); USES (Uses)
        (polyester-siloxane-; crosslinkable copolymers of vinyl silane and
        vinyl esters of branched fatty acid for coating compns.)
    Polyesters, uses
ΙT
    Polyesters, uses
    RL: IMF (Industrial manufacture); POF (Polymer in
    formulation); TEM (Technical or engineered material use); PREP
     (Preparation); USES (Uses)
```

```
(polysiloxane-; crosslinkable copolymers of vinyl silane and vinyl
        esters of branched fatty acid for coating compns.)
     Polyesters, uses
IT
     Polyesters, uses
     Polyesters, uses
     RL: IMF (Industrial manufacture); POF (Polymer in
     formulation); TEM (Technical or engineered material use); PREP
     (Preparation); USES (Uses)
        (polyurethane-polysiloxane-; crosslinkable copolymers of vinyl silane
        and vinyl esters of branched fatty acid for coating of)
TΤ
     Aminoplasts
     Aminoplasts
     RL: IMF (Industrial manufacture); POF (Polymer in
     formulation); TEM (Technical or engineered material use); PREP
     (Preparation); USES (Uses)
        (siloxane-, polyester-; crosslinkable copolymers of vinyl silane and
        vinyl esters of branched fatty acid for coating of)
IT
     75-13-8D, Isocyanic acid, esters, polymers
     RL: MOA (Modifier or additive use); USES (Uses)
        (Polyisocyanates, crosslinking agents; crosslinkable copolymers of
        vinyl silane and vinyl esters of branched fatty acid for coating
        compns.)
     226935-81-5P
                    226935-82-6P
                                   226935-84-8P
                                                   226935-85-9P, Cardura E
TΨ
     10-maleic anhydride-Silquest A 171-trimethylpentanediol-VeoVa 9 copolymer
                    226935-90-6P 227026-78-0P
     226935-88-2P
                                                 227026-79-1P
     RL: IMF (Industrial manufacture); POF (Polymer in
     formulation); TEM (Technical or engineered material use); PREP
     (Preparation); USES (Uses)
        (crosslinkable copolymers of vinyl silane and vinyl esters of branched
        fatty acid for coating compns.)
ΙT
     9003-08-1
     RL: MOA (Modifier or additive use); USES (Uses)
        (crosslinking agents; crosslinkable copolymers of vinyl
        silane and vinyl esters of branched fatty acid for
        coating compns.)
ΙT
     9003-08-1
     RL: MOA (Modifier or additive use); USES (Uses)
        (crosslinking agents; crosslinkable copolymers of vinyl
        silane and vinyl esters of branched fatty acid for
        coating compns.)
RN
     9003-08-1 HCAPLUS
CN
     1,3,5-Triazine-2,4,6-triamine, polymer with formaldehyde (9CI) (CA INDEX
     NAME)
     CM
          1
     CRN 108-78-1
     CMF C3 H6 N6
       NH<sub>2</sub>
```

CM 2

CRN 50-00-0 CMF C H2 O

 $H_2C = 0$ 

```
L76 ANSWER 11 OF 27 HCAPLUS COPYRIGHT 2004 ACS on STN
     1997:526154 HCAPLUS
AN
DN
     127:222070
ED
     Entered STN: 16 Aug 1997
TI
     Pigment compositions with good hue, coloring ability, and dispersibility
     Imagawa, Ippei; Koide, Masashi; Machida, Yasuaki; Ueki, Katsuyuki
IN
     Toyo Ink Mfg. Co., Ltd., Japan
PA
     Jpn. Kokai Tokkyo Koho, 11 pp.
SO
     CODEN: JKXXAF
DT
     Patent
LA
     Japanese
     ICM C09B067-20
IC
     ICS C08L003-00; C08L091-06; C08L101-00; C09C003-08
     42-12 (Coatings, Inks, and Related Products)
CC
     Section cross-reference(s): 37, 40, 41, 74
FAN.CNT 1
                                          APPLICATION NO.
     PATENT NO.
                       KIND DATE
                                                                 DATE
                                           ______
                        ----
                        A2
DF 09202861
PRAI JP 1996-12486
CLASS
                                                                19960129
                               19970805
                                          JP 1996-12486
                               19960129
 PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
 ______
JP 09202861 ICM ICS
                       C09B067-20
                       C08L003-00; C08L091-06; C08L101-00; C09C003-08
OS
    MARPAT 127:222070
AΒ
     Pigment compns. useful as colorants for printing inks, coatings,
     thermoplastics, etc. contain 0.01-90% pigments and 1-80% Me(CHR)nCH2CO2H
     (I; R = H, Me; n = 20-50). Thus, I (R = H, n = 20) 50, 131 P (polyethylene wax) 100, Lionol Yellow GGT 300, and Sumikathene G 808
     (polyethylene) 120 parts were mixed to prepare a pigment composition for
printing
     inks.
ST
     pigment fatty acid ink; thermoplastic pigment fatty acid; hue pigment
     fatty acid; dispersibility pigment fatty acid; coloring ability pigment
     fatty acid
ΙT
     Carbon black, uses
     RL: TEM (Technical or engineered material use); USES (Uses)
        (MB 45; pigments containing higher fatty acids for inks, coatings, and
        thermoplastics)
IT
    Fatty acids, uses
    RL: MOA (Modifier or additive use); PRP (Properties); TEM (Technical or
     engineered material use); USES (Uses)
        (long-chain; pigments containing higher fatty acids for inks, coatings, and
        thermoplastics)
IT
    Coating materials
    Electrophotographic toners
     Pigments, nonbiological
        (pigments containing higher fatty acids for inks, coatings, and
```

thermoplastics) ΙT Paraffin waxes, uses RL: MOA (Modifier or additive use); USES (Uses) (pigments containing higher fatty acids for inks, coatings, and thermoplastics) IT Polypropene fibers, uses RL: POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses) (pigments containing higher fatty acids for inks, coatings, and thermoplastics) ΙT Polyesters, uses RL: TEM (Technical or engineered material use); USES (Uses) (pigments containing higher fatty acids for inks, coatings, and thermoplastics) ΙT Inks (printing; pigments containing higher fatty acids for inks, coatings, and thermoplastics) ΤТ Plastics, uses RL: POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses) (thermoplastics; pigments containing higher fatty acids for inks, coatings, and thermoplastics) IT 9010-79-1, Ethylene-propylene copolymer RL: POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses) (APAO-UT 2385; pigments containing higher fatty acids for inks, coatings, and thermoplastics) 1047-16-1, Cinquasia Red YRT 759D ΙT RL: TEM (Technical or engineered material use); USES (Uses) (Cinquasia Red YRT 759D; pigments containing higher fatty acids for inks, coatings, and thermoplastics) ΙT 5280-80-8, Cromophtal Yellow GR RL: TEM (Technical or engineered material use); USES (Uses) (Cromophtal Yellow GR; pigments containing higher fatty acids for inks, coatings, and thermoplastics) 9003-53-6, Esbrite 2V IT RL: POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses) (Esbrite 2V; pigments containing higher fatty acids for inks, coatings, and thermoplastics) ΙT 106565-43-9 RL: POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses) (Hipol J 740, fibers; pigments containing higher fatty acids for inks, coatings, and thermoplastics) ΙT 147-14-8, Lionol Blue FG 7330 RL: TEM (Technical or engineered material use); USES (Uses) (Lionol Blue FG 7330; pigments containing higher fatty acids for inks, coatings, and thermoplastics) ΙT 14302-13-7, Lionol Green 2Y 301 RL: TEM (Technical or engineered material use); USES (Uses) (Lionol Green 2Y 301; pigments containing higher fatty acids for inks, coatings, and thermoplastics)

9002-88-4, Polyethylene RL: POF (Polymer in formulation); TEM (Technical or engineered

coatings, and thermoplastics)

RL: TEM (Technical or engineered material use); USES (Uses)

(Lionol Yellow GGT; pigments containing higher fatty acids for inks,

6358-85-6, Lionol Yellow GGT

ΙT

material use); USES (Uses) (Mirason 68, Sumikathene G 808, Sanwax 131P and Topcoat PW-M 10; pigments containing higher fatty acids for inks, coatings, and thermoplastics) ΙT 9003-07-0, Polypropylene RL: POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses) (Noblen JH-G, fibers; pigments containing higher fatty acids for inks, coatings, and thermoplastics) ΙT 24968-12-5, Novadur 5010R3 RL: TEM (Technical or engineered material use); USES (Uses) (Novadur 5010R3; pigments containing higher fatty acids for inks, coatings, and thermoplastics) IT 25067-34-9, Ethylene vinyl alcohol copolymer RL: TEM (Technical or engineered material use); USES (Uses) (Soarlite K; pigments containing higher fatty acids for inks, coatings, and thermoplastics) ΙT 24937-78-8, Ethylene-vinyl acetate copolymer RL: POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses) (Sumitate HE 10; pigments containing higher fatty acids for inks, coatings, and thermoplastics) IT 13463-67-7, Titanium oxide (TiO2), uses RL: TEM (Technical or engineered material use); USES (Uses) (Tipure R 101; pigments containing higher fatty acids for inks, coatings, and thermoplastics) ΙT 1309-37-1, Red iron oxide, uses RL: TEM (Technical or engineered material use); USES (Uses) (Toda Color 180ED; pigments containing higher fatty acids for inks, coatings, and thermoplastics) IT57455-37-5, Ultramarine Blue RL: TEM (Technical or engineered material use); USES (Uses) (Ultramarine Blue 2000; pigments containing higher fatty acids for inks, coatings, and thermoplastics) ΙT 188004-08-2P, Ethyl acrylate-formaldehyde-methacrylic acid-melamine-methyl methacrylate-styrene copolymer RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (coatings; pigments containing higher fatty acids for inks, coatings, and thermoplastics) ΙT 506-48-9, Octacosanoic acid 2433-96-7, Tricosanoic acid 38232-07-4, 106405-06-5 121730-38-9, Octatetracontanoic Heptatriacontanoic acid acid RL: MOA (Modifier or additive use); USES (Uses) (pigments containing higher fatty acids for inks, coatings, and thermoplastics) ΙT 28263-96-9, Ethyl acrylate-methacrylic acid-methyl methacrylate-styrene copolymer RL: POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses) (pigments containing higher fatty acids for inks, coatings, and thermoplastics) ΙT 1328-53-6, Lionol Green Y 102 26062-94-2 39283-39-1, Quinacridone Red RL: TEM (Technical or engineered material use); USES (Uses) (pigments containing higher fatty acids for inks, coatings, and thermoplastics) 188004-08-2P, Ethyl acrylate-formaldehyde-methacrylic ΙT acid-melamine-methyl methacrylate-styrene copolymer

RL: IMF (Industrial manufacture); TEM (Technical or engineered

material use); PREP (Preparation); USES (Uses)
 (coatings; pigments containing higher fatty acids for

inks, coatings, and thermoplastics)

RN 188004-08-2 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, polymer with ethenylbenzene, ethyl 2-propenoate, formaldehyde, methyl 2-methyl-2-propenoate and 1,3,5-triazine-2,4,6-triamine (9CI) (CA INDEX NAME)

CM 1

CRN 140-88-5 CMF C5 H8 O2

CM 2

CRN 108-78-1 CMF C3 H6 N6

CM 3

CRN 100-42-5 CMF C8 H8

 $H_2C = CH - Ph$ 

CM 4

CRN 80-62-6 CMF C5 H8 O2

$$\begin{array}{c|c} ^{\text{H}_2\text{C}} & \text{O} \\ \parallel & \parallel \\ \text{Me-C-C-OMe} \end{array}$$

CM 5

CRN 79-41-4 CMF C4 H6 O2

CM 6

CRN 50-00-0 CMF C H2 O

 $H_2C = O$ 

PATENT NO.

JP 08325359

-----

ICM

ICS

```
L76
    ANSWER 12 OF 27 HCAPLUS COPYRIGHT 2004 ACS on STN
    1997:174518 HCAPLUS
AN
DN
    126:173095
ED
    Entered STN: 14 Mar 1997
    Alkyd resin compositions, coatings containing them with good handling
ΤI
    properties and method for forming coating films
IN
    Takeda, Yoshiro; Yoda, Kohei
    Lion Corp, Japan
PA
    Jpn. Kokai Tokkyo Koho, 7 pp.
SO
    CODEN: JKXXAF
DT
    Patent
LA
    Japanese
TC
    ICM C08G059-50
    ICS C08G059-14; C08K005-17; C08L063-00; C08L067-08; C09D163-00;
         C09D167-08
CC
    42-8 (Coatings, Inks, and Related Products)
FAN.CNT 1
    PATENT NO.
                               DATE
                                          APPLICATION NO.
                                                                DATE
                       KIND
                                          -----
                       ____
    JP 08325359
                                          JP 1995-131640
                                                               19950530
                        A2
                               19961210
PRAI JP 1995-131640
                               19950530
CLASS
```

CLASS PATENT FAMILY CLASSIFICATION CODES

C08G059-50

C09D163-00; C09D167-08

AB Title compns. comprise (A) epoxy-modified alkyd resins or mixts. of alkyd resins and epoxide group-terminated compds. and (B)

RN[(CH2CH2O)mH](CH2CH2O)nH (R = C6-22 alkyl, alkenyl, or acyl; m + n = 1-10). Solvent-free coatings comprising the compns., melamine resin curing agents, and optionally neutralizing agents (for unreacted CO2H groups) are also claimed. Thus, oleic acid 226, trimethylolpropane 145, neopentyl glycol 112, phthalic anhydride 100, trimellitic anhydride 130, and Epikote 828 135 were heated at 150-220° for 4 h, mixed with 135 parts oleyl diethanolamide, neutralized with 85 parts Et3N, and diluted with H2O to give a transparent resin, 100 parts of which was mixed with 27 parts Cymel 325, 3.6 parts Carbon Black MA 100, and H2O, applied on an Al plate, and baked at 140° for 20 min to form coating film showing

C08G059-14; C08K005-17; C08L063-00; C08L067-08;

```
60°-gloss 91.2%, pencil hardness H, no foaming and sagging, and
     good resistance against alkali (2% NaOH, 48 h), acid (5% H2SO4, 48 h),
     water (50°, 200 h) and weather (300 h).
     coating epoxy alkyd melamine crosslinker; oleyl diethanolamide epoxy alkyd
ST
     coating; solventless epoxy modified alkyd coating; melamine resin
     crosslinked alkyd coating
ΙT
     Epoxy resins, uses
     Epoxy resins, uses
     RL: IMF (Industrial manufacture); POF (Polymer in formulation); PRP
     (Properties); TEM (Technical or engineered material use); PREP
     (Preparation); USES (Uses)
        (alkyd, crosslinked; compns. for solventless coatings with good
        handling properties)
ΙT
     Alkyd resins
     Alkyd resins
     RL: IMF (Industrial manufacture); POF (Polymer in formulation); PRP
     (Properties); TEM (Technical or engineered material use); PREP
     (Preparation); USES (Uses)
        (epoxy, crosslinked; compns. for solventless coatings with good
        handling properties)
ΙT
     Coating materials
        (solventless; epoxy-modified melamine resin-crosslinkable alkyd resin
        compns. for)
     Amides, uses
IT
     Amines, uses
     RL: MOA (Modifier or additive use); USES (Uses)
        (tertiary, handling improvers; in alkyd-epoxy resin compns. for
        solventless coatings)
ΙT
                                     120-40-1, Lauric diethanolamide
     93-83-4, Oleic diethanolamide
     136-26-5, Capric acid diethanolamide
                                            31587-79-8
     RL: MOA (Modifier or additive use); USES (Uses)
        (handling improvers; in alkyd-epoxy resin compns. for solventless
        coatings)
     187040-99-9P
IΤ
     RL: IMF (Industrial manufacture); POF (Polymer in formulation); PREP
     (Preparation); USES (Uses)
        (manufacture and use in solventless coatings with good handling properties)
ΙT
     187041-02-7
     RL: POF (Polymer in formulation); PRP (Properties); TEM
     (Technical or engineered material use); USES (Uses)
        (solventless coatings with good handling properties from)
IT
     187041-02-7
     RL: POF (Polymer in formulation); PRP (Properties); TEM
     (Technical or engineered material use); USES (Uses)
        (solventless coatings with good handling properties from)
RN
     187041-02-7 HCAPLUS
     5-Isobenzofurancarboxylic acid, 1,3-dihydro-1,3-dioxo-, polymer with
CN
     (chloromethyl)oxirane, 2,2-dimethyl-1,3-propanediol, 2-ethyl-2-
     (hydroxymethyl)-1,3-propanediol, formaldehyde, 1,3-isobenzofurandione,
     4,4'-(1-methylethylidene)bis[phenol] and 1,3,5-triazine-2,4,6-triamine,
     (9Z)-9-octadecenoate, compd. with N, N-diethylethanamine (9CI) (CA INDEX
     NAME)
     CM
          1
     CRN 121-44-8
     CMF C6 H15 N
```

CRN 187041-01-6 CMF

C18 H34 O2 . x (C15 H16 O2 . C9 H4 O5 . C8 H4 O3 . C6 H14 O3 . C5 H12 O2 . C3 H6 N6 . C3 H5 C1 O . C H2 O)x

3 CM

CRN 112-80-1 CMF C18 H34 O2

Double bond geometry as shown.

$$HO_2C$$
 (CH<sub>2</sub>)7 Z (CH<sub>2</sub>)7 Me

CM 4

CRN 187041-00-5

CMF (C15 H16 O2 . C9 H4 O5 . C8 H4 O3 . C6 H14 O3 . C5 H12 O2 . C3  ${\tt H6\ N6\ .\ C3\ H5\ Cl\ O\ .\ C\ H2\ O)\,x}$ 

CCI PMS

CM 5

CRN 552-30-7 CMF C9 H4 O5

CM 6

CRN 126-30-7 CMF C5 H12 O2

CRN 108-78-1 CMF C3 H6 N6

CM 8

CRN 106-89-8 CMF C3 H5 C1 O

CM 9

CRN 85-44-9 CMF C8 H4 O3

CM 10

CRN 80-05-7 CMF C15 H16 O2

CM 11

CRN 77-99-6 CMF C6 H14 O3

$$\begin{array}{c} \text{CH}_2-\text{OH} \\ | \\ \text{HO-CH}_2-\text{C-Et} \\ | \\ \text{CH}_2-\text{OH} \end{array}$$

CM 12

ICS

CRN 50-00-0 CMF C H2 O

H2C=0

```
L76 ANSWER 13 OF 27 HCAPLUS COPYRIGHT 2004 ACS on STN
    1997:111075 HCAPLUS
AN ·
DN
    126:119155
ED
    Entered STN: 17 Feb 1997
TΙ
    Thermal-curable aqueous coating compositions for metals
    Iwahashi, Masanori; Moriki, Juichiro; Oonishi, Kyoshi
IN
PA
    Dainippon Ink & Chemicals, Japan
SO
    Jpn. Kokai Tokkyo Koho, 19 pp.
    CODEN: JKXXAF
DT
    Patent
LA
    Japanese
IC
    ICM C09D163-00
    ICS C08G059-16; C09D161-06; C09D161-20; C09D175-00; C09D201-00
    42-10 (Coatings, Inks, and Related Products)
FAN.CNT 1
                           DATE APPLICATION NO.
    PATENT NO.
                    KIND DATE
                                                        DATE
    _____
                                     -----
                    ----
                    A2
                                                        19960327
    JP 08325509
                           19961210
                                    JP 1996-72148
PΙ
PRAI JP 1995-71484
                           19950329
CLASS
PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
______
JP 08325509
              ICM
                    C09D163-00
```

AB Title compns., showing good storage stability, coatability, adhesion, anticorrosion, and hardness, comprise (a) epoxy resins modified with ≥2 P-bonded OH group-containing compds. and carboxylic acids selected from (un)saturated carboxylic acid (derivs.), aromatic carboxylic acid (derivs.),

and oxycarboxylic acid (derivs.), (b) water-soluble or dispersible resins (excluding the above modified epoxy resins), and (c) hardeners selected from aminoplasts, phenoplasts, and blocked isocyanate resins, optionally and (d) pigments. A stainless steel plate was coated with a clear or white (containing TiO2) aqueous composition (storage-stable at 40° for 1 mo) containing Epiclon 1050 propionate phosphate salt with dimethylethanolamine,

C08G059-16; C09D161-06; C09D161-20; C09D175-00;

C09D201-00

ST

IT

IT

IT

IT

ΙT

IT

TΤ

ΙT

HCHO-benzoquanamine copolymer, and acrylic acid-Bu acrylate-Et acrylate-2-hydroxyethyl methacrylate-styrene copolymer dimethylethanolamine salt to a thickness of 5-7  $\mu m$  or 15-20  $\mu m$  for clear and white composition, resp., and baked at  $260^{\circ}$  for 45 s to form a surface showing good adhesion, anticorrosion, and hardness with high transparency and gloss for clear and white composition, resp. thermal curable acrylic epoxy aq coating; polyester epoxy thermal curable aq coating; alky resin epoxy thermal curable coating; storage stability coatability aq coating metal; epoxy resin phosphate carboxylate aq coating Epoxy resins, uses RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (acrylic or polyester- or alkyd resin-; storage-stable/thermalcurable/anticorrosive aqueous coatings with coatability for metals) Coating materials (anticorrosive, clear or colored; epoxy resin phosphate carboxylate/water-dispersible resin/hardener-containing thermal-curable aqueous coatings for metals) Fatty acids, reactions RL: RCT (Reactant); RACT (Reactant or reagent) (branched fatty acids, esters with epoxy resin phosphates; alkyd resin/melamine resin-containing thermal-curable aqueous coatings for metals) Fatty acids, reactions RL: RCT (Reactant); RACT (Reactant or reagent) (castor-oil and tall-oil; epoxy resin phosphate carboxylate/melamine resin-containing thermal-curable aqueous coatings for metals) Metals, miscellaneous RL: MSC (Miscellaneous) (epoxy resin phosphate carboxylate/water-dispersible resin/hardener-containing thermal-curable aqueous coatings for metals) Acrylic polymers, uses Alkyd resins RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (epoxy resin-; storage-stable/thermal-curable/anticorrosive aqueous coatings with coatability for metals) Aminoplasts RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (polymers with alkyd resins and epoxy resin phosphate carboxylates, salts with amines; storage-stable/thermal-curable/anticorrosive aqueous coatings with coatability for metals) 65-85-0DP, Benzoic acid, esters with epoxy resin phosphates, polymers with alkyd resins and melamine resin, salts with amines, uses 79-10-7DP, 2-Propenoic acid, esters with epoxy resin phosphates, polymers with alkyd resins and melamine resin, salts with amines, uses 88-99-3DP, 1,2-Benzenedicarboxylic acid, polymers with castor oil and tall oil fatty acids and polyols and epoxy resin phosphate carboxylates, salts with 108-01-0DP, salts with reaction products of alkyd resins amines, uses and epoxy resin phosphate carboxylates 115-77-5DP, polymers with castor oil and tall oil fatty acids and epoxy resin phosphate carboxylates, salts 121-44-8DP, salts with reaction products of alkyd resins and with amines epoxy resin phosphate carboxylates 121-91-5DP, 1,3-Benzenedicarboxylic

acid, polymers with castor oil and tall oil fatty acids and polyols and

phosphate carboxylates, salts with amines 7664-38-2DP, Phosphoric acid, esters with epoxy resin carboxylates, polymers with alkyd resins and

126-30-7DP,

epoxy resin phosphate carboxylates, salts with amines, uses

polymers with castor oil and tall oil fatty acids and epoxy resin

melamine resin, salts with amines, uses 9003-08-1DP, Super Beckamine S 695, polymers with alkyd resins and epoxy resin phosphate carboxylates, 25068-38-6DP, phosphates and carboxylates, polymers salts with amines with alkyd resins and melamine resin, salts with amines 186047-92-7P 186047-95-0P, uses 186047-98-3P 186048-01-1P **186048-04-4P** 186148-04-9P 186148-29-8P 186148-31-2P RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (storage-stable/thermal-curable/anticorrosive aqueous coatings with coatability for metals) TT 186048-04-4P RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (storage-stable/thermal-curable/anticorrosive aqueous coatings with coatability for metals) RN 186048-04-4 HCAPLUS CN 2-Propenoic acid, 2-methyl-, 2-hydroxyethyl ester, polymer with butyl 2-propenoate, (chloromethyl)oxirane polymer with 4,4'-(1methylethylidene)bis[phenol] octadecanoate phosphate, ethenylbenzene, ethyl 2-propenoate, formaldehyde, 6-phenyl-1,3,5-triazine-2,4-diamine and 2-propenoic acid, compd. with N, N-diethylethanamine and 2-(dimethylamino)ethanol (9CI) (CA INDEX NAME) CM CRN 121-44-8 CMF C6 H15 N Εt Et-N-Et 2 CM 108-01-0 CRN CMF C4 H11 N O  $Me_2N-CH_2-CH_2-OH$ CM 3 CRN 186048-03-3 (C18 H36 O2 . x (C15 H16 O2 . C3 H5 Cl O)x . C9 H9 N5 . C8 H8 . C7 CMF H12 O2 . C6 H10 O3 . C5 H8 O2 . C3 H4 O2 . C H2 O . x H3 O4 P)x CCI PMS CM4 CRN 868-77-9

CMF C6 H10 O3

CRN 141-32-2 CMF C7 H12 O2

$$\begin{array}{c} \text{O} \\ \parallel \\ \text{n-BuO-C-CH} \end{array}$$

CM 6

CRN 140-88-5 CMF C5 H8 O2

CM 7

CRN 100-42-5 CMF C8 H8

$$H_2C = CH - Ph$$

CM 8

CRN 91-76-9 CMF C9 H9 N5

CM S

CRN 79-10-7

TRUONG 10/642873 8/10/04 Page 43

CMF C3 H4 O2

CM 10

CRN 50-00-0 CMF C H2 O

 $H_2C = O$ 

CM 11

CRN 186048-02-2

CMF C18 H36 O2 . x (C15 H16 O2 . C3 H5 Cl O)x . x H3 O4 P

CM 12

CRN 7664-38-2 CMF H3 O4 P

CM 13

CRN 57-11-4 CMF C18 H36 O2

 ${\rm HO_2C^-}$  (CH<sub>2</sub>)<sub>16</sub>-Me

CM 14

CRN 25068-38-6

CMF (C15 H16 O2 . C3 H5 Cl O)x

CCI PMS

CM 15

CRN 106-89-8 CMF C3 H5 Cl O

CRN 80-05-7 CMF C15 H16 O2

```
ANSWER 14 OF 27 HCAPLUS COPYRIGHT 2004 ACS on STN
```

AN 1997:2492 HCAPLUS

DN 126:33155

ED Entered STN: 04 Jan 1997

Scratch-resistant coating composition for clearcoats and topcoats Huybrechts, Jozef; Kernaghan, Stuart Alexander; Vervoort, Robert ΤI

IN

E. I. Du Pont de Nemours & Co., USA; Huybrechts, Jozef; Kernaghan, Stuart PA Alexander; Vervoort, Robert

PCT Int. Appl., 22 pp. SO

CODEN: PIXXD2

DT Patent

LA English

IC

ICM C09D133-06 ICS C09D167-00; C09D175-04; C08G018-40

42-10 (Coatings, Inks, and Related Products) CC

FAN.	CNT	1	_	•	•						-							
	PATENT NO.			KIND DATE				APPLICATION NO.						DATE				
ΡI	WO 9634924			A1 19961107 CN, JP, KR, MX, U										19960429				
		RW: AT,								GB	GR,	IE,	IT,	LU,	MC,	NL,	PT,	SE
	AU	9656329			A1		1996	1121		ΑU	1996-	56329	9		1	9960	425	
	ΑU	718129			В2		2000	0406										
	CA	2219887			AA		1996	1107		CA	1996-	22198	387		1	99604	429	
	EΡ	824573			A1		1998	0225		EΡ	1996-	91326	50		1	9960	429	
	ΕP	824573			В1		2000	1206										
		R: BE,	DE,	ES,		•	IT,	NL,	SE,	PΤ	ı							
	CN	1189850			Α		1998	0805		CN	1996-	1951	79			99604		
	BR	9608425			Α		1999	0330			1996-					9960		
	JΡ	11506477			Т2		1999	0608		JP	1996-	53339	99		1	99604	429	
	ES	2153573			Т3		2001	0301		ES	1996-	91326	50		1	99604	429	
	PT	824573			${f T}$		2001	0430		PT	1996-	91326	50		1	99604	129	
	US	5977256			Α		1999	1102		US	1997-	93066	56		1	9971	029	
PRAI	US	1995-4332	284		Α		1995	0502										
	WO	1996-US59	950		W		1996	0429										
CLASS	3																	

PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES

C09D133-06 WO 9634924 ICM C09D167-00; C09D175-04; C08G018-40 ICS A crosslinkable coating composition comprises (1) .apprx.15-70% acrylic polyol AB (weight-average mol. weight .apprx.2500-40,000, OH value 50-180 mg KOH/g, and Tg -30 to +70°); (2) 5-60% optionally substituted polyester polyol (weight-average mol. weight .apprx.2000-80,000, OH value .apprx.50-220 mg KOH/q, optional substituent selected from ≥1 member of the group trialkoxysilyl and urethane) and the polyol comprising ≥10% by weight of the hydrogenated reaction product of  $\geq 1$  of dimer fatty acid and polyol, polyacid and dimer alc., and dimer fatty acid and dimer alc.; and (3) .apprx.10-60% crosslinking agent selected from ≥1 alkoxylated melamine-formaldehyde adduct and a polyisocyanate. The composition is useful for coating metallic substrates, especially as clearcoats and topcoats. The improved properties are derived from the dimer acid/alc. component of the polyester polyol. Thus, an acrylic polyol, a polyester polyol containing dimer fatty acid, and melamine-formaldehyde polymer were formulated into a clearcoat composition which exhibited excellent scratch resistance and good appearance. ST scratch resistant crosslinkable coating compn; acrylic polyol polyester polyol coating compn; dimer fatty acid polyester polyol coating; clearcoat compn scratch resistant; topcoat compn scratch resistant ΙT Aminoplasts RL: MOA (Modifier or additive use); TEM (Technical or engineered material use); USES (Uses) (Luwipal, crosslinking agent; scratch-resistant crosslinkable coating compns. containing acrylic polyol, dimer fatty acid polyester polyol, and crosslinking agent) ΙT Crosslinking agents (isocyanates and melamine-formaldehyde resins; scratch-resistant crosslinkable coating compns. containing acrylic polyol, dimer fatty acid polyester polyol, and crosslinking agent) IT Acrylic polymers, uses Polyesters, uses RL: IMF (Industrial manufacture); POF (Polymer in formulation); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (scratch-resistant crosslinkable coating compns. containing acrylic polyol, dimer fatty acid polyester polyol, and crosslinking agent) IT Coating materials (scratch-resistant, clearcoats and topcoats; scratch-resistant crosslinkable coating compns. containing acrylic polyol, dimer fatty acid polyester polyol, and crosslinking agent) 9003-08-1, Melamine-formaldehyde copolymer TΨ RL: MOA (Modifier or additive use); TEM (Technical or engineered material use); USES (Uses) (Luwipal, crosslinking agent; scratch-resistant crosslinkable coating compns. containing acrylic polyol, dimer fatty acid polyester polyol, and crosslinking agent) 51097-42-8P, Acrylic acid-butyl methacrylate-2-ethylhexyl ΙT acrylate-2-hydroxyethyl methacrylate-styrene copolymer 55774-94-2P, Acrylic acid-butyl methacrylate-2-ethylhexyl methacrylate-2-hydroxyethyl 94798-18-2P, Butyl acrylate-butyl methacrylate-styrene copolymer

RL: IMF (Industrial manufacture); POF (Polymer in formulation); PRP (Properties); TEM (Technical or engineered material

184642-81-7P

184642-86-2P

184642-78-2P

184642-83-9P

184642-82-8P

methacrylate-2-hydroxyethyl acrylate-styrene copolymer

184642-80-6P

184642-85-1P

184642-79-3P

184642-84-0P

use); PREP (Preparation); USES (Uses)

(scratch-resistant crosslinkable coating compns. containing acrylic polyol, dimer fatty acid polyester polyol, and crosslinking agent)

IT 9003-08-1, Melamine-formaldehyde copolymer

RL: MOA (Modifier or additive use); TEM (Technical or engineered material use); USES (Uses)

(Luwipal, crosslinking agent; scratch-resistant crosslinkable coating compns. containing acrylic polyol, dimer <u>fatty acid</u> polyester polyol, and crosslinking agent)

RN 9003-08-1 HCAPLUS

CN 1,3,5-Triazine-2,4,6-triamine, polymer with formaldehyde (9CI) (CA INDEX NAME)

CM 1

CRN 108-78-1 CMF C3 H6 N6

CM 2

CRN 50-00-0 CMF C H2 O

H2C==O

L76 ANSWER 15 OF 27 HCAPLUS COPYRIGHT 2004 ACS on STN

AN 1996:171905 HCAPLUS

DN 124:205109

ED Entered STN: 26 Mar 1996

TI Alkyd resin solutions for anticorrosive coatings containing melamine resins

IN Vaszilcsin, Ileana; Dalea, Ion

PA Intreprinderea de Lacuri si Vopsele "Azur" Timisoara, Rom.

SO Rom., 3 pp. CODEN: RUXXA3

DT Patent

LA Romanian

IC ICM C08G063-12

CC 42-8 (Coatings, Inks, and Related Products)

FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE
PI RO 104961 B1 19941210 RO 1989-140037 19890603
PRAI RO 1989-140037 19890603

CLASS

PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES

```
______
                ICM
                       C08G063-12
RO 104961
     Title solns. are manufactured from soybean-oil fatty acid 19, phthalic
AB
     anhydride 26.7, trimethylolpropane 24.6, Ph3PO3 (50% solution) 0.04, BuOH
     6.5, and xylene 51.5 parts and exhibit acid index 25 mg KOH/g resin,
     4-mm-diameter cup viscosity 90-125 s, solids content 53-57%, and I color
     index ≤20 mg/100 g (50% solution). These solns. replace alkyd resin
     solns. prepared from castor oil instead of soybean-oil fatty acids and
     provide coatings that cure at room temperature in 24 h or at 80° in 1 h.
     alkyd coating soybean fatty acid; melamine resin crosslinker alkyd
ST
     coating; trimethylolpropane alkyd coating anticorrosive; phthalic alkyd
     coating anticorrosive
ΙT
     Alkyd resins
     RL: IMF (Industrial manufacture); POF (Polymer in
     formulation); PREP (Preparation); USES (Uses)
        (esters, with soybean-oil fatty acids; alkyd resin solns. for
        anticorrosive coatings containing melamine resins)
ΙT
     Coating materials
        (anticorrosive, alkyd resin solns. for anticorrosive coatings containing
        melamine resins)
ΙT
     Fatty acids, uses
     RL: IMF (Industrial manufacture); POF (Polymer in
     formulation); PREP (Preparation); USES (Uses)
        (soya, esters, with alkyd resins; alkyd resin solns. for anticorrosive
        coatings containing melamine resins)
IT
     30525-36-1DP, Phthalic anhydride-trimethylolpropane copolymer, esters with
     soybean-oil fatty acids
     RL: IMF (Industrial manufacture); POF (Polymer in
     formulation); PREP (Preparation); USES (Uses)
        (alkyd resin solns. for anticorrosive coatings containing melamine resins)
IT
     174572-34-ODP, Formaldehyde-melamine-phthalic anhydride-
     trimethylolpropane copolymer, esters with soybean-oil
     fatty acids
     RL: IMF (Industrial manufacture); PRP (Properties); TEM
     (Technical or engineered material use); PREP (Preparation); USES
        (cured coating; alkyd resin solns. for anticorrosive coatings containing
        melamine resins)
     174572-34-0DP, Formaldehyde-melamine-phthalic anhydride-
TΨ
     trimethylolpropane copolymer, esters with soybean-oil
     fatty acids
     RL: IMF (Industrial manufacture); PRP (Properties); TEM
     (Technical or engineered material use); PREP (Preparation); USES
     (Uses)
        (cured coating; alkyd resin solns. for anticorrosive coatings containing
       melamine resins)
    174572-34-0 HCAPLUS
RN
     Formaldehyde, polymer with 2-ethyl-2-(hydroxymethyl)-1,3-propanediol,
CN
     1,3-isobenzofurandione and 1,3,5-triazine-2,4,6-triamine (9CI) (CA INDEX
    NAME)
    CM
        1
    CRN 108-78-1
    CMF C3 H6 N6
```

CRN 85-44-9 CMF C8 H4 O3

3 CM

CRN 77-99-6 CMF C6 H14 O3

$$\begin{array}{c} \text{CH}_2-\text{OH} \\ | \\ \text{HO-CH}_2-\text{C-Et} \\ | \\ \text{CH}_2-\text{OH} \end{array}$$

CM 4

CRN 50-00-0 CMF C H2 O

Н2С = О

L76 ANSWER 16 OF 27 HCAPLUS COPYRIGHT 2004 ACS on STN

1995:441468 HCAPLUS AN

123:115588 DN

Entered STN: 25 Mar 1995 ED

Resin compositions for water-thinned coatings ΤI

Shima, Toshihide; Oosumi, Tatsuya; Yoshida, Michiro Sanyo Chemical Ind Ltd, Japan Jpn. Kokai Tokkyo Koho, 5 pp. ΙN

PA

SO

CODEN: JKXXAF

DTPatent

Japanese LA

IC ICM C09D167-00

ICS C09D161-32

42-10 (Coatings, Inks, and Related Products)

Section cross-reference(s): 55, 56

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE		
				~			
ΡI	JP 07011188	A2	19950113	JP 1994-113689	19940428		
PRAI	JP 1993-125258		19930428				

CLASS

PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES

JP 07011188 ICM C09D167-00 C09D161-32 ICS

The compns., useful for top coats of metal cans, comprise (A) polyesters AB composed of polyol-alkylene oxide adducts and polycarboxylic acids and (B) amino resins. Thus, trimethylolpropane-propylene oxide adduct 48, bisphenol A-ethylene oxide adduct 5, trimethylolpropane 9, pentaerythritol 11, stearic acid 7, phthalic anhydride 15, and trimellitic acid anhydride 5 parts were polymerized at 160° to give a polyester, 25 parts of which was blended with Cymel 327 28, H2O 35, Bu cellosolve 12, and diethanolamine 0.6 part, applied on an Al plate, and heated at 230  $^{\circ}$ for 2 min to form a coating showing pencil hardness 3 H, du-Pont impact strength 50, cross-cut adhesion 100/100 after 30-min immersion in H2O at 30°, and good appearance.

ST polyester aminoplast coating can; water resistance coating polyester aminoplast; impact resistance coating polyester melamine

ΙT

(amino resin-polyester-based water-thinned coatings with good impact and water resistance for cans)

IT Polyesters, uses

> RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(acrylic-aminoplast-, amino resin-polyester-based water-thinned coatings with good impact and water resistance for cans)

IT Aminoplasts

> RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(acrylic-polyester-, amino resin-polyester-based water-thinned coatings with good impact and water resistance for cans)

ΙT Aminoplasts

> RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(alkyd resin-, amino resin-polyester-based water-thinned coatings with good impact and water resistance for cans)

IT Alkyd resins

Polyesters, uses

RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(aminoplast-, amino resin-polyester-based water-thinned coatings with good impact and water resistance for cans)

ΙT Coating materials

(impact- and water-resistant, water-thinned, amino resin-polyesterbased water-thinned coatings with good impact and water resistance for

IT Aminoplasts

> RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

```
(polyester-, amino resin-polyester-based water-thinned coatings with
        good impact and water resistance for cans)
     165248-51-1P 165248-52-2P 166019-91-6P 166019-93-8P
IT
     RL: IMF (Industrial manufacture); PRP (Properties); TEM
     (Technical or engineered material use); PREP (Preparation); USES
     (Uses)
        (amino resin-polyester-based water-thinned coatings with good impact
        and water resistance for cans)
ΙT
     7429-90-5, Aluminum, uses
     RL: TEM (Technical or engineered material use); USES (Uses)
        (amino resin-polyester-based water-thinned coatings with good impact
        and water resistance for cans)
     166019-91-6P 166019-93-8P
ΙT
     RL: IMF (Industrial manufacture); PRP (Properties); TEM
     (Technical or engineered material use); PREP (Preparation); USES
     (Uses)
        (amino resin-polyester-based water-thinned coatings with good impact
        and water resistance for cans)
     166019-91-6 HCAPLUS
RN
CN
     5-Isobenzofurancarboxylic acid, 1,3-dihydro-1,3-dioxo-, polymer with
     2,2-bis(hydroxymethyl)-1,3-propanediol, 2-ethyl-2-(hydroxymethyl)-1,3-
     propanediol, formaldehyde, α-hydro-ω-hydroxypoly[oxy(methyl-
     1,2-ethanediyl)] ether with 2-ethyl-2-(hydroxymethyl)-1,3-propanediol
     (3:1), 1,3-isobenzofurandione, \alpha,\alpha'-[(1-methylethylidene)di-4,1-phenylene]bis[\omega-hydroxypoly(oxy-1,2-ethanediyl)] and
     1,3,5-triazine-2,4,6-triamine, octadecanoate (ester) (9CI) (CA INDEX
     NAME)
     CM
          1
     CRN 57-11-4
     CMF C18 H36 O2
HO_2C^- (CH<sub>2</sub>)<sub>16</sub>-Me
          2
     CM
     CRN
          166019-90-5
     CMF
          (C9 H4 O5 . C8 H4 O3 . C6 H14 O3 . C5 H12 O4 . C3 H6 N6 . (C3 H6 O)n
           (C3 H6 O)n (C3 H6 O)n C6 H14 O3 . (C2 H4 O)n (C2 H4 O)n C15 H16 O2 .
          C H2 O)x
     CCI
         PMS
          CM
                3
          CRN 32492-61-8
               (C2 H4 O)n (C2 H4 O)n C15 H16 O2
          CMF
          CCI PMS
```

HO 
$$CH_2-CH_2-O$$
  $Me$   $Me$   $Me$   $Me$ 

CRN 25723-16-4

(C3 H6 O)n (C3 H6 O)n (C3 H6 O)n C6 H14 O3 CMF

CCI IDS, PMS

CM5

CRN 552-30-7 CMF C9 H4 O5

CM 6

CRN 115-77-5 C5 H12 O4 CMF

$$\begin{array}{c} \text{CH}_2-\text{OH} \\ | \\ \text{HO-CH}_2-\text{C-CH}_2-\text{OH} \\ | \\ \text{CH}_2-\text{OH} \end{array}$$

TRUONG 10/642873 8/10/04 Page 52

CM 7

CRN 108-78-1 CMF C3 H6 N6

CM 8

CRN 85-44-9 CMF C8 H4 O3

CM 9

CRN 77-99-6 CMF C6 H14 O3

$$\begin{array}{c} \text{CH}_2-\text{OH} \\ | \\ \text{HO-CH}_2-\text{C-Et} \\ | \\ \text{CH}_2-\text{OH} \end{array}$$

CM 10

CRN 50-00-0 CMF C H2 O

 $H_2C = O$ 

RN 166019-93-8 HCAPLUS 5-Isobenzofurancarboxylic acid, 1,3-dihydro-1,3-dioxo-, polymer with 2,2-bis(hydroxymethyl)-1,3-propanediol, butyl 2-propenoate, ethenylbenzene, 2-ethyl-2-(hydroxymethyl)-1,3-propanediol, formaldehyde,  $\alpha$ -hydro- $\omega$ -hydroxypoly[oxy(methyl-1,2-ethanediyl)] ether with 2-ethyl-2-(hydroxymethyl)-1,3-propanediol (3:1), 2-hydroxyethyl

2-methyl-2-propenoate, 1,3-isobenzofurandione,  $\alpha,\alpha'-[(1-methylethylidene)di-4,1-phenylene]bis[<math>\omega$ -hydroxypoly(oxy-1,2-ethanediyl)], 2-propenoic acid and 1,3,5-triazine-2,4,6-triamine, octadecanoate (ester), compd. with 2-(dimethylamino)ethanol (9CI) (CAINDEX NAME)

CM 1

CRN 108-01-0 CMF C4 H11 N O

 $Me_2N-CH_2-CH_2-OH$ 

CM 2

CRN 57-11-4 CMF C18 H36 O2

 $HO_2C^-$  (CH<sub>2</sub>)<sub>16</sub>-Me

CM 3

CRN 166019-92-7

CMF (C9 H4 O5 . C8 H8 . C8 H4 O3 . C7 H12 O2 . C6 H14 O3 . C6 H10 O3 . C5 H12 O4 . C3 H6 N6 . (C3 H6 O)n (C3 H6 O)n (C3 H6 O)n C6 H14 O3 . C3 H4 O2 . (C2 H4 O)n (C2 H4 O)n C15 H16 O2 . C H2 O)x

CCI PMS

CM 4

CRN 32492-61-8

CMF (C2 H4 O)n (C2 H4 O)n C15 H16 O2

CCI PMS

HO 
$$CH_2-CH_2-O$$
  $Me$   $Me$   $Me$   $Me$ 

CM 5

CRN 25723-16-4

CMF (C3 H6 O)n (C3 H6 O)n (C3 H6 O)n C6 H14 O3

CCI IDS, PMS

$$CH_2 - CH_2 - CH_3H_6) - OH_2 - CH_2 - CH_3H_6) - OH_3H_6$$
 $CH_2 - CH_3H_6 - OH_3H_6 - OH_3H_6$ 

CRN 868-77-9 CMF C6 H10 O3

CM 7

CRN 552-30-7 CMF C9 H4 O5

CM 8

CRN 141-32-2 CMF C7 H12 O2

CM 9

CRN 115-77-5 CMF C5 H12 O4

$$\begin{array}{c} \text{CH}_2-\text{OH} \\ | \\ \text{HO-CH}_2-\text{C-CH}_2-\text{OH} \\ | \\ \text{CH}_2-\text{OH} \end{array}$$

CRN 108-78-1 CMF C3 H6 N6

CM 11

CRN 100-42-5 CMF C8 H8

 $H_2C = CH - Ph$ 

CM 12

CRN 85-44-9 CMF C8 H4 O3

CM 13

CRN 79-10-7 CMF C3 H4 O2

CRN 77-99-6 CMF C6 H14 O3

$$\begin{array}{c} \text{CH}_2-\text{OH} \\ | \\ \text{HO-CH}_2-\text{C-Et} \\ | \\ \text{CH}_2-\text{OH} \end{array}$$

CM 15

CRN 50-00-0 CMF C H2 O

 $H_2C = O$ 

```
L76 ANSWER 17 OF 27 HCAPLUS COPYRIGHT 2004 ACS on STN
    1994:9822 HCAPLUS
AN
```

DN 120:9822

ED Entered STN: 08 Jan 1994

Fireproofing agents for polyurethanes ΤI

Hashimoto, Yoichi IN

Matsushita Electric Works Ltd, Japan PΑ

Jpn. Kokai Tokkyo Koho, 3 pp. SO

CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM C08L061-20

ICS C08K005-09; C08L075-04

37-6 (Plastics Manufacture and Processing)

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 05230330	A2	19930907	JP 1992-36247	19920224
PRAI	JP 1992-36247		19920224		

CLASS

PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES
TD 05220220	TCM	C081 061 - 20

JP 05230330 C08L061-20

ICS C08K005-09; C08L075-04

Title agents comprising highly condensed amino resins and 0.1-100 phr AB higher fatty acid metal salts are blended in polyols. Thus, treating urea and HCHO at a mol ratio of 1:1.5 at 95° for 1 h and further treating the resulting resin at 150° for 1 h gave a highly condensed urea resin, 100 parts of which was blended with 1 part Zn stearate to give a fireproofing agent. A mixture of 80 parts polypropylene glycol and 20 parts the fireproofing agent formed no precipitate during 1-wk. storage at 20°.

ST urea resin fireproofing agent polyurethane; zinc stearate aminoplast fireproofing polyurethane; polyoxyalkylene zinc stearate aminoplast fireproofing

ΙT Fireproofing agents

> (amino resins and fatty acid metal salts, polyols containing, for polyurethanes)

Urethane polymers, miscellaneous IT

RL: POF (Polymer in formulation); USES (Uses)

(fireproofing agents for, polyols containing amino resins and fatty acid metal salts for)

IT Aminoplasts

RL: USES (Uses)

(fireproofing agents, containing fatty acid metal salts, polyols containing, for polyurethanes)

IT Fatty acids, compounds

RL: USES (Uses)

(metal salts, amino resins containing, fireproofing agents, polyols containing,

for polyurethanes)

ΙT 557-05-1, Zinc stearate

RL: USES (Uses)

(amino resins containing, fireproofing agents, polyols containing, for polyurethanes)

25322-69-4, Polypropylene glycol ΙT 25322-68-3, Polyethylene glycol

RL: USES (Uses)

(containing amino resins and fatty acid metal salts, for fire-resistant polyurethanes)

9003-08-1P, Formaldehyde-melamine copolymer 9011-05-6P, Formaldehyde-urea copolymer IT

RL: PREP (Preparation)

(preparation of, fireproofing agents, containing fatty acid metal salts, polyols containing, for polyurethanes)

ΙT 9003-08-1P, Formaldehyde-melamine copolymer

RL: PREP (Preparation)

(preparation of, fireproofing agents, containing fatty acid metal salts, polyols containing, for polyurethanes)

RN 9003-08-1 HCAPLUS

1,3,5-Triazine-2,4,6-triamine, polymer with formaldehyde (9CI) (CA INDEX CN NAME)

CM 1

CRN 108-78-1 CMF C3 H6 N6

2 CM

50-00-0 CRN CMF C H2 O

 $H_2C = 0$ 

```
ANSWER 18 OF 27 HCAPLUS COPYRIGHT 2004 ACS on STN
L76
    1993:498102 HCAPLUS
AN
DN
    119:98102
ED
    Entered STN: 04 Sep 1993
TΙ
    Aqueous-dispersion coatings containing epoxy-denatured alkyd resins
    Taniguchi, Hitoshi; Ishidoya, Masahiro; Nakashima, Ken; Ogawa, Hisao
IN
    Nippon Oil and Fats Co., Ltd., Japan U.S., 9 pp. Cont. of U.S. Ser. No. 206,477, abandoned.
PA
SO
    CODEN: USXXAM
DT
    Patent
LA
    English
    ICM C08K003-20
IC
NCL
    523403000
    42-10 (Coatings, Inks, and Related Products)
FAN.CNT 1
                      KIND
                                         APPLICATION NO.
    PATENT NO.
                              DATE
                                                                DATE
                               _____
                                          -----
                       ----
                                                                 _____
                       Α
                                                                19910624
    US 5202364
                               19930413
                                         US 1991-719668
PRAI US 1988-206477
                               19880613
CLASS
 PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
             ICM C08KUUU
NCL 523403000
US 5202364
                      C08K003-20
```

AB Aqueous dispersions that provide anticorrosive coatings with good appearance on vertical surfaces contain products of the neutralization of reaction products having acid value 10-30 prepared by reaction of epoxy resins and epoxy-modified alkyd resins. The viscosity of the dispersions are controllable by addition of organic solvents while maintaining the storage stability. Thus, reaction of lauric acid, adipic acid, Epikote 828, trimellitic anhydride, neopentyl glycol, and trimethylolpropane and reaction of the resulting product (I) with Epikote 1007 (II) gave a product (III) with acid number 16. Neutralizing III with dimethylethanolamine (IV), dispersing the neutralized product in water, and mixing the dispersion with a black pigment paste, a melamine resin crosslinker, addnl. IV, p-toluenesulfonic acid, and addnl. water gave a composition that exhibited better storage stability after addition of organic solvents than a similar composition in which I and II were presence as a mixture

and not a prereacted product.

- ST epoxy alkyd dispersion coating; laurate alkyd epoxy coating; storage stable epoxy alkyd coating; anticorrosive epoxy alkyd coating; trimethylolpropane alkyd epoxy coating; neopentyl glycol alkyd epoxy coating; trimellitate alkyd epoxy coating; adipate alkyd epoxy coating; bisphenol A epoxy alkyd coating
- IT Epoxy resins, uses
  - RL: TEM (Technical or engineered material use); USES (Uses) (alkyd-, coatings, aqueous-dispersion, with good storage stability and solvent dilutability, for vertical surfaces)
- IT Coating materials

(anticorrosive, dispersion, water-thinned, alkyd-epoxy, with good storage stability and solvent dilutability, for vertical surfaces)

IT Alkyd resins
RL: TEM (Technical or engineered material use); USES (Uses)

(epoxy, coatings, aqueous-dispersion, with good storage stability and solvent dilutability, for vertical surfaces)

IT 149368-95-6P 149369-00-6P

RL: PREP (Preparation)

(manufacture of, as anticorrosive coatings for vertical surfaces)
IT 25068-38-6DP, Epikote 1007, reaction products with epoxy-modified alkyd
resins 68647-08-5P 111308-82-8P 126902-73-6P

RL: PREP (Preparation)

(manufacture of, for aqueous dispersion anticorrosive coatings with good storage

stability and solvent dilutability)

IT 149368-95-6P 149369-00-6P

RL: PREP (Preparation)

(manufacture of, as anticorrosive coatings for vertical surfaces)

RN 149368-95-6 HCAPLUS

CN Hexanedioic acid, polymer with (chloromethyl)oxirane, 1,3-dihydro-1,3-dioxo-5-isobenzofurancarboxylic acid, 2,2-dimethyl-1,3-propanediol, 2-ethyl-2-(hydroxymethyl)-1,3-propanediol, formaldehyde, 4,4'-(1-methylethylidene)bis[phenol] and 1,3,5-triazine-2,4,6-triamine, dodecanoate (ester) (9CI) (CA INDEX NAME)

CM 1

CRN 143-07-7 CMF C12 H24 O2

 $HO_2C^-$  (CH<sub>2</sub>)<sub>10</sub>-Me

CM 2

CRN 163418-53-9

CMF (C15 H16 O2 . C9 H4 O5 . C6 H14 O3 . C6 H10 O4 . C5 H12 O2 . C3 H6 N6 . C3 H5 Cl O . C H2 O)  $\times$ 

CCI PMS

CM 3

CRN 552-30-7 CMF C9 H4 O5

CM 4

CRN 126-30-7 CMF C5 H12 O2

$$\begin{array}{c} \text{Me} \\ \mid \\ \text{HO-CH}_2\text{-C-CH}_2\text{-OH} \\ \mid \\ \text{Me} \end{array}$$

CRN 124-04-9 CMF C6 H10 O4

$$_{\rm HO_2C^-}$$
 (CH<sub>2</sub>)<sub>4</sub>-CO<sub>2</sub>H

CM 6

CRN 108-78-1 CMF C3 H6 N6

CM 7

CRN 106-89-8 CMF C3 H5 C1 O

CM 8

CRN 80-05-7 CMF C15 H16 O2

TRUONG 10/642873 8/10/04 Page 61

CM 9

CRN 77-99-6 CMF C6 H14 O3

$$\begin{array}{c} \text{CH}_2-\text{OH} \\ | \\ \text{HO-CH}_2-\text{C-Et} \\ | \\ \text{CH}_2-\text{OH} \end{array}$$

CM 10

CRN 50-00-0 CMF C H2 O

 $H_2C = 0$ 

RN 149369-00-6 HCAPLUS

CN Hexanedioic acid, polymer with (chloromethyl)oxirane, 1,3-dihydro-1,3-dioxo-5-isobenzofurancarboxylic acid, 2,2-dimethyl-1,3-propanediol, 2-ethyl-2-(hydroxymethyl)-1,3-propanediol, formaldehyde, 4,4'-(1-methylethylidene)bis[phenol] and 1,3,5-triazine-2,4,6-triamine, (Z,Z)-9,12-octadecadienoate (ester) (9CI) (CA INDEX NAME)

CM 1

CRN 60-33-3 CMF C18 H32 O2

Double bond geometry as shown.

$$HO_2C$$
 (CH<sub>2</sub>)<sub>7</sub>  $Z$   $Z$  (CH<sub>2</sub>)<sub>4</sub>  $Me$ 

CM 2

CRN 163418-53-9

CMF (C15 H16 O2 . C9 H4 O5 . C6 H14 O3 . C6 H10 O4 . C5 H12 O2 . C3 H6 N6 . C3 H5 C1 O . C H2 O) x

CCI PMS

CM 3

CRN 552-30-7 CMF C9 H4 O5

CRN 126-30-7 CMF C5 H12 O2

$$\begin{array}{c} \text{Me} \\ \mid \\ \text{HO-CH}_2\text{-C-CH}_2\text{-OH} \\ \mid \\ \text{Me} \end{array}$$

CM 5

CRN 124-04-9 CMF C6 H10 O4

 ${\rm HO_2C-}$  (CH<sub>2</sub>)<sub>4</sub>-CO<sub>2</sub>H

CM 6

CRN 108-78-1 CMF C3 H6 N6

CM 7

CRN 106-89-8 CMF C3 H5 Cl O

CRN 80-05-7 CMF C15 H16 O2

CM 9

77-99-6 CRN CMF C6 H14 O3

$$\begin{array}{c} \text{CH}_2-\text{OH} \\ | \\ \text{HO-CH}_2-\text{C-Et} \\ | \\ \text{CH}_2-\text{OH} \end{array}$$

CM10

CRN 50-00-0 CMF C H2 O

 $H_2C = 0$ 

L76 ANSWER 19 OF 27 HCAPLUS COPYRIGHT 2004 ACS on STN

1991:658415 HCAPLUS AN

DN 115:258415

Entered STN: 14 Dec 1991 ED

ΤI Topcoat compositions for automobiles

Enokida, Yutaka; Tsunoda, Takeshi; Nakamichi, Toshihiko Nippon Oil and Fats Co., Ltd., Japan ΙN

PΑ

Jpn. Kokai Tokkyo Koho, 10 pp. SO

CODEN: JKXXAF

DΤ Patent

LA Japanese

IC ICM C09D167-02

ICI C09D167-02, C09D161-20

CC 42-10 (Coatings, Inks, and Related Products)

Section cross-reference(s): 35

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE		
PΙ	JP 03126778	A2	19910529	JP 1989-265961	19891012		

PRAI JP 1989-265961

19891012

CLASS

AB The title compns. comprise polyesters with acid value <30 mg KOH/g, OH value 30-200 mg KOH/g, surface tension 33-39 dyne/cm, and softening point 5-45° and melamine resins at mix ratio 60:40 - 90:10. Thus, a bonderized steel sheet with a 20 μm electrophoretic coat and a 30 μm intermediate coat was coated with a mixture of adipic acid-isophthalic acid-neopentyl glycol-trimethylolpropane copolymer isononanoate (OH value 120, acid value 9.6, surface tension 36.1 dyne/cm, and softening point 36°) 42.0, TiO2 29.0, U-Van 20S 18.0, Modaflow 0.5, Solvesso 100 8.5, and BuOH 2.0 parts to 30 μm thickness and baked at 140°. The coated specimen with good appearance showed no change when immersed in H2O at 40° for 240 h, pencil hardness F, good recoatability, and good flexibility while controls containing polyesters with off-specification properties or containing polyesters and melamine resins at off-specification ratios showed degradation in some of the properties.

ST polyester melamine topcoat automobile

IT Epoxides

IT

RL: USES (Uses)

(C12-14-alkyl, polymers, with diacids and diols and melamine resins, topcoats for automobiles, AOEX-24)

IT Coating materials

(topcoats, blends of polyesters and melamine resins, for automobiles) 77-99-6DP, Trimethylolpropane, polymers with adipic acid and AOEX 24 and isophthalic acid and trimethylolpropane and U-Van 20S 121-91-5DP, 1,3-Benzenedicarboxylic acid, polymers with adipic acid and AOEX 24 and neopentyl glycol and trimethylpropane and U-Van 20S 124-04-9DP, Hexanedioic acid, polymers with AOEX 24 and isophthalic acid and neopentyl glycol and trimethylolpropane and U-Van 20S 126-30-7DP, Neopentyl glycol, polymers with adipic acid and AOEX 24 and isophthalic acid and trimethylolpropane and U-Van 20S 137223-61-1P 137621-67-1P 137621-68-2P 137621-69-3P

RL: PREP (Preparation)

(preparation of, topcoats for automobiles)

IT 137223-61-1P

RL: PREP (Preparation)

(preparation of, topcoats for automobiles)

RN 137223-61-1 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, polymer with 2,2-dimethyl-1,3-propanediol, 2-ethyl-2-(hydroxymethyl)-1,3-propanediol and hexanedioic acid, dodecanoate, polymer with formaldehyde and 1,3,5-triazine-2,4,6-triamine (9CI) (CA INDEX NAME)

CM 1

CRN 108-78-1 CMF C3 H6 N6

CRN 50-00-0 CMF C H2 O

 $H_2C = O$ 

CM 3

CRN 127646-43-9

CMF C12 H24 O2 . x (C8 H6 O4 . C6 H14 O3 . C6 H10 O4 . C5 H12 O2) x

CM 4

CRN 143-07-7 CMF C12 H24 O2

 $HO_2C^-$  (CH<sub>2</sub>)<sub>10</sub>-Me

CM 5

CRN 25950-34-9

CMF (C8 H6 O4 . C6 H14 O3 . C6 H10 O4 . C5 H12 O2) x

CCI PMS

CM 6

CRN 126-30-7 CMF C5 H12 O2

$$\begin{array}{c} \text{Me} \\ \mid \\ \text{HO-CH}_2\text{--C-CH}_2\text{--OH} \\ \mid \\ \text{Me} \end{array}$$

CM 7

CRN 124-04-9 CMF C6 H10 O4  $HO_2C-(CH_2)_4-CO_2H$ 

CM 8

CRN 121-91-5 CMF C8 H6 O4

CM 9

CRN 77-99-6 CMF C6 H14 O3

$$\begin{array}{c} \text{CH}_2-\text{OH} \\ | \\ \text{HO-CH}_2-\text{C-Et} \\ | \\ \text{CH}_2-\text{OH} \end{array}$$

L76 ANSWER 20 OF 27 HCAPLUS COPYRIGHT 2004 ACS on STN

AN 1991:610299 HCAPLUS

DN 115:210299

ED Entered STN: 15 Nov 1991

TI Aqueous dispersions of crosslinked polymer microparticles for use in coatings

IN Hille, Hans Dieter; Massone, Matthias

PA Bollig und Kemper K.-G., Germany

SO Ger. Offen., 9 pp.

CODEN: GWXXBX

DT Patent

LA German

IC ICM C08J003-03

ICS C08L067-02; C08L075-04; C08L033-14; C08L061-20; C09D005-02; C09D167-02; C09D175-04; C09D133-14; C09D005-38

ICI C09D167-02, C09D161-20; C09D175-04, C09D161-20; C09D133-14, C09D161-20
CC 42-7 (Coatings, Inks, and Related Products)

FAN.CNT 1

17114.	PATENT NO.					KIN	D	DATE			APPL	DATE					
ΡI	DE 3940316				A1	_	1991	DE 1989-3940316						19891206			
	CA 2070807			AA		1991	0607		CA 1990-2070807				19901204				
	CA	2070	807			С		2001	0227								
	WO	WO 9108269				A1		19910613		WO 1990-EP2096			19901204	204			
		W:	ΑU,	BR,	CA,	FI,	KR,	NO,	SU,	US							
		RW:	ΑT,	BE,	CH,	DE,	DK,	ES,	FR,	GB,	GR,	ΙT,	LU,	NL,	SE		

```
TRUONG 10/642873 8/10/04
                              Page 67
     AU 9168943
                                 19910626
                                             AU 1991-68943
                                                                    19901204
                          A1
     AU 652729
                          B2
                                 19940908
     BR 9007898
                                             BR 1990-7898
                                                                    19901204
                          Α
                                 19920915
     EP 502934
                                             EP 1991-900211
                          Α1
                                 19920916
                                                                    19901204
     EP 502934
                          В1
                                 19941102
         R: AT, BE, CH, DE, DK, ES, FR, GB, IT, LI, NL, SE
                                 19950116
                                             ES 1991-900211
                                                                    19901204
     ES 2064076
                          т3
     RU 2104290
                          C1
                                 19980210
                                             RU 1990-5052553
                                                                    19901204
     NO 9202228
                                             NO 1992-2228
                                                                    19920605
                          Α
                                 19920730
     US 5412023
                          Α
                                 19950502
                                            US 1992-910080
                                                                    19920710
PRAI DE 1989-3940316
                          Α
                                 19891206
     WO 1990-EP2096
                          Α
                                 19901204
CLASS
 PATENT NO.
                 CLASS PATENT FAMILY CLASSIFICATION CODES
                        ______
 DE 3940316
                 ICM
                        C08J003-03
                        C08L067-02; C08L075-04; C08L033-14; C08L061-20;
                 ICS
                        C09D005-02; C09D167-02; C09D175-04; C09D133-14;
                        C09D005-38
                        C09D167-02, C09D161-20; C09D175-04, C09D161-20; C09D133-14, C09D161-20
                 ICI
     The title dispersions are prepared by heating polyester, polyurethane, or
AB
     polyacrylate polyols (acid number 5-268) bearing sufficient ionic groups to
     form stable aqueous dispersions, optionally dissolved in organic solvents, with
     reactive aminoplasts in aqueous media. Adding 687.5 parts 5% aqueous
Me2NCH2CH2OH
     over 10 min to 1050 parts 77% MIBK solution of a polyester (acid number 34)
     prepared from trimethylolpropane 670, 1,6-hexanediol 354, isomerized soya
     acids 1400, phthalic anhydride 592, and trimellitic anhydride 384 parts
     and 495 parts BuOH solution of melamine resin stirred at 40-48°,
     adding 1457.5 parts H2O over 15 min at 48-68°, heating at
     85° for 3 h, neutralizing with Me2NCH2CH2OH, and distilling MIBK gave a
     34% resin dispersion with pH 8.37 and DIN viscosity 21 s. A mixture of this
     composition, an acrylic polymer thickener, and nonleafing Al bronze was sprayed
     on steel, dried at 50°, topcoated with a polyacrylate-melamine resin composition, and baked at 130° to give a metal-effect coating.
     metal effect coating binder; binder coating microparticle emulsion;
ST
     polyester crosslinked emulsion coating; melamine resin crosslinker
     coating; emulsion polymer microparticle coating
ΙT
     Crosslinking agents
        (melamine resin, for polyester microparticle emulsions for coatings)
ΙT
     Coating materials
        (metal-effect, water-thinned, crosslinked polyester microparticle
        emulsions for use in)
IT
     Fatty acids, compounds
     RL: USES (Uses)
        (polyunsatd., conjugated, maleated, polymers with polyols and
        dicarboxylic acids, for crosslinked microparticle emulsions for
        coatings)
TΤ
     Fatty acids, esters
     RL: USES (Uses)
        (soya, esters, with alkyd resins, in microparticle emulsions for use in
        coatings)
ΙT
     108-31-6D, 2,5-Furandione, reaction products with unsatd. fatty acids,
     polymers with hexanediol, isophthalic acid and trimellitic anhydride,
     reaction products with melamine resins 121-91-5D, 1,3-
     Benzenedicarboxylic acid, polymers with hexanediol, trimellitic anhydride
     and maleated unsatd. fatty acids, reaction products with melamine resins
     552-30-7D, polymers with hexanediol, isophthalic acid and maleated unsatd.
```

fatty acids, reaction products with melamine resins 629-11-8D, 1,6-Hexanediol, polymers with isophthalic acid, trimellitic anhydride and maleated unsatd. fatty acids, reaction products with melamine resins 9003-08-1D, Formaldehyde-melamine copolymer, reaction products with maleated fatty acid-based polyesters 136837-45-1 136837-47-3D, esters with isomerized soya fatty acids 136837-48-4 RL: USES (Uses) (crosslinked microparticle emulsions, for use in coatings) IT 136894-87-6P RL: PREP (Preparation) (manufacture of, for crosslinked microparticle emulsions for coatings) 9003-08-1D, Formaldehyde-melamine copolymer, reaction IT products with maleated fatty acid-based polyesters RL: USES (Uses) (crosslinked microparticle emulsions, for use in coatings) 9003-08-1 HCAPLUS RN 1,3,5-Triazine-2,4,6-triamine, polymer with formaldehyde (9CI) (CA INDEX CN NAME) CM 1 CRN 108-78-1 CMF C3 H6 N6 NH2 H<sub>2</sub>N 2 CM 50-00-0 CRN C H2 O CMF  $H_2C = 0$ 

L76 ANSWER 21 OF 27 HCAPLUS COPYRIGHT 2004 ACS on STN 1990:613128 HCAPLUS AN 113:213128 DN Entered STN: 08 Dec 1990 ED Polymeric resins from renewable resources. II. Synthesis and ΤI characterization of flame-retardant prepolymers from cardanol Pillai, C. K. S.; Prasad, V. S.; Sudha, J. D.; Bera, S. C.; Menon, A. R. ΑU Reg. Res. Lab., CSIR, Trivandrum, 695019, India CS SO Journal of Applied Polymer Science (1990), 41(9-10), 2487-501 CODEN: JAPNAB; ISSN: 0021-8995 DT Journal LA English CC 37-3 (Plastics Manufacture and Processing)

Phosphorylated cardanol (I) prepolymers (II) were obtained by simultaneous AB phosphorylation and oligomerization of I, an unsatd. pentadecylphenol extracted from the nuts of the plant Anacardium occidentale. Although gel permeation chromatog. showed the presence of only oligomeric species in the system, the high viscosity of (0.35-0.45) + 106 cps was due to involvement of H bonding. IR spectra of II gave bands at 1030 cm-1 and 1240 cm-1, indicating P-O-C linkages. The decrease in iodine value and the absence of vinyl IR bands at 895 cm-1 and 907 cm-1 indicated oligomerization. NMR spectra of II showed partial loss of unsatn. carbonium ion-initiated mechanism is indicated for the oligomerization. II was highly reactive with aldehydes, amines, and isocyanates. Highly insol. and infusible thermoset products could be obtained. TGA studies showed a 2-stage decomposition with improved thermal stability above 500° for II compared to I-HCHO (novolac-type) resin (III). cured with hexamethylenetetramine (IV) had tensile strength of 16.9-21.5 MN/m2 and impact strength of 1.63-2.04 J compared to 24-48 MN/m2 and 1.35 J, resp., for phenol-HCHO resin prepared under similar conditions. The lap shear strength of II-bonded wood pieces was  $400 \pm 10 \text{ N/cm2}$  compared to  $60 \pm 10$  N/cm2 for III, indicating that phosphorylation improves bonding characteristics. With a P content of 7.9%, II showed good fire-retardant properties. Vertical burning studies showed no propagation of fire or any afterglow. IV-cured II had a limiting O-index value of 35, which increased to 42.0 on bromination. ST cardanol phosphorylation oligomerization mechanism; crosslinked cardanol polymer fire resistance ΙT Cashew (nutshell liquid, phosphorylated, crosslinked polymers containing, preparation and fire resistance of) ΙT Phosphorylation, synthetic (of cardanol, with orthophosphoric acid, with simultaneous oligomerization, mechanism of) ΙT Adhesives (phosphorylated cardanol homopolymers, for wood) ΙT Heat-resistant materials (phosphorylated cardanol-containing crosslinked polymers) IT Fire-resistant materials (phosphorylated cardanol-containing crosslinked polymers, bromination effect on) ΙT Phenolic resins, compounds RL: PEP (Physical, engineering or chemical process); PROC (Process) (cardanol-based, phosphorylated, preparation and crosslinking of) IT Polymerization (oligomerization, of cardanol, with simultaneous phosphorylation, mechanism of) TΤ Phenolic resins, preparation Urethane polymers, preparation RL: SPN (Synthetic preparation); PREP (Preparation) (phosphorylated cardanol-based, preparation and fire resistance of crosslinked) ΙT 37330-39-5, Cardanol RL: RCT (Reactant); RACT (Reactant or reagent) (oligomerization of, with simultaneous phosphorylation, mechanism of) ΤΨ 7664-38-2, Phosphoric acid, reactions RL: RCT (Reactant); RACT (Reactant or reagent)

(phosphorylation by, in simultaneous cardanol oligomerization,

IT 130285-13-1DP, Cardanol homopolymer, phosphorylated
RL: RCT (Reactant); SPN (Synthetic preparation); PREP

mechanism of)

```
(Preparation); RACT (Reactant or reagent)
        (preparation and characterization and crosslinking of)
IT
     37311-73-2DP, Cardanol-formaldehyde copolymer, phosphorylated
     125433-24-1DP, phosphorylated
                                    130572-33-7DP, phosphorylated
     130572-34-8DP, phosphorylated
     RL: PEP (Physical, engineering or chemical process); SPN (Synthetic
     preparation); PREP (Preparation); PROC (Process)
        (preparation and characterization of)
IT
     37311-73-2P, Cardanol-formaldehyde copolymer
     RL: PEP (Physical, engineering or chemical process); SPN (Synthetic
     preparation); PREP (Preparation); PROC (Process)
        (preparation and crosslinking of)
ΙT
     9062-06-0DP, phosphorylated, brominated
                                                130572-32-6P
     RL: SPN (Synthetic preparation); PREP (Preparation)
        (preparation and fire resistance of)
ΙT
     130572-34-8DP, phosphorylated
     RL: PEP (Physical, engineering or chemical process); SPN (Synthetic
     preparation); PREP (Preparation); PROC (Process)
        (preparation and characterization of)
     130572-34-8 <u>HCAPLUS</u>
RN
     1,3,5-Triazine-2,4,6-triamine, polymer with Cardanol (9CI)
CN
                                                                   (CA INDEX
     NAME)
     CM
          37330-39-5
     CRN
     CMF
          Unspecified
     CCI
         MAN
*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***
     CM
          2
     CRN 108-78-1
     CMF C3 H6 N6
       NH2
             NH<sub>2</sub>
     ANSWER 22 OF 27 HCAPLUS COPYRIGHT 2004 ACS on STN
L76
     1990:236973 HCAPLUS
ΑN
DN
     112:236973
ED
     Entered STN: 23 Jun 1990
ΤI
     High-solids acrylic-polyester coatings containing hydroxy ester reactive
     diluents
IN
     Toman, Perry A.
PA
     Glidden Co., USA
SO
     U.S., 8 pp. Cont.-in-part of U.S. 4,182,523.
     CODEN: USXXAM
     Patent
DT
LA
     English
IC
     ICM C08L063-00
```

NCL

NCL 525107000

CC 42-10 (Coatings, Inks, and Related Products)

525107000

FAN.CNT 2

	V					
	PATENT NO.		KIND	DATE	APPLICATION NO.	DATE
ΡI	US 4877838		Α	19891031	US 1988-289257	19881223
	US 4812523		Α	19890314	US 1988-147440	19880125
PRAI US 1988-147440			19880125			
CLAS	S					
PAT	ENT NO.	CLASS	PATENT	FAMILY CLAS	SSIFICATION CODES	
US 4877838 ICM			C08L063	-00		

AB Reaction products of epoxides and carboxylic acids are useful as reactive diluents for aminoplast- or glycoluril-crosslinked, high-solids coatings containing hydroxy acrylic polymers and hydroxy polyesters. Thus, a composition

containing 40:20:2.6:37.6 Bu acrylate-2-hydroxyethyl acrylate-methacrylic acid-styrene copolymer 21.1, BuOAc 16.81, TiO2 100, 26.45:30.06:54.79:1.5 adipic acid-isophthalic acid-neopentyl glycol-trimethylolpropane copolymer 22.86, 379.6:206.4 Cardura E-dodecenylsuccinic anhydride reaction product (I, acid number 3.8) 16.06, Cymel 303 crosslinker 35, thixatropic gel 10.26, surfactant 0.55, ethylene glycol mono-Bu ether acetate 0.55, BuOH 5.6, and dinonylnaphthalene sulfonic acid 1 g exhibited lower viscosity and similar film properties in salt spray, water fog, and accelerated weathering tests, compared with a similar composition not containing I.

- ST hydroxy ester reactive diluent coating; high solids acrylic polyester coating; epoxide carboxlate reactive diluent coating; dodecenyl succinate epoxide reactive diluent coating; aminoplast crosslinked acrylic polyester coating; glycoluril crosslinked acrylic polyester coating; styrene acrylic polyester coating; acrylate copolymer polyster coating; hydroxyethyl acrylate copolymer polyester coating; methacrylic polyester coating; polyisophthalate acrylic high solids coating; polyadipate acrylic high solids coating; neopentyl glycol polyester acrylic coating; trimethylolpropane polyester acrylic coating
- IT Fatty acids, polymers

RL: PROC (Process)

(C18-unsatd., dimers, reaction products, with epoxides, manufacture of, for reactive diluents for high-solids hydroxy group-containing acrylic-polyester coatings)

IT Fatty acids, esters

RL: USES (Uses)

(C9-11-branched, esters, with glycidol, reaction products with dodecenyl succinic anhydride, manufacture of, for reactive diluents for high-solids hydroxy group-containing acrylic-polyester coatings)

IT Coating materials

(high-solids, acrylic-polyester, containing hydroxy ester reactive diluents and amino crosslinking agents)

IT Fatty acids, polymers

RL: PROC (Process)

(unsatd., dimers, reaction products, with epoxides, manufacture of, for reactive diluents for high-solids hydroxy group-containing acrylic-polyester coatings)

IT 25950-34-9, Adipic acid-isophthalic acid-neopentyl glycol-

trimethylolpropane copolymer

RL: TEM (Technical or engineered material use); USES (Uses) (coatings, high-solids, containing hydroxy group-containing acrylic polymers and hydroxy ester reactive diluents and amino crosslinking agents)

IT 27553-55-5, Butyl acrylate-2-hydroxyethyl methacrylate-methacrylic

```
acid-styrene copolymer
     RL: TEM (Technical or engineered material use); USES (Uses)
        (coatings, high-solids, containing polyesters and hydroxy ester reactive
        diluents and amino crosslinking agents)
ΙT
                   127211-01-2P
                                  127331-32-2P 127483-38-9P
     127211-00-1P
     RL: TEM (Technical or engineered material use); PREP (Preparation)
     ; USES (Uses)
        (coatings, high-solids, manufacture of)
TΤ
     75-56-9, reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (esterification by, of dodecenyl succinic anhydride)
     143-07-7, Lauric acid, reactions
ΙT
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (esterification of, by diglycidyl dimethylmethylene glutarate)
TT
     17557-23-2
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (esterification of, by lauric acid)
     25377-73-5, Dodecenylsuccinic anhydride
TΤ
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (esterification of, by propylene oxide)
ΙT
     75-56-9DP, esters with fatty acid dimers
                                                9058-77-9P
     RL: PREP (Preparation)
        (manufacture of, for high-solids hydroxy group-containing acrylic-polyester
        coatings)
ΤТ
     9080-23-3DP, Empol 1018, reaction product with glycidyl neodecanoate
     25377-73-5DP, Dodecenylsuccinic anhydride, reaction products with glycidyl
     esters of branched fatty acids 26761-45-5DP, Cardura E-10, reaction
     products with fatty acid dimers and trimers 59111-86-3P, Bisphenol
     A-epichlorohydrin copolymer laurate 59978-87-9P
                                                          127176-35-6P
                                                 127289-99-0P
     127210-99-5P 127289-36-5P 127289-55-8P
                                                                 127290-22-6DP,
     Pripol 1009, reaction product with glycidyl neodecanoate
                                                                 127304-09-0P
     RL: PREP (Preparation)
        (manufacture of, for reactive diluents for high-solids hydroxy
group-containing
        acrylic-polyester coatings)
ΙT
     26761-45-5, Cardura E-10
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with diacids and hydroxy acids)
     106-14-9 123-99-9, Nonanedioic acid, reactions
TΤ
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with glycidyl neodecanoate)
     100639-41-6, Diglycidyl 2,2-dimethyl-4-methylene glutarate
IT
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with lauric acid)
TΨ
     127483-38-9P
     RL: TEM (Technical or engineered material use); PREP (Preparation)
     ; USES (Uses)
        (coatings, high-solids, manufacture of)
    -127483=38-9 HCAPLUS
ŔŊ
     1,3-Benzenedicarboxylic acid, polymer with bis(oxiranylmethyl)
CN
     2,2-dimethyl-4-methylenepentanedioate homopolymer dodecanoate, butyl
     2-propenoate, 2,2-dimethyl-1,3-propanediol, ethenylbenzene,
     2-ethyl-2-(hydroxymethyl)-1,3-propanediol, formaldehyde, hexanedioic acid,
     2-hydroxyethyl 2-propenoate, 2-methyl-2-propenoic acid and
     1,3,5-triazine-2,4,6-triamine (9CI) (CA INDEX NAME)
     _CM____1--
```

CMF C5 H8 O3

$$\begin{array}{c} & \circ \\ || \\ \text{HO-CH}_2\text{-CH}_2\text{-O-C-CH-----} \text{CH}_2 \end{array}$$

CM 2

CRN 141-32-2 CMF C7 H12 O2

CM 3

CRN 126-30-7 CMF C5 H12 O2

$$\begin{array}{c} \text{Me} \\ \mid \\ \text{HO-CH}_2\text{--C-CH}_2\text{--OH} \\ \mid \\ \text{Me} \end{array}$$

CM 4

CRN 124-04-9 CMF C6 H10 O4

$$HO_2C-(CH_2)_4-CO_2H$$

CM 5

CRN 121-91-5 CMF C8 H6 O4

CM 6

CRN 108-78-1 CMF C3 H6 N6

CM 7

CRN 100-42-5 CMF C8 H8

 $_{\rm H2C} = _{\rm CH-Ph}$ 

CM 8

CRN 79-41-4 CMF C4 H6 O2

CM 9

CRN 77-99-6 CMF C6 H14 O3

$$\begin{array}{c} \text{CH}_2-\text{OH} \\ | \\ \text{HO-CH}_2-\text{C-Et} \\ | \\ \text{CH}_2-\text{OH} \end{array}$$

CM 10

CRN 50-00-0 CMF C H2 O

 $H_2C = O$ 

CM 11

CRN 127289-55-8

CMF (C14 H20 O6)x . x C12 H24 O2

CM 12

CRN 143-07-7 CMF C12 H24 O2

 $HO_2C^-$  (CH<sub>2</sub>)<sub>10</sub>-Me

CM 13

CRN 199542-57-9 CMF (C14 H20 O6)x

CCI PMS

CM 14

CRN 100639-41-6 CMF C14 H20 O6

L76 ANSWER 23 OF 27 HCAPLUS COPYRIGHT 2004 ACS on STN

AN 1988:512199 HCAPLUS

DN 109:112199

ED Entered STN: 01 Oct 1988

TI Formation of chip-resistant coatings

IN Kasari, Akira; Iwase, Osamu; Osumimoto, Hiroshi; Udagawa, Takashi; Fujii, Yasuhiro

PA Kansai Paint Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 16 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM B05D007-14

ICS B05D003-02; B05D005-00; B05D007-24

CC 42-10 (Coatings, Inks, and Related Products)

Section cross-reference(s): 55

FAN.CNT 1

IIIII. CIVI I					
PATENT NO.	KIND	DATE	APPLICATION NO.	DATE	
PI JP 63080884	A2	19880411	JP 1986-225360	19860924	
JP 2512907	B2	19960703			
PRAI JP 1986-225360		19860924			
CLASS					

PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES

JP 63080884 ICM B05D007-14

ICS B05D003-02; B05D005-00; B05D007-24

Chip-resistant coatings are manufactured by forming a middle coat having tensile breaking elongation >10% and strength at break >20 kg/cm2 with a thermosetting organic solution-type polyester resin, adjusting the viscosity of the above polyester resin to 5-500 P  $(20^{\circ})$  after coating and before curing, forming a topcoating having tensile breaking elongation <40% and strength at break >100 kg/cm2 with thermosetting coating material having equal or smaller surface tension than that for the above polyester. Compared with the above polyester coating, the topcoats has a smaller tensile break strength elongation (absolute value >5) and a larger strength at breaking (≥80 kg/cm2). Thus, phosphated steel was primered with Elecron 9200 (an epoxy polyamide cationic electrodeposition coating), coated with a second coat containing a short soybean oil-modified alkyd resin, a melamine resin, Ti white, and baryta, coated with barrier coat having elongation 60%, strength at break 100 kg/cm2, and surface tension 32.0 and containing hexahydrophthalic acid-adipic acid-trimethylolpropane-1,6hexanediol copolymer, U-Van 20SE-60, and pigments, topcoated with metallic coating having elongation 20%, strength at break 300 kg/cm2, and surface tension 30.8 and containing styrene-Et methacrylate-Bu methacrylatehydroxyethyl methacrylate-acrylic acid copolymer, U-Van 20 SE-60, and pigments, and coated with clear coat containing acrylic resins and U-Van 20SE-760.

ST chip resistant multilayer coating; alkyd resin soybean modified coating; metallic acrylic topcoat chip resistant; polyhexahydrophthalate coating chip resistant; polyadipate coating chip resistant; trimethylolpropane polyester coating chip resistant; hexanediol polyester coating chip resistant; styrene acrylic coating chip resistant; methacrylate copolymer coating chip resistant

IT Coating materials

(chip-resistant, multilayer, acrylic-polyester, on steel)

IT Fatty acids, esters

RL: USES (Uses)

(coco, esters, with polyesters, chip-resistant multilayer coatings containing melamine resins and, on steel)

IT Fatty acids, esters

RL: USES (Uses)

(soya, alkyd resins, chip-resistant multilayer coatings containing melamine resins and, on steel)

IT Fatty acids, esters

RL: USES (Uses)

(tall-oil, esters, with polyesters, chip-resistant multilayer coatings containing melamine resins and, on steel)

IT 85-44-9D, Phthalic anhydride, alkyd resin derivs., soybean-oil-modified 100-21-0D, Terephthalic acid, alkyd resin derivs., soybean-oil-modified RL: USES (Uses)

(chip-resistant multilayer coatings containing melamine resins and, for steel)

IT 12597-69-2

RL: MSC (Miscellaneous)

(coating materials, chip-resistant, multilayer, acrylic-polyester, on steel)

IT 116363-10-1P

RL: PREP (Preparation)

(manufacture of, as clear toplayer in chip-resistant coatings on steel)
IT 110871-08-4P, Acrylic acid-butyl methacrylate-ethyl methacrylatehydroxyethyl methacrylate-styrene-U-Van 20SE60 copolymer 116322-54-4P,

Adipic acid-1,6-hexanediol-hexahydrophthalic acid-trimethylolpropane-U-Van 20SE60 copolymer 116322-55-5DP, Adipic acid-1,6-hexanediolhexahydrophthalic acid-neopentyl glycol-trimethylolpropane-U-Van 20SE 60 copolymer, coconut-oil fatty acid-modified 116322-56-6P, Adipic acid-1,3-Butanediol-1,6-Hexanediol-sebacic acid-trimethylolpropane-U-Van 20SE60 copolymer RL: PREP (Preparation) (manufacture of, as intermediate layer in chip-resistant coatings, for steel) 116334-57-7P ΙT RL: PREP (Preparation) (manufacture of, as intermediate metallic layer in chip-resistant coatings, for steel) IT 116322-57-7DP, coconut oil fatty acid-modified 116322-58-8P RL: PREP (Preparation) (manufacture of, as pigmented intermediate layer in chip-resistant coatings on steel) 116322-55-5DP, Adipic acid-1,6-hexanediol-hexahydrophthalic IT acid-neopentyl glycol-trimethylolpropane-U-Van 20SE 60 copolymer , coconut-oil fatty acid-modified RL: PREP (Preparation) (manufacture of, as intermediate layer in chip-resistant coatings, for steel) 116322-55-5 HCAPLUS\_ RN 1,2-Cyclohexanedicarboxylic acid, polymer with 2,2-dimethyl-1,3-CN propanediol, 2-ethyl-2-(hydroxymethyl)-1,3-propanediol, formaldehyde, hexanedioic acid, 1,6-hexanediol and 1,3,5-triazine-2,4,6-triamine (9CI) (CA INDEX NAME) CM-CRN 1687-30-5 CMF C8 H12 O4 CO<sub>2</sub>H CO2H 2 CM CRN 629-11-8 CMF C6 H14 O2  $^{\rm HO-}$  (CH<sub>2</sub>)<sub>6</sub> $^{\rm -}$  OH 3 CM CRN 126-30-7

CMF C5 H12 O2

CM 4

CRN 124-04-9 CMF C6 H10 O4

 $HO_2C-(CH_2)_4-CO_2H$ 

CM 5

CRN 108-78-1 CMF C3 H6 N6

CM 6

CRN 77-99-6 CMF C6 H14 O3

$$\begin{array}{c} \text{CH}_2-\text{OH} \\ | \\ \text{HO-CH}_2-\text{C-Et} \\ | \\ \text{CH}_2-\text{OH} \end{array}$$

CM 7

CRN 50-00-0 CMF C H2 O

 $H_2C = O$ 

L76 ANSWER 24 OF 27 HCAPLUS COPYRIGHT 2004 ACS on STN AN 1987:158012 HCAPLUS

```
106:158012
DN
ED
     Entered STN: 15 May 1987
     Synthesis and some properties of oligo(amino esters)
TΤ
     Mikhailova, T. B.; Semina, R. A.; Klygina, R. V.; Livshits, R. M.
ΑU
     GIPI LKP, Moscow, USSR
CS
     Lakokrasochnye Materialy i Ikh Primenenie (1987), (1), 8-9
SO
     CODEN: LAMAAD; ISSN: 0023-737X
DΤ
     Journal
LA
     Russian
     42-10 (Coatings, Inks, and Related Products)
CC
     A melamine-HCHO oligomer (GM-3) [9003-08-1], was modified with
AB
     tall-oil fatty acids and adipic acid [124-04-9] at 200° to give
     film formers having number-average mol. weight 2100-2600. The prepared film
formers
     gave cured coatings having better hardness, gloss, and color than the
     coatings prepared from their alkyd analogs. The best results were achieved
     at GM-3-(tall-oil fatty acids)-adipic acid mol ratio 1:2:0.5. A model
     study using stearic acid [57-11-4] instead of tall-oil fatty
     acids revealed that there was no polycondensation of GM-3 in the absence
     of adipic acid, but this reaction occurred in its presence.
     aminoplast fatty acid modification coating; melamine resin fatty acid
ST
     modification; adipic acid modification melamine resin; tall oil acid
     modification aminoplast
ΙT
     Coating materials
        (melamine resins modified with adipic and tall-oil fatty acids, preparation
        and properties of)
ΙT
     Crosslinking
        (of melamine resin coatings modified with adipic and tall-oil fatty
        acids)
     Fatty_acids, compounds
ΙT
     RL: SPN (Synthetic preparation); PREP (Preparation)
        (tall-oil, reaction products with adipic acid and melamine resins,
        coatings, preparation properties of)
It:
    .124-04-9DP, reaction products with melamine resins and tall-oil fatty
     acids 9003-08-1DP, Melamine-formaldehyde copolymer,
     reaction products with adipic and tall-oil fatty acids
     RL: SPN (Synthetic preparation); TEM (Technical or engineered
     material use); PREP (Preparation); USES (Uses)
        (coatings, preparation and properties of)
ΙT
     9003-08-1, Melamine-formaldehyde copolymer
     RL: USES (Uses)
        (oligomeric, N-isolation of GM3, with carboxylic acids)
     124-04-9, reactions
ΙT
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (N-isolation of melamine resins with tall-oil fatty acids and)
IT
     57-11-4, Stearic acid, uses and miscellaneous
     RL: USES (Uses)
        (N-isolation with, of melamine resins)
IT
     9003-08-1DP, Melamine-formaldehyde copolymer, reaction
     products with adipic and tall-oil fatty acids
     RL: SPN (Synthetic preparation); TEM (Technical or engineered
     material use); PREP (Preparation); USES (Uses)
        (coatings, preparation and properties of)
RN
     9003-08-1 HCAPLUS
CN
     1,3,5-Triazine-2,4,6-triamine, polymer with formaldehyde (9CI) (CA INDEX
     NAME)
     CM
          1
```

CRN 108-78-1 CMF C3 H6 N6

CM 2

CRN 50-00-0 CMF C H2 O

 $H_2C=0$ 

OS

AΒ

```
L76 ANSWER 25 OF 27 HCAPLUS COPYRIGHT 2004 ACS on STN
    1985:613396 HCAPLUS
AN
DN
    103:213396
    Entered STN: 28 Dec 1985
ED
    Modified lipase
ΤI
    Inada, Yuji
ΙN
    Bellex Corp., Japan
PA
SO
    Eur. Pat. Appl., 21 pp.
    CODEN: EPXXDW
DT
    Patent
LA
    English
    ICM C12N009-20
IC
    ICS C12N009-96
ICA C12P007-64; C12P033-00
    16-4 (Fermentation and Bioindustrial Chemistry)
    Section cross-reference(s): 7
FAN.CNT 1
                                        APPLICATION NO.
    PATENT NO.
                      KIND
                            DATE
                                                              DATE
                      ----
                             -----
                                        -----
                        A2
                                        EP 1985-300102
PΙ
    EP 149520
                             19850724
                                                              19850107
    EP 149520
                       A3
                             19871125
    EP 149520
                       В1
                             19910904
       R: CH, DE, FR, GB, IT, LI, NL
    JP 60156395
                       A2
                             19850816
                                        JP 1984-6129
                                                              19840117
    JP 05036029
                       В4
                             19930528
    US 4645741
                             19870224
                                         US 1984-687635
                                                              19841231
                       Α
PRAI JP 1984-6129
                             19840117
CLASS
              CLASS PATENT FAMILY CLASSIFICATION CODES
PATENT NO.
_____
EP 149520
               ICM
                      C12N009-20
               ICS
                      C12N009-96
                      C12P007-64; C12P033-00
               ICA
```

A chemical modified lipase is prepared which is modified with a straight chain comprising a substituted polyalkylene glycol having a hydrophobic group at

CASREACT 103:213396

```
a terminal end. The modified enzyme is soluble in both water and organic
     solvent, allowing for contact with organic solvents without enzyme
     deactivation. Thus, lipoprotein lipase [9004-02-8] from Pseudomonas
     fluorescens was reacted with 2,4-bis(methoxypolyoxyethylene)-6-chloro-s-
     triazine [72708-10-2] at 37° for 1 h. The enzyme was
     purified by conventional means to obtain a lipase preparation containing 52%
of its
     NH2 groups modified with the triazine derivative The modified lipase was
     added to a benzene [71-43-2] solution containing stearic acid [57-11-4
     ] and lauryl alc. [112-53-8] and the reaction was carried out at 37% for
     20 min. Modified lipase exhibited maximum lauryl stearate [5303-25-3]
     synthesis activity of 4.5 µmol/min/mg protein.
ST
     lipase polyalkylene glycol modification soly; org solvent soly modified
     lipase
IT
     Solvents
        (organic, lipase chemical modified derivative solubility in)
     57-11-4, biological studies
IT
     RL: BIOL (Biological study)
        (lauryl stearate preparation from lauryl alc. and, with lipoprotein lipase
        polyoxyethylenetriazine modified derivative)
     112-53-8
IT
     RL: BIOL (Biological study)
        (lauryl stearate preparation from stearate and, with lipoprotein lipase
        polyoxyethylenetriazine modified derivative)
     67-66-3, properties 71-43-2, properties 108-88-3, properties 127-18-4, properties
TΨ
                                                 71-55-6
                                                            79-01-6, properties
     RL: PRP (Properties)
        (lipase chemical modified derivative solubility in)
IT
     9001-62-1DP, polyalkylene glycol-modified derivs.
                                                         9004-02-8DP, derivs.
     25322-68-3DP, substituted derivs., reaction products with lipase
     RL: PREP (Preparation)
        (preparation and activity and solubility of)
                  92451-01-9P
                                99126-66-6P 99126-67-7P 99126-68-8P
IT
     80506-64-5P
     RL: PREP (Preparation)
        (preparation and lipase modification with)
     72708-10-2P
IT
     RL: PREP (Preparation)
        (preparation and lipoprotein lipase modification with)
TТ
                  99126-65-5P
     99126-64-4P
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP
     (Preparation); RACT (Reactant or reagent)
        (preparation and reaction with hydrazine hydrate)
TΤ
     67665-19-4P
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP
     (Preparation); RACT (Reactant or reagent)
        (preparation and reaction with hydroxysuccinimide)
ΙT
     58320-73-3P
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP
     (Preparation); RACT (Reactant or reagent)
        (preparation and reaction with potassium phthalimide)
TΤ
                  99126-69-9P
     86469-86-5P
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP
     (Preparation); RACT (Reactant or reagent)
        (preparation and reaction with sodium nitrite)
IT
     106-33-2P
    RL: PREP (Preparation)
        (preparation of, by trilaurin alcoholysis with lipase modified form)
IT
     5303-25-3P
     RL: PREP (Preparation)
```

```
(preparation of, from stearate and lauryl alc. with lipoprotein lipase
        polyoxyethylenetriazine modified derivative)
ΙT
     41005-65-6P
     RL: PREP (Preparation)
        (preparation of, in presence of lipase modified derivative)
TΤ
     111-82-0P
     RL: PREP (Preparation)
        (preparation of, with lipase modified form)
TΤ
     334-88-3
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with chloroacetic anhydride and monomethoxypolyethylene
        glycol)
TΨ
     538-24-9
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with ethanol in presence of lipase modified form)
ΤТ
     83-46-5
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with laurate)
TΨ
     623-42-7
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with laurate in presence of lipase modified form)
     105-36-2 118-48-9 45669-18-9
TΨ
    'RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with methoxypolyethylene glycol)
TΤ
     6066-82-6
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with methoxypolyethylene glycol acetic acid ether)
ΙT
     98-59-9
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with methoxypolyethylene glycol and chloroform and
        pyridine)
IT
     110-86-1, reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with methoxypolyethylene glycol and chloroform and tosyl
        chloride)
ΙT
     67-66-3, reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with methoxypolyethylene glycol and pyridine and tosyl
        chloride)
     108-77-0
IT
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with monomethoxypolyethylene glycol)
IT
     541-88-8
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with monomethoxypolyethylene glycol and diazomethane)
ΙT
     9004-02-8
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with polyoxyethylenetriazine derivative, enzyme solubility in
        relation to)
ΙT
     143-07-7, reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with sitosterol or methylbutyrate)
IT
     99143-92-7
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with tosylated methoxypolyethylene glycol)
IT
     9004-74-4
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with trichlorotriazine)
IT
     64-17-5, reactions
```

```
RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with trilaurin in presence of lipase modified form)
    ANSWER 26 OF 27 HCAPLUS COPYRIGHT 2004 ACS on STN
ΑN
     1984:176526 HCAPLUS
     100:176526
DN
     Entered STN: 26 May 1984
ED
ΤI
     Water-soluble film-forming products of polymer-analog reactions of vinyl
     acetate-maleic anhydride copolymer
     Smirnov, G. A.; Agafonov, G. I.; Kostyuchenko, G. F.; Smekalova, M. V.;
ΑU
     Mil'to, S. P.
     Yarosl. Politekh. Inst., Yaroslavl, USSR
CS
     Deposited Doc. (1982), SPSTL 599 Khp-D82, 7 pp. Avail.: SPSTL
SO
DT
     Report
LA
     Russian
     42-10 (Coatings, Inks, and Related Products)
CC
AΒ
     The reaction products of maleic anhydride-vinyl acetate copolymer
     (I) with allyl alc., ethylene glycol monomethacrylate (II), or tall oil
     fatty acid ethylene glycol monoester (FEM) were prepared
     in the presence of Et3N [121-44-8] and used for H2O-soluble, oil- and
     qasoline-resistant coatings. FEM-based coatings had higher resistance to
     H2O and lower hardness than those based on reaction products of I with
     allyl alc. or II. Coatings with optimum properties were obtained from
     reaction products of I with FEM at 1:0.8 maleic anhydride-FEM mol ratio
     crosslinked with 20% melamine-formaldehyde copolymer
     9003-08-1].
     vinyl acetate copolymer coating; maleic anhydride copolymer coating; alc
ST
     modified vinyl acetate copolymer
ΙT
    Crosslinking agents
        (melamine resins, for maleic anhydride-vinyl acetate copolymer reaction
        products with hydroxy-containing compds., for coatings)
ΙT
     Esterification catalysts
        (triethylamine, for maleic anhydride copolymers with hydroxy-containing
        compds.)
TΤ
    Coating materials
        (vinyl acetate polymer-based, alc.-modified, preparation and properties of)
ΙT
     Fatty acids, esters
     RL: SPN (Synthetic preparation); PREP (Preparation)
        (tall-oil, esters with ethylene glycol, reaction products with maleic
        anhydride-vinyl acetate copolymer, preparation and coating properties of)
ΙT
     121-44-8, uses and miscellaneous
     RL: CAT (Catalyst use); USES (Uses)
        (catalysts, for maleic anhydride copolymer esterification with
       hydroxy-containing compds.)
     9003-08-1
IΤ
     RL: USES (Uses)
        (crosslinking with, of maleic anhydride-vinyl acetate copolymer
        reaction product with hydroxy-containing compds., for coatings)
     107-18-6D, reaction products with maleic anhydride-vinyl acetate copolymer
TΤ
     107-21-1D, esters with tall oil fatty acids, reaction products with maleic
     acid-vinyl acetate copolymer 868-77-9D, reaction products with maleic
                                       9011-07-8D, reaction product with
     anhydride-vinyl acetate copolymer
     hydroxy-containing compds.
    RL: USES (Uses)
        (water-soluble preparation and coating properties of)
L76 ANSWER 27 OF 27 HCAPLUS COPYRIGHT 2004 ACS on STN
     1980:640560 HCAPLUS
AN
     93:240560
DN
```

```
ED
    Entered STN: 12 May 1984
    Polymeric polyamino-1, 3, 5-triazines containing at least one
ΤI
    polyalkylpiperidine group and their use in stabilizing an organic material
IN
    Nikles, Erwin
PA
    Ciba-Geigy A.-G., Switz.
SO
    Eur. Pat. Appl., 49 pp.
    CODEN: EPXXDW
DT
    Patent
T.A
    German
IC
    C08G073-06; C08G073-02
CC
    36-6 (Plastics Manufacture and Processing)
    Section cross-reference(s): 27
FAN.CNT 1
    PATENT NO.
                           DATE
                                     APPLICATION NO.
                     KIND
                                                         DATE
    _____
                     ____
    EP 13665
                            19800723
                                     EP 1980-810008
                                                          19800109
PΙ
                      A1
    EP 13665
                           19830629
                      В1
       R: BE, CH, DE, FR, GB, IT, NL
    US 4315859 A
                            19820216
                                     US 1980-109733
                                                          19800104
    EP 61785
                                     EP 1982-104017
                                                          19800109
                      Α1
                            19821006
    EP 61785
                            19850220
                      В1
    EP 61785
                           19880316
                      В2
       R: BE, CH, DE, FR, GB, IT, NL
    JP 55098180 A2 19800725
                                    JP 1980-3451
                                                          19800116
                                     US 1981-319352
    US 4528374
                            19850709
                                                          19811109
                      Α
PRAI CH 1979-357
                            19790115
    US 1980-109733
                            19800104
    EP 1980-810008
                           19800109
CLASS
PATENT NO.
            CLASS PATENT FAMILY CLASSIFICATION CODES
C08G073-06IC C08G073-02
EP 13665
             IC
GT
```

AB 1,7-Bis(2,2,6,6-tetramethyl-4-piperidyl)-1,4,7-triazaheptane (I) [68617-75-4], 1,10-bis(2,2,6,6-tetramethyl-4-piperidyl)-1,4,7,10-tetraazadecane [75577-94-5], 1-(2,2,6,6-tetramethyl-4-piperidyl)-1,4,7-triazaheptane [75578-13-1], 1-cyclohexyl-7-(2,2,6,6-tetramethyl-4-piperidyl)-1,4,7-triazaheptane [75578-18-6], and 12 similar compds. are prepared The compds. are copolymd. with cyanuric chloride (II), 2-amino-4,6-dichloro-s-triazine (III), 2,4-dichloro-6-(dimethylamino)-s-triazine, or another dichlorotriazine derivative or copolymd. with II (i.e., to give a Cl-substituted copolymer) and treated with an amine, diamine, mercaptan, alc., KCN, NaN3, Ac2O, BzCl, or a similar compound to prepare polymers which are useful as stabilizers for organic materials such as polyolefins and as hardeners for epoxy resins. Thus,

```
-
```

```
triacetoneamine [826-36-8] and diethylenetriamine [111-40-0] were used
     to prepare I which was copolymd. with III. The copolymer
     [75578-39-1] (mol. weight 2140) was used (0.3%) as a stabilizer in
     polypropene fibers. The fibers lost 50% of their strength during 3500 h
     in UV light, compared with 420 h for unstabilized fibers.
ST
     amine polymer light stabilizer; triazine polymer crosslinking agent;
     piperidine triazine polymer; crosslinker epoxy amine polymer;
     polypropylene fiber light stabilization
IT
     Epoxy resins, uses and miscellaneous
     RL: USES (Uses)
        (hardeners for, polyamines containing tetramethylpiperidyl and triazinyl
        groups as)
ΙT
     Polypropene fibers, uses and miscellaneous
     RL: USES (Uses)
        (light stabilizers for, tetramethylpiperidylamino and triazinyl
        group-containing polymers as)
ΙT
     Light stabilizers
        (tetramethylpiperidylamino and triazinyl group-containing polymers)
ΙT
     Crosslinking agents
        (tetramethylpiperidylamino and triazinyl group-containing polymers, for
        epoxy resins)
TT
     Polymerization
        (condensation, of diamines containing tetramethylpiperidyl groups with
        dichlorotriazines)
ΙT
     109-73-9DP, reaction products with bis(tetramethylpiperidylaminoethyl)amin
     e-cyanuric chloride copolymers 765-30-0DP, reaction products with
     bis(tetramethylpiperidylaminoethyl)amine-cyanuric chloride copolymers
                   75577-96-7P
                                 75577<del>-</del>97-8P
                                                75577-98-9P
     75577-95-6P
                                                              75577-99-0P
     75578-00-6P
                   75578-01-7P
                                 75578-02-8P
                                               75578-03-9P
                                                              75578-04-0DP,
                                     75578-04-0P
     reaction products with amines
                                                    75578-05-1P
                                                                  75578-06-2P
     75578-07-3P
                                               75578-10-8P
                   75578-08-4P
                                 75578-09-5P
                                                              75578-11-9P
     75578-12-0P
                   75578-14-2P
                                 75578-15-3P
                                                75578-16-4P
                                                              75578-17-5P
                   75578-21-1DP, reaction products with amines
     75578-19-7P
                                                                  75578-21-1P
     75578-23-3DP, reaction products with amines
                                                    75578-23-3P
                                                                  75578-25-5P
     75578-27-7DP, reaction products with amines
                                                   75578-27-7P
                                                                  75578-28-8P
     75578-31-3P
                   75578-33-5DP, reaction products with amines
                                                                  75578-33-5P
     75578-35-7DP, reaction products with benzyl bromide 75578-35-7P
                   75578-38-0P
                                75578-39-1P
                                               75587-12-1DP, derivs.
     75578-37-9P
     75587-12-1P
     RL: PREP (Preparation)
        (manufacture of, as stabilizers and epoxy hardeners)
ΙT
     68617-75-4P
                   75577-94-5P
                                 75578-13-1P
                                               75578-18-6P
                                                              75578-20-0P
     75578-22-2P
                                               75578-29-9P
                   75578-24-4P
                                 75578-26-6P
                                                              75578-30-2P
     75578-32-4P
                   75578-34-6P
                                 75578-36-8P
                                               75589-45-6P
                                                              75589-46-7P
     75589-47-8P
     RL: PREP (Preparation)
        (preparation of)
IT
     826-36-8
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with amines)
IΤ
     67-64-1, reactions
                         78-84-2
                                    100-52-7, reactions
                                                           108-94-1, reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with bis(tetramethylpiperidylaminoethyl)amine)
TΨ
     5554-54-1
                61682-93-7
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with diethylenetriamine)
                                                              13274-42-5
TΤ
     105-83-9
                111-40-0
                           112-24-3
                                      2157-24-6
                                                  4605-14-5
     25497-48-7
                  75589-44-5
     RL: RCT (Reactant); RACT (Reactant or reagent)
```

(reaction of, with triacetoneamine) IT 64-04-0D, reaction products with bis(tetramethylpiperidylaminoethyl)aminecyanuric chloride copolymers 71-36-3D, reaction products with bis(tetramethylpiperidylaminoethyl)amine-cyanuric chloride copolymers 78-81-9D, reaction products with bis(tetramethylpiperidylaminoethyl)aminecyanuric chloride copolymers 98-88-4D, reaction products with bis(tetramethylpiperidylaminoethyl)amine-cyanuric chloride copolymers 100-39-0D, reaction products with bis(tetramethylpiperidylaminopropoxy)but 100-46-9D, reaction products with ane-cyanuric chloride copolymers bis(tetramethylpiperidylaminoethyl)amine-cyanuric chloride copolymers 104-75-6D, reaction products with bis(tetramethylpiperidylaminoethyl)aminecyanuric chloride copolymers 107-11-9D, reaction products with bis(tetramethylpiperidylaminoethyl)amine-cyanuric chloride copolymers 107-15-3D, reaction products with bis(tetramethylpiperidylaminoethyl)aminecyanuric chloride copolymers 108-24-7D, reaction products with bis(tetramethylpiperidylaminoethyl)amine-cyanuric chloride copolymers 108-91-8D, reaction products with bis(tetramethylpiperidylaminoethyl)aminecyanuric chloride copolymers 108-93-0D, reaction products with bis(tetramethylpiperidylaminoethyl)amine-cyanuric chloride copolymers 109-89-7D, reaction products with bis(tetramethylpiperidylaminoethyl)aminecyanuric chloride copolymers 110-89-4D, reaction products with bis(tetramethylpiperidylaminoethyl)amine-cyanuric chloride copolymers 110-91-8D, reaction products with bis(tetramethylpiperidylaminoethyl)aminecyanuric chloride copolymers 111-26-2D, reaction products with bis(tetramethylpiperidylaminoethyl)amine-cyanuric chloride copolymers 111-86-4D, reaction products with bis(tetramethylpiperidylaminoethyl)amine-111-92-2D, reaction products with cyanuric chloride copolymers bis(tetramethylpiperidylaminoethyl)amine-cyanuric chloride copolymers 112-55-0D, reaction products with bis(tetramethylpiperidylaminoeth 112-90-3D, reaction yl)amine-cyanuric chloride copolymers products with bis(tetramethylpiperidylaminoethyl)amine-cyanuric chloride 123-75-1D, reaction products with bis(tetramethylpiperidylaminoethyl)amine-cyanuric chloride copolymers 124-09-4D, reaction products with bis(tetramethylpiperidylaminoethyl)aminecyanuric chloride copolymers 124-30-1D, reaction products with bis(tetramethylpiperidylaminoethyl)amine-cyanuric chloride copolymers 141-43-5D, reaction products with bis(tetramethylpiperidylaminoethyl)aminecyanuric chloride copolymers 151-50-8D, reaction products with bis(tetramethylpiperidylaminoethyl)amine-cyanuric chloride copolymers 302-01-2D, reaction products with bis(tetramethylpiperidylaminoethyl)aminecyanuric chloride copolymers 822-98-0D, reaction products with bis(tetramethylpiperidylaminoethyl)amine-cyanuric chloride copolymers 5332-73-0D, reaction products with bis(tetramethylpiperidylaminoethyl)amin e-cyanuric chloride copolymers 13952-84-6D, reaction products with bis(tetramethylpiperidylaminoethyl)amine-cyanuric chloride copolymers 26628-22-8D, reaction products with bis(tetramethylpiperidylaminoethyl)ami ne-cyanuric chloride copolymers 36768-62-4D, reaction products with bis(tetramethylpiperidylaminoethyl)amine-cyanuric chloride copolymers RL: USES (Uses) (stabilizers and epoxy hardeners) ΤT 112-55-0D, reaction products with bis(tetramethylpiperidylaminoeth yl)amine-cyanuric chloride copolymers RL: USES (Uses) (stabilizers and epoxy hardeners) RN 112-55-0 HCAPLUS 1-Dodecanethiol (7CI, 8CI, 9CI) (CA INDEX NAME) CN

 $HS-(CH_2)_{11}-Me$ 

```
=> => D OUE
L50
         134516 SEA FILE=REGISTRY ABB=ON 46.492.16/RID
          16503 SEA FILE=REGISTRY ABB=ON L50 AND PMS/CI
L51
L52
            174 SEA FILE=REGISTRY ABB=ON CARDANOL
              3 SEA FILE=REGISTRY ABB=ON LAURIC ACID/CN OR MYRISTIC ACID/CN
L53
                OR PALMITIC ACID/CN
L54
              7 SEA FILE=REGISTRY ABB=ON STEARIC ACID/CN OR ARACHIDIC ACID/CN
                OR PALMITOLEIC ACID/CN OR OLEIC ACID/CN OR RICINOLEIC ACID/CN
                OR LINOLEIC ACID/CN OR ARACHIDONIC ACID/CN
L55
              4 SEA FILE=REGISTRY ABB=ON DODECYL MERCAPTAN/CN OR PHENYL
                MERCAPTAN/CN OR LAURYL THIOGLYCOLATE/CN OR OCTYL THIOGLYCOLATE/
L56
          27181 SEA FILE=HCAPLUS ABB=ON
                                        L51
L57
         127435 SEA FILE=HCAPLUS ABB=ON
                                         L52 OR L53 OR L54 OR L55
L60
           7482 SEA FILE=HCAPLUS ABB=ON
                                         L56 (L) COPOLYMER?
L62
             10 SEA FILE=HCAPLUS ABB=ON
                                         L56(L)L57
                                         L56(L) FATTY ACID#(L) COPOLYMER?
L65
             41 SEA FILE=HCAPLUS ABB=ON
L66
            836 SEA FILE=HCAPLUS ABB=ON
                                         L57 (L) COPOLYMER?
L67
             10 SEA FILE=HCAPLUS ABB=ON
                                         L60 AND L66
L68
             56 SEA FILE=HCAPLUS ABB=ON
                                         L62 OR L65 OR L67
                                         L68 AND (PREP OR IMF OR SPN OR
L69
             16 SEA FILE=HCAPLUS ABB=ON
                POF)/RL
           7564 SEA FILE=REGISTRY ABB=ON 554-63-8/CRN OR 143-07-7/CRN OR
L70
                57-10-3/CRN OR 506-32-1/CRN OR 506-30-9/CRN OR 373-49-9/CRN OR
                141-22-0/CRN OR 112-80-1/CRN
           5141 SEA FILE=REGISTRY ABB=ON 60-33-3/CRN OR 57-11-4/CRN
L71
L72
             35 SEA FILE=REGISTRY ABB=ON L51 AND (L70 OR L71)
              1 SEA FILE=REGISTRY ABB=ON L51 AND L52
L73
             22 SEA FILE=HCAPLUS ABB=ON L72 OR L73
L74
             12 SEA FILE=HCAPLUS ABB=ON L74(L) (POF OR PREP OR IMF OR SPN)/RL
L75
             27 SEA FILE=HCAPLUS ABB=ON L69 OR L75
L76
             49 SEA FILE=HCAPLUS ABB=ON
                                         (L68 OR L74) NOT L76
L77
             39 SEA FILE=HCAPLUS ABB=ON
                                        L77 AND ?MELAMINE?
L78
             32 SEA FILE=HCAPLUS ABB=ON L78 AND COPOLYMER?
L80
```

## => D L80 BIB ABS HITIND HITSTR 1-32

L80 ANSWER 1 OF 32 HCAPLUS COPYRIGHT 2004 ACS on STN

AN 1993:561751 HCAPLUS

DN 119:161751

TI Melamine resin-containing materials for injection molding

IN Sawada, Nobuyuki; Myagawa, Takayuki; Yamamoto, Kazuo; Myata, Kazuyoshi

PA Nippon Carbide Kogyo Kk, Japan

SO Jpn. Kokai Tokkyo Koho, 10 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 05163419	A2	19930629	JP 1991-350381	19911211
	JP 3144867	B2	20010312		

PRAI JP 1991-350381 19911211 The title materials, showing good cylinder stability and giving moldings with a good appearance, contain pulp 20-60, higher fatty acids 0.1-5, and HCHO-melamine resins 100 parts. A resin prepared from 1260 g melamine and 1380 g 37% aqueous HCHO was mixed (3000 g) with 500 g pulp, kneaded and dried at 90° to form a popcorn-like product, mixed (600 g) with 900 g powdered melamine resin, 300 g powdered pulp, 13.5 g stearic acid, TiO2, red pigment, and additives, pulverized, melt kneaded, extruded, pulverized, and used to prepared injection moldings showing good crack resistance and gloss 95%. IC ICM C08L061-28 ICS B29C039-00; B29C045-00; C08K005-09; C08L001-00 37-6 (Plastics Manufacture and Processing) CC Section cross-reference(s): 38, 43 ST melamine resin pulp injection molding; fatty acid melamine resin molding; stearic acid melamine resin molding; gloss molding melamine resin pulp; crack resistance melamine resin pulp IT Pulp, cellulose (melamine resins containing fatty acids and, injection-moldable) IT Molding of plastics and rubbers (injection, melamine resin-fatty acid-pulp mixts. for) 9003-08-1, Melamine-formaldehyde copolymer IT RL: USES (Uses) (injection-moldable, containing pulp and fatty acids) ΙT 57-11-4, Stearic acid, miscellaneous 112-85-6, Behenic acid RL: MSC (Miscellaneous) (melamine resins containing pulp and, injection-moldable) IT 9004-34-6 RL: USES (Uses) (pulp, melamine resins containing fatty acids and, injection-moldable) ΙT 9003-08-1, Melamine-formaldehyde copolymer RL: USES (Uses) (injection-moldable, containing pulp and fatty acids) RN 9003-08-1 HCAPLUS 1,3,5-Triazine-2,4,6-triamine, polymer with formaldehyde (9CI) (CA INDEX CN NAME) CM 1 CRN 108-78-1 CMF C3 H6 N6

CM 2

CRN 50-00-0 CMF C H2 O

## $H_2C = 0$

- L80 ANSWER 2 OF 32 HCAPLUS COPYRIGHT 2004 ACS on STN
- AN 1991:561159 HCAPLUS
- DN 115:161159
- TI Aqueous dispersion of copolymers bearing perfluoroalkyl groups
- IN Roettger, Jutta; Passon, Karl Heinz; Maurer, Werner; Meyer, Rolf Volker; Kortmann, Wilfried; Selinger, Peter
- PA Bayer A.-G., Germany
- SO Eur. Pat. Appl., 12 pp.

CODEN: EPXXDW

- DT Patent
- LA German
- FAN. CNT 1

PAN.	CNI I				
	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
ΡI	EP 424765	A2	19910502	EP 1990-119662	19901013
	EP 424765	A3	19920729		
	EP 424765	B1	19940803		
	R: BE, DE, FR,	GB			
	DE 3935859	A1	19910502	DE 1989-3935859	19891027
	US 5093398	A	19920303	US 1990-592142	19901003
	JP 03174418	A2	19910729	JP 1990-282059	19901022
	CA 2028503	AA	19910428	CA 1990-2028503	19901025
	CA 2028503	С	20011218		
PRAI	DE 1989-3935859	Α	19891027		

- AB The title dispersions, useful as water- and oilproofing finishes for textiles, paper, and leather, contain (graft) copolymers of unsatd. compds. bearing C≥6 perfluoroalkyl groups and comonomers free of ester groups, and esters bearing linear C≥6 groups and 1,2-substituted vinyl groups or no vinyl groups. Peroxide-initiated emulsion polymerization of 2-(N-methylperfluorooctanesulfonamido)ethyl methacrylate 43.4, octadecyl methacrylate 12.7, and vinyl acetate 18.45 parts in the presence of .apprx.29 g glycerol monooleate (I) gave a 14.4% emulsion (average particle size 367 nm) of a polymer containing 20.1% F. A 40:60
  - mixture of this dispersion and a colloidal siloxane dispersion was diluted to 2.5% solids, sprayed (1% solids) on a polyamide carpet (basis atomic 500 g/m2, 30% residual moisture), dried for 5-15 min at 125-150°, and conditioned at 23° and 65% relative humidity for 24 h to give a carpet with oilproofing rating (AATCC method 118, 5 best) 3-4 and waterproofing (resistance to aqueous iso-PrOH, 40% better than 30%) 40-50%; vs. 2 and 40, resp., without I.
- IC ICM B01F017-36
  - ICS C08K005-10; C08L033-14
- CC 40-9 (Textiles and Fibers)
  - Section cross-reference(s): 35, 43, 45
- ST fluoropolymer oilproofing waterproofing finish; oilproofing finish fluoropolymer emulsion; waterproofing finish fluoropolymer emulsion; nylon carpet oilproofing waterproofing; glycerol oleate fluoropolymer emulsion; methacrylate perfluoroalkyl copolymer emulsion; vinyl acetate copolymer emulsion
- IT 105-59-9D, 2,2'-(Methylimino) diethanol, reaction products with melamine resins and behenic acid, graft polymers with (methylperfluorooctanesulfonamido) ethyl methacrylate, octadecyl methacrylate and vinyl acetate 108-05-4D, Acetic acid ethenyl ester,

graft polymers with melamine resin-behenic acid-(methylimino)diethanol reaction products, octadecyl methacrylate and (methylperfluorooctanesulfonamido)ethyl methacrylate 112-85-6D, Behenic acid, reaction products with melamine resins and (methylimino)diethanol, poly graft polymers with (methylperfluorooctanesulfonamido)ethyl methacrylate, octadecyl methacrylate and vinyl acetate 9003-08-1D, Formaldehydemelamine copolymer, reaction products with behenic acid and (methylimino)diethanol, graft polymers with (methylperfluorooctanesulfonamido) ethyl methacrylate, octadecyl 14650-24-9D, graft polymers with methacrylate and vinyl acetate melamine resin-behenic acid-(methylimino)diethanol condensation products, octadecyl methacrylate and vinyl acetate 32360-05-7D, graft polymers with melamine resin-behenic acid-(methylimino)diethanol reaction products, (methylperfluorooctanesulfonamido)ethyl methacrylate and vinyl acetate 136403-96-8 RL: USES (Uses) (emulsions, containing fatty acid esters, as oilproofing and waterproofing finishes) 9003-08-1D, Formaldehyde-melamine copolymer, reaction products with behenic acid and (methylimino)diethanol, graft polymers with (methylperfluorooctanesulfonamido)ethyl methacrylate, octadecyl methacrylate and vinyl acetate RL: USES (Uses) (emulsions, containing fatty acid esters, as oilproofing and waterproofing finishes)

9003-08-1 HCAPLUS

1,3,5-Triazine-2,4,6-triamine, polymer with formaldehyde (9CI) (CA INDEX NAME)

CM 1

TΤ

RN

CN

CRN 108-78-1 CMF C3 H6 N6

2 CM

CRN 50-00-0 CMF C H2 O

 $H_2C = 0$ 

L80 ANSWER 3 OF 32 HCAPLUS COPYRIGHT 2004 ACS on STN AN 1990:181468 HCAPLUS DN 112:181468 ΤI Oxidative film formation of melamine-formaldehyde oligomers

```
modified with unsaturated fatty acids
     Semina, R. A.; Mikhailova, T. B.; Makotkin, A. V.; Livshits, R. M.
ΑU
     GIPI, USSR
CS
     Lakokrasochnye Materialy i Ikh Primenenie (1989), (4), 17-20
SO
     CODEN: LAMAAD; ISSN: 0023-737X
DT
     Journal
LA
     Russian
     Modification of oligomeric melamine-HCHO resin (I) with tall-oil
AR
     fatty acids gave film-forming materials with low content of volatile
     compds. capable of oxidative crosslinking. The obtained coatings
     exhibited properties similar to those of alkyd coatings and superior to
     those of nonmodified I-based coatings. Investigations conducted on model
     oligomers obtained by acidolysis of hexamethoxymethylmelamine
     with linoleic and linolenic acids showed that a significant role in
     crosslinking I oligomers plays homocondensation of MeO groups. This
     reaction can be catalyzed by products of oxidation of unsatd. fatty acid
     residues.
     42-10 (Coatings, Inks, and Related Products)
CC
     aminoplast fatty acid modification coating; melamine resin fatty
ST
     acid modification; tall oil acid modification aminoplast; linoleic acid
     modification; linolenic acid modification aminoplast
     Coating materials
IT
        (melamine resin modified with tall-oil fatty acids, oxidative
        crosslinking and properties of)
ΙT
     Glass temperature and transition
        (of coatings prepared by oxidative crosslinking of tall-oil fatty
        acid-modified oligomeric melamine resins)
IT
     Crosslinking
        (oxidative, of melamine resin-based coatings modified with
        tall-oil fatty acids)
ΙT
     Fatty acids, compounds
     RL: USES (Uses)
        (tall-oil, reaction products, with oligomeric melamine
        resins, coatings, oxidative crosslinking of)
TΨ
     126540-35-0
                 126540-36-1
     RL: USES (Uses)
        (oligomeric, as model for fatty acid-modified melamine
        -formaldehyde resin coating, oxidative crosslinking of)
TΨ
     9003-08-1D, Formaldehyde-melamine copolymer,
     reaction products with tall-oil fatty acids
     RL: USES (Uses)
        (oligomeric, coatings, oxidative crosslinking and properties of)
TΤ
     126540-35-0
     RL: USES (Uses)
        (oligomeric, as model for fatty acid-modified melamine
        -formaldehyde resin coating, oxidative crosslinking of)
RN
     126540-35-0 HCAPLUS
     9,12-Octadecadienoic acid (9Z,12Z)-, polymer with N,N,N',N',N'',N''-
CN
     hexakis (methoxymethyl)-1,3,5-triazine-2,4,6-triamine (9CI) (CA INDEX
     NAME)
     CM
          1
     CRN 3089-11-0
     CMF C15 H30 N6 O6
```

$$\begin{array}{c|c} \text{MeO-CH}_2 \\ \text{N-CH}_2\text{-OMe} \\ \\ \text{MeO-CH}_2\text{-N} \\ \text{N-CH}_2\text{-OMe} \\ \\ \text{MeO-CH}_2 \\ \text{CH}_2\text{-OMe} \\ \end{array}$$

CM 2

CRN 60-33-3 CMF C18 H32 O2

Double bond geometry as shown.

IT 9003-08-1D, Formaldehyde-melamine copolymer,

reaction products with tall-oil fatty acids

RL: USES (Uses)

(oligomeric, coatings, oxidative crosslinking and properties of)

RN 9003-08-1 HCAPLUS

CN 1,3,5-Triazine-2,4,6-triamine, polymer with formaldehyde (9CI) (CA INDEX NAME)

CM 1

CRN 108-78-1 CMF C3 H6 N6

CM 2

CRN 50-00-0 CMF C H2 O

 $H_2C = 0$ 

```
L80 ANSWER 4 OF 32 HCAPLUS COPYRIGHT 2004 ACS on STN
ΑN
     1990:58191 HCAPLUS
DN
     112:58191
     Composition for antipilling finishing of textile materials
ΤI
IN
     Podshibyakina, K. D.; Lakhman, L. F.
     Latvian Scientific-Research Institute of Light Industry, USSR
PA
SO
     U.S.S.R.
     From: Otkrytiya, Izobret. 1989, (31), 151.
     CODEN: URXXAF
DT
     Patent
LA
    Russian
FAN.CNT 1
     PATENT NO.
                       KIND
                               DATE
                                          APPLICATION NO.
                                                                  DATE
                               -----
                        ____
                                           -----
                                                                  _____
                         A1
    SU 1502672
                                19890823
                                          SU 1987-4287768
                                                                 19870721
PΙ
PRAI SU 1987-4287768
                               19870721
    In the title finishing of wool and wool blend fabrics with
     methylolmelamine derivs., pilling resistance of the fabrics is
     improved if the fabrics are finished with compns. comprising Me or
     ethylene glycol esters of penta- and hexamethylolmelamine,
     poly(oxypropylene, glycol)-MDI block copolymer (Latur 1052-3F),
     and aqueous 30-40% dispersions Me sulfate salts of 2-[N-
     (methyldiethanol)amino]ethyl esters of tall-oil carboxylic acids. An aqueous
     antipilling composition contains 1.0-1.3% Me or ethylene glycol esters of
     penta- and hexamethylolmelamine, 2.0-2.5% polyurethane latex
     (Latur 1052-3F), and 1-2% aqueous dispersion of Me sulfate of
     2-[N-(methyldiethanol)amino]ethyl ester of tall-oil carboxylic acids.
     ICM D06M015-61
IC
     ICS D06M015-564
     40-9 (Textiles and Fibers)
CC
     melamine resin antipilling agent wool; wool textile antipilling
ST
     finishing; polyurethane antipilling finish wool textile; fatty acid ester
     antipilling finish
IT
     Urethane polymers, uses and miscellaneous
     RL: USES (Uses)
        (antipilling finishes, containing melamine resins and tall-oil
        fatty acid Me sulfate salts, for wool and wool blends)
IT
     Fatty acids, esters
     RL: USES (Uses)
        (tall-oil, esters, with 2-[N-(methyldiethanol)amino]ethanol, Me sulfate
        salts, antipilling finishes, containing melamine resins and
       polyurethanes, for wool and wool blend)
ΙT
     Textiles
        (wool, antipilling finishing of, with melamine resins containing
        polyurethanes and Me sulfate salts of (methyldiethanol)ethyl ester of
        tall-oil fatty acids)
ΙT
     9003-08-1, Formaldehyde-melamine copolymer
     RL: USES (Uses)
        (antipilling agents, with polyurethanes and fatty
       acid ester salts, for wool and wool blend fabrics)
IT
     9048-57-1
     RL: USES (Uses)
        (antipilling finishes (Latur 1052-3F), containing melamine resins
        and fatty acid ester salts, for wool and wool blend fabrics)
ΙT
     9003-08-1, Formaldehyde-melamine copolymer
    RL: USES (Uses)
        (antipilling agents, with polyurethanes and fatty
       acid ester salts, for wool and wool blend fabrics)
    9003-08-1 HCAPLUS
RN
```

CN 1,3,5-Triazine-2,4,6-triamine, polymer with formaldehyde (9CI) (CA INDEX NAME)

CM 1

CRN 108-78-1 CMF C3 H6 N6

CM 2

CRN 50-00-0 CMF C H2 O

H2C=0

L80 ANSWER 5 OF 32 HCAPLUS COPYRIGHT 2004 ACS on STN

AN 1989:576584 HCAPLUS

DN 111:176584

TI Melamine resin sizing agents for paper

IN Goetze, Thomas; Kamutzki, Walter; Schoen, Manfred

PA Cassella A.-G., Fed. Rep. Ger.

SO Ger. Offen., 7 pp.

CODEN: GWXXBX

DT Patent

LA German

FAN.CNT 1

		_									
	PA:	rent :	NO.			KINI	)	DATE		APPLICATION NO. DA	ATE
							-				
PI	DE	3741	282			A1		1989	0615	DE 1987-3741282	9871205
	WO	8905	376			A1		1989	0615	WO 1988-EP1080 19	9881129
		W:	BR,	FI,	JP,	US					
		RW:	AT,	BE,	CH,	DE,	FR,	, GB,	ΙT,	NL, SE	
	BR	8807	832	•	·	A		1990	1023	BR 1988-7832	9881129
	EP	3931	05			A1		1990	1024	EP 1989-900215 19	9881129
		R:	AT,	BE,	CH,	DE,	FR,	GB,	ΙT,	LI, NL, SĒ	
	JP	0350	3787			Т2		1991	0822	JP 1989-500249 19	9881129
	ES	2012	576			Α6		1990	0401	ES 1988-3710 19	9881205
PRAI	DE	1987	-374	1282				1987	1205		
	WO	1988	-EP1	080				1988	1129		

OS MARPAT 111:176584

AB Sizing agents for paper which are storage-stable and readily emulsifiable contain melamines bearing 1-5 CH2OH or alkoxymethyl groups, 1-4 ROCH2, RCO2CH2, or RCONHCH2 groups (R = C8-28 hydrocarbyl), and 0-2 (melaminomethoxy)methyl or melaminomethyl groups. Heating 195 g 95% 1:5.8 melamine-HCHO resin, 216.4 g C18H37OH, and 10 mL 55% HNO3 at 60° in vacuo for 3 h gave a modified resin. A mixture of this resin

```
18, 87% cationic starch 10.3, dispersant 1, and H2O 262 g was used to size
     paper at concentration 0.5, 1, and 2% (based on solids), resulting in sizing
     rating (≥20 very good, 10-20 good, 5-10 fair, 1-5 poor) 0.2, >40,
     and >40, resp.
IC
     ICM D21H001-28
     ICS D21H003-48
     C08J003-06, C08L061-28
ICI
     43-7 (Cellulose, Lignin, Paper, and Other Wood Products)
CC
     Section cross-reference(s): 38
ST
     sizing agent paper; melamine resin modified size; octadecyl
     ether melamine resin size
ΙT
     Sizes
        (melamine resin fatty alkyl ethers and fatty acid esters,
        storage-stable and emulsifiable, for paper)
ΙT
     Paper
        (sizes for, storage-stable and emulsifiable melamine resin
        fatty acid esters and fatty alkyl ethers as)
IT
     Fatty acids, esters
     RL: USES (Uses)
        (esters, with melamine resins, sizes for paper)
     57-10-3D, Palmitic acid, esters with melamine resins
IΤ
                                                             57-11-4D.
     Stearic acid, esters with melamine resins 108-78-1D,
     1,3,5-Triazine-2,4,6-triamine, methylolated, fatty acid esters and fatty
     alkyl ethers 112-53-8D, 1-Dodecanol, ethers with melamine
              112-80-1, Oleic acid, uses and miscellaneous 112-92-5D,
     1-Octadecanol, ethers with melamine resins 124-26-5D,
     Stearamide, reaction products with epoxy resins 9003-08-1D,
     Formaldehyde-melamine copolymer, fatty
     acid esters and fatty alkyl ethers
     RL: USES (Uses)
        (sizes, storage-stable and emulsifiable, for paper)
     9003-08-1D, Formaldehyde-melamine copolymer,
TΤ
     fatty acid esters and fatty alkyl ethers
     RL: USES (Uses)
        (sizes, storage-stable and emulsifiable, for paper)
     9003-08-1 HCAPLUS
RN
     1,3,5-Triazine-2,4,6-triamine, polymer with formaldehyde (9CI) (CA INDEX
CN
     NAME)
     CM
          1
     CRN 108-78-1
     CMF C3 H6 N6
       NH<sub>2</sub>
            NH2
          2
     CM
     CRN
         50-00-0
     CMF
         C H2 O
```

H2C== 0

```
ANSWER 6 OF 32 HCAPLUS COPYRIGHT 2004 ACS on STN
T.80
     1988:592279 HCAPLUS
ΑN
DN
     109:192279
     Stoving primers for metals, containing dicyclopentadienyl maleates
TΙ
     Zimmermann, Frank; Freitag, Werner; Lohs, Werner; Sarfert, Werner;
IN
     Thielemann, Volkmar; Kurtessis, Nikos; Becker, Sieghard
     VEB Lackfabrik Dresden, Ger. Dem. Rep.
PΑ
     Ger. (East), 3 pp.
SO
     CODEN: GEXXA8
DT
     Patent
LA
     German
FAN.CNT 1
                       KIND
                                          APPLICATION NO.
                                                                 DATE
     PATENT NO.
                                DATE
                        ____
                                            _____
                                                                  19861003
                                19880113 DD 1986-294996
     DD 253252
                         A1
PRAI DD 1986-294996
                                19861003
     The title compns., with improved use properties, contain esters of 1:1
     dicyclopentadienyl maleate (I) with aliphatic dicarboxylic acid-diol diesters
     (I functionality 1.5-2) 50-60, alkylated melamine resins of
     medium reactivity 20-30, and esters of nondrying C8-18 fatty acids
     (.apprx.40%) with bisphenol A epoxy resins (epoxy equivalent 800-1025) 15-25%.
     A mixture of 58:25:17 I-1,3-butanediol-adipic acid ester (viscosity of an
     80% xylene solution 2.1 Pa-s) 55, butylated melamine resin 25, and
     ester of 40% coco fatty acids with bisphenol A epoxy resin 20% gave a
     coating with crosscut adhesion 1, gloss 110%, and Erichsen indentation 8
     mm before, and 1-2, 105, and 6.5, resp., after, 24 h at 200^{\circ}; vs.
     1, 90, 4, 4, 40, and 0.5, resp., for an alkyd-phenolic resin-rosin
     coating.
     ICM C09D003-49
IC
CC
     42-8 (Coatings, Inks, and Related Products)
     dicyclopentadienyl maleate coating; adipate butanediol maleate coating;
ST
     melamine resin butylated coating; epoxy resin ester coating; coco
     fatty acid ester coating
IT
     Coating materials
        (dicyclopentadienyl maleates-alkylated melamine resins-epoxy
        resin fatty acid esters, for metals)
TΨ
     107-88-0D, 1,3-Butanediol, esters with dicyclopentadienylmaleate and
     adipic acid 124-04-9D, Adipic acid, esters with butanediol and
     dicyclopentadienylmaleate 28347-17-3D, esters with butanediol and adipic
     acid
     RL: TEM (Technical or engineered material use); USES (Uses)
        (coatings, containing alkylated melamine resins and epoxy resin
        fatty acid esters, for metals)
IT
     9003-08-1, Formaldehyde-melamine copolymer
     RL: TEM (Technical or engineered material use); USES (Uses)
        (coatings, containing dicyclopentadienyl maleates and epoxy resin
        fatty acid esters, for metals)
ΙT
     25068-38-6D, esters with coco fatty acids
     RL: TEM (Technical or engineered material use); USES (Uses)
        (coatings, containing dicyclopentadienylmaleates and alkylated
        melamine resins, for metals)
IT
     9003-08-1, Formaldehyde-melamine copolymer
```

RL: TEM (Technical or engineered material use); USES (Uses)

(coatings, containing dicyclopentadienyl maleates and epoxy resin

fatty acid esters, for metals)

RN 9003-08-1 HCAPLUS

CN 1,3,5-Triazine-2,4,6-triamine, polymer with formaldehyde (9CI) (CA INDEX NAME)

CM 1

CRN 108-78-1 CMF C3 H6 N6

CM 2

CRN 50-00-0 CMF C H2 O

 $H_2C = 0$ 

L80 ANSWER 7 OF 32 HCAPLUS COPYRIGHT 2004 ACS on STN

AN 1987:479829 HCAPLUS

DN 107:79829

TI Manufacture of microcapsules

IN Fukuo, Hidetoshi; Onoguchi, Tomio

PA Sakura Color Products Corp., Japan

SO Jpn. Kokai Tokkyo Koho, 4 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 2

T LITA.	CNIZ					
	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE	
ΡI	JP 62001451	A2	19870107	JP 1985-137825	19850626	
	JP 06018637	B4	19940316			
	US 4753759	Α	19880628	US 1986-892783	19860801	
PRAI	JP 1985-137825		19850626			
	JP 1985-236176		19851021			

AB The title microcapsules, useful for encapsulating leuco dyes in the manufacture of thermal inks or inks for carbonless paper, are prepared by emulsifying hydrophobic, slightly volatile organic compds. in aqueous solns. of acrylic acid-itaconic acid copolymers, adding urea or melamine and HCHO, adjusting the pH to 2.5-6.0, and forming coatings of urea-HCHO or melamine-HCHO copolymers around the organic compds.

Thus, an aqueous solution of crystal violet lactone in KMC 113 was added to an aqueous solution of Jurymer 50P, urea, and resorcinol at 50°, the mixture was stirred at 5000 rpm, aqueous NaOH was added to pH 4.0, 37% aqueous HCHO was added, and the mixture was heated at 55°. The resulting slurry of microcapsules was sprayed onto paper to prepare a carbonless paper printing

in blue. IC ICM B01J013-02 ICA B41M005-12; B41M005-18 43-7 (Cellulose, Lignin, Paper, and Other Wood Products) Section cross-reference(s): 42 microencapsulation leuco dye thermal ink; carbonless paper ink dye ST microencapsulation; urea formaldehyde copolymer microcapsule dye; melamine formaldehyde copolymer microcapsule dye; acrylic itaconic copolymer emulsifier microencapsulation TΤ Encapsulation (micro-, of leuco dyes with urea- or melamine-formaldehyde copolymer, for thermal inks and carbonless paper inks) ΙT 25948-33-8, Acrylic acid-itaconic acid copolymer RL: USES (Uses) (emulsifiers, for leuco dyes, in microencapsulation with urea- or melamine-formaldehyde copolymer) 29512-49-0 IT RL: PROC (Process) (microencapsulation of, with melamine-formaldehyde copolymer, for carbonless paper ink) ΙT 98724-23-3, O-DCF RL: PROC (Process) (microencapsulation of, with melamine-formaldehyde copolymer, for thermal ink) TΤ 1552-42-7, Crystal violet lactone RL: PROC (Process) (microencapsulation of, with urea-formaldehyde copolymer, for carbonless paper ink) ΙT 9003-08-1, Melamine-formaldehyde copolymer 9011-05-6, Urea-formaldehyde copolymer RL: USES (Uses) (microencapsulation with, of leuco dye solns., for carbonless paper) IT 143-07-7, Lauric acid, uses and miscellaneous 544-63-8, Myristic acid, uses and miscellaneous 38640-62-9, KMC 113 663917-53-1, Hisol SAS 296 RL: USES (Uses) (solvents, for leuco dye, in microencapsulation with urea- or melamine-formaldehyde copolymer) TΨ 9003-08-1, Melamine-formaldehyde copolymer RL: USES (Uses) (microencapsulation with, of leuco dye solns., for carbonless paper) RN 9003-08-1 HCAPLUS 1,3,5-Triazine-2,4,6-triamine, polymer with formaldehyde (9CI) (CA INDEX CN NAME.) CM 1 CRN 108-78-1 CMF C3 H6 N6

CM 2

CRN 50-00-0 CMF C H2 O

 $H_2C = 0$ 

IT 143-07-7, Lauric acid, uses and miscellaneous 544-63-8,

Myristic acid, uses and miscellaneous

RL: USES (Uses)

(solvents, for leuco dye, in microencapsulation with urea- or melamine-formaldehyde copolymer)

RN 143-07-7 HCAPLUS

CN Dodecanoic acid (9CI) (CA INDEX NAME)

 $HO_2C^-(CH_2)_{10}^-Me$ 

RN 544-63-8 HCAPLUS

CN Tetradecanoic acid (9CI) (CA INDEX NAME)

 $HO_2C^-$  (CH<sub>2</sub>)<sub>12</sub>-Me

L80 ANSWER 8 OF 32 HCAPLUS COPYRIGHT 2004 ACS on STN

AN 1987:158065 HCAPLUS

DN 106:158065

TI Thermosetting high solids primer composition comprising epoxy ester resin and hydroxy-reactive crosslinking agent

IN Kordomenos, Panagiotis I.; Maker, David L.; Mros, Gordon R.; Nordstrom, John D.

PA Ford Motor Co., USA

SO U.S., 11 pp.

CODEN: USXXAM
Patent

LA English

FAN.CNT 1

DT

DATE APPLICATION NO. PATENT NO. KIND DATE ----\_\_\_\_\_ -----US 4626578 19861202 PΤ Α US 1984-685637 19841224 PRAI US 1984-685637 19841224

AB Solvent-based thermosetting primers for automobiles comprise aliphatic diol-extended, fatty acid-terminated epoxy ester resin (number-average mol. weight

1000-5000) and aminoplast or blocked isocyanate crosslinking agents (0.5-1.6 OH-reactive groups/epoxy resin OH group). Thus, a formulation comprised 1257:901:836 Epon 829-polycaprolactone diol-soybean fatty acid copolymer (I) 270, I-containing millbase 800, melamine resin (Resimine 717) 110, Mn naphthenate 4, and BuOH 35 parts.

IC ICM C08G018-80

NCL 525484000

CC 42-9 (Coatings, Inks, and Related Products)

IT 4767-03-7D, Dimethylol propionic acid, polymers with epoxy resins and

fatty acids and aminoplasts or blocked polyisocyanates 5314-31-8D, reaction products with polyisocyanates, polymers with epoxy resins and diols and fatty acids 9003-08-1D, Formaldehyde-melamine copolymer, polymers with epoxy resins and diols and fatty acids 9016-87-9D, Papi 580, reaction products with Me amyl ketoxime, polymers with epoxy resins and diols and fatty acids 25068-38-6D, Epon 829, polymers with diols and fatty acids and aminoplasts or blocked polyisocyanates 29611-97-0D, Araldite RD-2, polymers with diols and fatty acids and aminoplasts or blocked isocyanates 50327-24-7D, PCP 0200, polymers with epoxy resins and fatty acids and aminoplasts or blocked isocyanates 54986-73-1D, Desmodur IL, polymers with epoxy resins and diols and fatty acids 81774-35-8D, DER 333, polymers with diols and fatty acids and aminoplasts or blocked isocyanates 84286-52-2D, Desmodur L-2291a, polymers with epoxy resins and diols and fatty acids

RL: TEM (Technical or engineered material use); USES (Uses) (coatings, for automobiles)

IT 9003-08-1D, Formaldehyde-melamine copolymer,

polymers with epoxy resins and diols and fatty acids

RL: TEM (Technical or engineered material use); USES (Uses)

(coatings, for automobiles)

RN 9003-08-1 HCAPLUS

CN 1,3,5-Triazine-2,4,6-triamine, polymer with formaldehyde (9CI) (CA INDEX NAME)

CM 1

CRN 108-78-1 CMF C3 H6 N6

CM 2

CRN 50-00-0 CMF C H2 O

 $H_2C = 0$ 

L80 ANSWER 9 OF 32 HCAPLUS COPYRIGHT 2004 ACS on STN

AN 1983:524167 HCAPLUS

DN 99:124167

TI Water-thinned thermosetting coating materials with good resistant to foaming during baking

PA Kansai Paint Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 6 pp.

CODEN: JKXXAF

DT Patent

```
Japanese
T.A
FAN.CNT 1
     PATENT NO.
                       KIND DATE
                                          APPLICATION NO.
                                                                  DATE
                       JP 58040370
                         A2 19830309 JP 1981-138900 19810903
PΙ
                         B4 19900108
     JP 02000390
PRAI JP 1981-138900
                               19810903
     Foaming of water-thinned thermosetting resin coating materials during
     baking is prevented by 0.1-10% (based on solids) benzoin (I) [119-53-9]
     or C1-4 alkyl ether of I. Thus, a composition of 66% solid alkyd resin
solution
     (prepared by polymerizing neopentyl glycol 31, trimethylolpropane 7, phthalic
     anhydride 32, tetrahydrophthalic anhydride 17, trimellitic anhydride 5,
     and tall-oil fatty acid 15 parts, diethanolamine salt)
     22.5, TiO2 80, baryta 20, carbon black 0.4, Nopco DF 122 NS 1.5, and H2O
     30 parts with 1.0% (based on total solids of final coating) I was milled
     20 h and mixed with the above alkyd resin solution 100, 88% solids
     melamine-formaldehyde copolymer [9003-08-1]
     solution 22.4, EtO(CH2)2O(CH2)2OH 10, surfactant 1, and H2O 50 parts to give
     a coating material. When the above composition was applied to a steel plate through a spray gun, set 10 min, and baked 20 min at 160°, the maximum
     thickness of dry coating without foaming was 52 \mu (30 \mu without I).
IC
     C09D005-00
CC
     42-7 (Coatings, Inks, and Related Products)
L80 ANSWER 10 OF 32 HCAPLUS COPYRIGHT 2004 ACS on STN
     1983:455128 HCAPLUS
ΑN
     99:55128
DN
     High-solids alkyd resin coating materials
ΤI
PA
     Daicel Chemical Industries, Ltd., Japan
SO
     Jpn. Kokai Tokkyo Koho, 5 pp.
     CODEN: JKXXAF
DΤ
     Patent
LA
     Japanese
FAN.CNT 1
                    KIND DATE APPLICATION NO.
                                                                DATE
     PATENT NO.
                               -----
                                           ______
                        ----
     JP 58021422 A2 19830208 JP 1981-120506
JP 1981-120506 19810731
                                                                  19810731
PΤ
PRAI JP 1981-120506
     High-solids alkyd resin coating materials having OH number 20-300, and acid
     number 0.1-20 are prepared from polyols, polycarboxylic acids or anhydrides,
     0-50% vegetable or animal oils or fatty acids, and
     5-80\% \epsilon-caprolactone (I). Thus, a mixture of phthalic anhydride
     (II) 592, soya fatty acid 520, glycerol 386, ethylene
     glycol 65, I 273, (BuO)4Ti 0.0076, and xylene 50 parts was heated 10 h at
     160-210° to reach acid number 0.23, cooled to 150°, stirred
     with 50 parts II at 150° for 2 h, and thinned with xylene to 79.3%
     solids to give copolymer (III) solution having acid number 6.3, OH number
     120.6, and viscosity 900 cP at 25°. A 70:30 (solids) mixture of the
     III solution and Yuban 20SE [9003-08-1] was applied to a steel plate and baked 15 min at 120° to form a 50-60 \mu coating.
IC
     C08G063-46; C08G063-48
ICA C09D003-64
     42-8 (Coatings, Inks, and Related Products)
CC
     Section cross-reference(s): 55
ST
     alkyd resin high solid; caprolactone copolymer coating; soya
     fatty acid copolymer; melamine resin crosslinking
ΙT
     Crosslinking agents
```

(melamine-formaldehyde copolymer, for high-solids alkyd resin coatings) IT Coating materials (high-solids, alkyd resins, containing melamine resins) ΙT 56-81-5D, polymers with phthalic anhydride, glycol, caprolactone, and soya fatty acid 85-44-9D, polymers with glycerol, glycol, caprolactone, and soya fatty acid 107-21-1D, polymers with phthalic anhydride, glycerol, caprolactone, and soya fatty acid 126-30-7D, polymers with phthalic anhydride, glycerol, caprolactone, and soya fatty acid 502-44-3D, polymers with phthalic anhydride, glycerol, and soya fatty acid RL: TEM (Technical or engineered material use); USES (Uses) (coatings, high-solids, containing melamine resins) L80 ANSWER 11 OF 32 HCAPLUS COPYRIGHT 2004 ACS on STN 1983:200003 HCAPLUS AN 98:200003 DN ΤI Soldering resists Toshiba Chemical Products Co., Ltd., Japan PΑ SO Jpn. Kokai Tokkyo Koho, 6 pp. CODEN: JKXXAF DT Patent LA Japanese FAN.CNT 1 PATENT NO. KIND DATE APPLICATION NO. DATE ---------\_\_\_\_\_ JP 57198697 19821206 JP 1981-83222 A2 19810531 PΙ PRAI JP 1981-83222 19810531 Soldering resists contain 70-95 parts of phenolic resin-modified alkyd resins (1-10% phenolic resin, acid number ≥20) and 5-30 parts aminoplasts. Thus, a mixture of soya fatty acid 140, trimethylolpropane 147, isophthalic acid 166, and CKM 1282 30 g was heated at 170-210° until the acid number reached 30, cooled to 170°, thinned with 500 g BuO(CH2)2OH, cooled to 50°, and mixed with 130 g S 6831M (melamine-formaldehyde copolymer) [ 9003-08-1] to give a soldering-resist ink. A Cu foil-phenolic resin laminate was printed on the Cu side with the above ink and baked 15 min at 180°. H05K003-28; C08L067-08 IC ICI C08L067-08, C08L061-32 42-12 (Coatings, Inks, and Related Products) STsoldering resist ink; phenolic resin modified alkyd; melamine formaldehyde copolymer blend IT Soldering (resists, phenolic resin-modified alkyd resin and melamine -formaldehyde copolymers as) IΤ 77-99-6D, polymers with unsatd. fatty acids, isophthalic acid, and phenolic resins 121-91-5D, polymers with unsatd. fatty acids, triols, and phenolic resins 839-90-7D, polymers with unsatd. fatty acids, isophthalic acid, and phenolic resins RL: USES (Uses) (soldering resists, containing melamine-formaldehyde copolymer) L80 ANSWER 12 OF 32 HCAPLUS COPYRIGHT 2004 ACS on STN 1983:181226 HCAPLUS AN DN 98:181226 TIHeat-resistant baking enamel IN Adamec, Jozef; Drobny, Frantisek PΑ Czech.

```
Czech., 3 pp.
SO
     CODEN: CZXXA9
DT
     Patent
LA
     Slovak
FAN.CNT 1
                        KIND DATE
                                            APPLICATION NO.
                                                                      DATE
     PATENT NO.
                          ____
                                 -----
                                              -----
     CS 200964
                           В
                                             CS 1979-588
                                                                       19790126
PΙ
                                  19801031
PRAI CS 1979-588
                                  19790126
     The title baking enamel suitable for hot surfaces (heat-storage stoves,
     central heating elements) contains an alkyd resin prepared from glycerol,
     phthalic anhydride, and glycidyl esters of branched-chain fatty
     acids 30-80, formaldehyde-melamine resin (I)
     9003-08-1] 10-40, pigments 5-50, Ba linoleate (II) [24871-38-3]
     0.2-2, a 50% solution of Me Ph siloxane in ethylene glycol mono-Bu ether
     (III) [111-76-2] 0.1-1.0 parts, and optionally III 1.0-10.0, BuOAc
     0.5-8.0, and copolymers of acrylic acid, Bu acrylate,
     hydroxypropyl acrylate, and methylolacrylamide or of Me methacrylate and styrene 0.01-0.2 parts. A typical composition comprised an alkyd resin (acid number 3.2 mg KOH/g) 55.0, I (acid number 0.2 mg KOH/g) 20.0, Ti white 25.0,
     yellow Fe2O3 0.1, red Fe2O3 0.1, carbon black 0.1, an acrylic
     copolymer (acid number 32 mg KOH/g) 0.1, II 0.5, 50% Me Ph siloxane
     in III 0.1, and BuOAc 4.0 parts.
IC
     C09D003-64
     42-8 (Coatings, Inks, and Related Products)
     heat resistant baking enamel; alkyd resin heat resistant enamel;
ST
     melamine formaldehyde resin baking enamel; barium linoleate heat
     resistant enamel; siloxane heat resistant enamel; titanium oxide heat
     resistant enamel; iron oxide heat resistant enamel; carbon black heat
     resistant enamel
L80 ANSWER 13 OF 32 HCAPLUS COPYRIGHT 2004 ACS on STN
AN
     1983:17570 HCAPLUS
     98:17570
DN
     Pelletized aminoplast molding compositions
TI
     Matsushita Electric Works, Ltd., Japan
PA
     Jpn. Kokai Tokkyo Koho, 2 pp.
SO
     CODEN: JKXXAF
DT
     Patent
LA
     Japanese
FAN.CNT 1
     PATENT NO. KIND DATE APPLICATION NO.
                                                                     DATE
                                              _____
                                  19820731 JP 1981-8718
                          A2
     JP 57123224
                                                                       19810122
PRAI JP 1981-8718
                                 19810122
     Aminoplast pellets are prepared with good processability by adding 0.5-5%
     (based on resin) lubricants to the resin. Thus, a composition of
     melamine-formaldehyde copolymer [9003-08-1]
     (H2O content 10%) 60, phthalic acid 0.5, zinc stearate 1, coloring agent 1 part was kneaded, ground, extruded with 2 parts stearic acid (I) [
     57-11-4], and pelletized. The above pellets were molded 2 min at
     145^{\circ} and 100 \text{ kg/cm2} to give level colored moldings.
IC
     C08J003-12; C08L061-20
     37-6 (Plastics Manufacture and Processing)
CC
     melamine formaldehyde resin lubricant; stearic acid lubricant
ST
IT
     Lubricants
        (stearic acid, for melamine resin molding compds.)
IT
     57-11-4, uses and miscellaneous
     RL: USES (Uses)
```

(lubricants, for melamine resin molding compds.) L80 ANSWER 14 OF 32 HCAPLUS COPYRIGHT 2004 ACS on STN AN 1983:5505 HCAPLUS DN 98:5505 TI High-solids coating materials Nippon Oils & Fats Co., Ltd., Japan PΑ Jpn. Kokai Tokkyo Koho, 6 pp. CODEN: JKXXAF DT Patent LA Japanese FAN.CNT 1 DATE APPLICATION NO. KIND DATE PATENT NO. \_\_\_\_\_ \_\_\_\_ -----\_\_\_\_\_ JP 57123268 A2 19820731 JP 1981-8698 19810123 PRAI JP 1981-8698 19810123 High-solids coating materials contain an alkyd resin (number-average mol. 700-2000) 100, an acrylic polymer (number-average mol. weight 2000-5000, glass-transition temperature 0-50°) 20-100, and aminoplasts 30-100 parts. Thus, a composition of an 80% solids alkyd resin (derived from coco fatty acid 40, phthalic anhydride 28.3, pentaerythritol 17, and trimethylolpropane 14.7 parts) 247, a 60% copolymer [63747-55-7] solution (derived from styrene 10, Me methacrylate 53, Bu methacrylate 7, 2-ethylhexyl acrylate 22, 2-hydroxyethyl methacrylate 7, and acrylic acid 1 part) 83, a melamine-formaldehyde copolymer [9003-08-1] solution 151, TiO2 300, Solvesso 100 30, BuOH 10, and BuOCH2CH2OH 10 parts was milled, thinned, applied to a primed steel plate, left 10 min, and baked 30 min at 140° to form a coating having pencil hardness F, falling-dart (500 g, 0.5-in. diameter) impact strength 40 cm, Erichsen test indentation 3.0 mm, gloss retention after 24 mos of outdoor exposure in Florida 60%, and good acid and gasoline resistance. C09D003-66; C09D005-40 IC CC 42-8 (Coatings, Inks, and Related Products) L80 ANSWER 15 OF 32 HCAPLUS COPYRIGHT 2004 ACS on STN 1982:599656 HCAPLUS ΑN 97:199656 DN High bond strength phenol-modified polyester coating composition ΤI ΙN Thomas, Charles H. PA Westinghouse Electric Corp. , USA U.S., 5 pp. SO CODEN: USXXAM DT Patent LA English FAN.CNT 1 KIND DATE PATENT NO. APPLICATION NO. DATE -----' \_\_\_\_\_ ----\_\_\_\_\_ Α US 1981-226339 US 4347335 19810119 PΙ 19820831 PRAI US 1981-226339 19810119 comprise a polyester prepared from an aromatic dicarboxylic acid,

Water-reducible polyester compns. useful as elec. insulating coil coatings comprise a polyester prepared from an aromatic dicarboxylic acid, tris(hydroxyalkyl) isocyanurate, polyol, unsatd. fatty acid, and a tricarboxylic acid, the polyester being modified by formaldehyde-phenol copolymer (I) [9003-35-4] and a hexamethoxymethylmelamine resin (II) [27936-91-0].

Thus, a polyester was prepared from linoleic acid 535, trimethylolpropane 502, and isophthalic acid 4495 g. The mixture was heated to 176° and

then to 238° over 3 h to acid number 9.5. Then, at 145°, 147.5 g trimellitic anhydride was added and the mixture was heated 3 h at 168°. Then 486.8 g Bu cellosolve was added and the polyester [83670-94-4] was cooled. To 65 g polyester was added 13.7 g I and 24.6 g II. To the composition was added 24.1 g water and 6.9 g Bu cellosolve. mixture was neutralized to pH 9.2 with dimethylethanolamine. The modified polyester coating had good elec. properties. C09D003-52; C09D003-56; C09D003-66 IC NCL 524538000 CC 42-8 (Coatings, Inks, and Related Products) ST polyester phenolic melamine coating; elec insulation polyester ΙT 83670-94-4 RL: TEM (Technical or engineered material use); USES (Uses) (coatings, containing phenolic and melamine resins, elec. insulating) L80 ANSWER 16 OF 32 HCAPLUS COPYRIGHT 2004 ACS on STN 1982:584042 HCAPLUS AN DN 97:184042 ΤI Water-thinned metallic base coating materials Nippon Oils & Fats Co., Ltd., Japan PΑ Jpn. Kokai Tokkyo Koho, 6 pp. SO CODEN: JKXXAF DT Patent LA Japanese FAN.CNT 1 PATENT NO. KIND DATE APPLICATION NO. DATE \_\_\_\_\_ \_\_\_\_ JP 1980-165660 PΙ JP 57090063 A2 19820604 19801125 JP 63028113 B4 19880607 PRAI JP 1980-165660 19801125 Water-thinned metallic base coating materials are prepared by esterifying an allyl alc.-styrene copolymer(I) with monocarboxylic acids and then polycarboxylic acid, neutralizing with NH3 or amines, and dispersing in water with powdered metal. Thus, a mixture of I (weight-average mol. weight 1700) 38.8, dehydrated castor oil fatty acid 17.6, and xylene 1.6 parts was heated 2.5 h at 230°, cooled to 130°, mixed with 5.9 parts succinic anhydride, heated 30 min at 130°, cooled, and mixed with EtO(CH2)20H 15.1, iso-PrOH 15.1, and Et3N 5.8 parts to give a varnish. A composition containing Al paste 11.0, iso-PrOH 23.3, EtO(CH2)2OH 6, above varnish 53.5, and Sumimal [9003-08-1] 6.2 parts was stirred to give a coating material. The coating material was thinned, applied to a steel plate, and heated for 5 min at 80° to form a  $16-\mu$  coating. The plate was top coated with a clear acrylic resin composition and baked 20 min at 150° to form a 27- $\mu$  top layer. C09D005-38; C09D003-733 IC CC 42-7 (Coatings, Inks, and Related Products) metallic coating material; steel metallic coating material; allyl alc ST copolymer coating; styrene copolymer coating; fatty acid ester copolymer; succinic anhydride ester copolymer; amine salt copolymer coating; melamine resin crosslinking agent ΙT Crosslinking agents (melamine-formaldehyde copolymers, for water-thinned allyl alc.-styrene copolymer ester salt coatings containing metal powders) IT Fatty acids, esters

RL: USES (Uses) (vegetable-oil, esters with allyl alc.-styrene copolymers and polycarboxylic acids, amine salts, for water-thinned metallic coatings) IT Coating materials (water-thinned, allyl alc.-styrene copolymer ester amine salts, containing powdered aluminum and melamine resin) IT 85-44-9D, esters with allyl alc.-styrene copolymer and fatty acids, triethylamine salt 108-30-5D, esters with allyl alc.-styrene copolymers and fatty acids, triethylamine salt 552-30-7D, esters with allyl alc.-styrene copolymer and fatty acids, triethylamine RL: TEM (Technical or engineered material use); USES (Uses) (coatings, water-thinned, containing powdered aluminum and melamine resin) ΙT 25119-62-4D, esters with fatty acids and polycarboxylic acids, triethylamine salt RL: TEM (Technical or engineered material use); USES (Uses) (coatings, water-thinned, containing powdered aluminum and melamine resins) ΙT 9003-08-1 RL: MOA (Modifier or additive use); USES (Uses) (crosslinking agents, for water-thinned allyl alc.-styrene copolymer ester salt coatings containing metal powders) L80 ANSWER 17 OF 32 HCAPLUS COPYRIGHT 2004 ACS on STN 1981:482494 HCAPLUS AN 95:82494 DN Water-thinned coating materials TIShikoku Kaken Kogyo K. K., Japan PΑ Jpn. Kokai Tokkyo Koho, 5 pp. SO CODEN: JKXXAF DΤ Patent LA Japanese FAN.CNT 1 PATENT NO. KIND DATE APPLICATION NO. DATE \_\_\_\_ -----\_\_\_\_\_ \_\_\_\_\_ JP 1979-99754 JP 56024463 A2 19810309 19790804 PT JP 60036189 19850819 В4 PRAI JP 1979-99754 19790804 Compns. of water-soluble silicate salt solution 100, water-soluble polymer solution 0.5-50, divalent metal oxides 1-50, ammonium salts of C14-20 fatty acid 0.2-15, and silica-coated CaSO3 1-50 parts (solids) are useful as water-thinned coating materials. Thus, a composition of 31% solids Na silicate (Na20-SiO2 molar ratio 1:3) solution 100, 20% solids melamine-formaldehyde copolymer [9003-08-1] solution 30, ZnO 3, calcium oleate [142-17-6] 1, and silica-coated CaSO3 4 parts (solids) was applied to an asbestos slate to 300 g/m2 and dried 1 wk to form a coating having pencil hardness 6H and excellent resistance to alkali, chems., hot water, and soiling. IC C09D001-04; C04B041-32 42-7 (Coatings, Inks, and Related Products) CC ST sodium silicate coating; melamine formaldehyde copolymer coating; water thinned coating; zinc oxide coating additive; calcium sulfite coating additive; silica coated calcium sulfite; oleate calcium coating compn IT Coating materials (water-thinned, melamine-formaldehyde copolymer and

sodium silicate-based)

TΤ 142-17-6 1314-13-2, uses and miscellaneous 10257-55-3 RL: USES (Uses) (coatings containing, melamine-formaldehyde copolymer and sodium silicate-based water-thinned) ΙT 1344-09-8 RL: TEM (Technical or engineered material use); USES (Uses) (coatings, water-thinned, containing melamine-formaldehyde copolymers) L80 ANSWER 18 OF 32 HCAPLUS COPYRIGHT 2004 ACS on STN 1980:78277 HCAPLUS AN DN 92:78277 ΤI Quick-drying coating materials Matsuhira, Osamu; Fujishima, Minoru; Ebisawa, Katsuo ΙN PA Hitachi Chemical Co., Ltd., Japan SO Jpn. Kokai Tokkyo Koho, 4 pp. CODEN: JKXXAF DΤ Patent LA Japanese FAN.CNT 1 DATE APPLICATION NO. KIND DATE PATENT NO. DATE --------------\_\_\_\_\_ A2 B4 JP 1978-24168 JP 54129029 19791006 19780302 PΙ JP 55022516 19800617 PRAI JP 1978-24168 19780302 Compns. of 30-70 weight % urethane-modified alkyd resin from 80-95 parts oil-modified alkyd resin and 5-20 parts polyisocyanates and 30-70 weight % aminoplasts are useful as quick-drying coating materials. Thus, 90 parts soya fatty acid-modified phthalic anhydride-glycerol copolymer having acid number 5 and OH number 270 was treated in xylene with 10 parts hexamethylene diisocyanate at 100° for 4 h and diluted with xylene to 55% solids. A composition of 60 parts (solids) of the above solution and 40 parts formaldehyde-melamine-urea copolymer [25036-13-9] was heated at 100° and diluted with BuOH to 55% solids. The above solution and 10% of a 1:1 mixture of p-MeC6H4SO3H and MeOH were mixed, applied to a substrate, and dried 5 min at  $70^{\circ}$  to form a coating having good resistance to solvents, water, and stains. IC C09D003-66 42-9 (Coatings, Inks, and Related Products) CC L80 ANSWER 19 OF 32 HCAPLUS COPYRIGHT 2004 ACS on STN 1979:153563 HCAPLUS AN DN 90:153563 Thermosetting coating materials ΤI IN Mochizuki, Minoru; Ito, Yoshiyasu Toyo Ink Mfg. Co., Ltd., Japan PΑ SO Jpn. Kokai Tokkyo Koho, 3 pp. CODEN: JKXXAF DΤ Patent LA Japanese FAN.CNT 1 KIND DATE PATENT NO. APPLICATION NO. DATE --------------\_\_\_\_\_ PI JP 53140333 A2 19781207 PRAI JP 1977-39436 19770408 JP 1977-39436 19770408 19781207

AB Aminoplast-based thermosetting coating materials containing >10 weight% (solids)
resins having acid number ≥40 form high-quality coatings after rapid

resins having acid number ≥40 form high-quality coatings after rapid baking. Thus, a mixture of coconut oil **fatty acid** 12.6,

neopentyl glycol 19.7, isophthalic acid 3.4, and xylene 2 parts was heated at 200° until the acid number reached 8, 14.3 parts trimellitic anhydride added, and heated at 200° to give a polyester (I) having acid number 50. A mixture of BuOCH2CH2OH 25, Solvesso 100 25, Me methacrylate 21, Bu methacrylate 21, acrylic acid 2, N-butoxyacrylamide 6 , and Bz2O2  $\bar{1}$  part was heated at 100° to give a **copolymer** (II) [69735-78-0] solution A composition of I 40, II 30, and melamine -formaldehyde copolymer [9003-08-1] 30 parts (solids) was diluted with Solvesso 150 to 50% solids, applied to a tin plate and baked 5 min at 160° to give a coating having comparable qualities to those of a similar coating without I baked 10 min at 170°. IC C09D003-06 CC 42-9 (Coatings, Inks, and Related Products) ST aminoplast acrylic polyester coating; melamine formaldehyde copolymer coating; trimellitic anhydride polyester L80 ANSWER 20 OF 32 HCAPLUS COPYRIGHT 2004 ACS on STN 1979:139156 HCAPLUS ANDN 90:139156 TIAlkyd resin-based high-solids coating materials Sato, Tetsuya; Tawada, Hirohisa; Okazaki, Yasuhisa; Watanabe, Nobuyuki; IN Takai, Makoto; Onoda, Koji Miyoshi Oil and Fat Co., Ltd., Japan PA SO Jpn. Kokai Tokkyo Koho, 4 pp. CODEN: JKXXAF DT Patent LA Japanese FAN.CNT 1 PATENT NO. KIND DATE APPLICATION NO. DATE ----\_\_\_\_\_ -----A2 JP 53130748 19781115 JP 1977-44584 19770420 PRAI JP 1977-44584 19770420 Compns. of 50-90 parts alkyd resins having acid number 2-14, OH number 95-400, and solution viscosity (90% in xylene) 1-3 P at 25° and containing 3-7:3-7 mixts. of C12-22 linear fatty acid glycidyl ester and  $\alpha$ -alkyl C9-21 **fatty acid** glycidyl ester and 10-50 parts aminoplasts are useful as high-solids coatings. Thus, C12-15 fatty acid glycidyl ester 196, C12-15  $\alpha$ -branched fatty acid glycidyl ester 84, coconut oil fatty acid 200, trimethylolpropane 270, ethylene glycol 190, phthalic anhydride 300, and adipic acid 150 parts were heated in xylene to give a copolymer having acid number 7.5, OH number 247, and soln viscosity (90% in xylene) 2.15 P at 25°. A composition of the above copolymer 70, melamine-formaldehyde copolymer [ 9003-08-1] 30, and TiO2 100 parts (solids) was thinned with 1:1 xylene-BuOH to 81.5% solids to give a coating material which was applied to a steel plate and baked 20 min at 145° to form a coating having gloss 93.1 and 71.6% before and after 500 h of irradiation in a weatherometer, resp., pencil hardness H, and impact strength (500 g dart) 35 cm. IC C08L067-08 CC 42-9 (Coatings, Inks, and Related Products) alkyd high solid coating; fatty acid glycidyl ester; epoxy modified alkyl STresin; melamine resin blend coating ΙT Coconut oil RL: USES (Uses) (fatty acids, alkyd resins modified by, as high-solids coatings, containing fatty acid glycidyl esters and melamine resins) ΙT Fatty acids, esters RL: USES (Uses)

(glycidyl esters, alkyd resin coatings containing melamine resins and, high-solids)

IT Coating materials

(high-solids, alkyd resins, containing fatty acid glycidyl esters and melamine resins)

IT 556-52-5D, esters with linear and branched fatty acids

RL: USES (Uses)

(alkyd resin coatings containing melamine resins and, high-solids)

IT 40526-34-9D, esters with coconut oil fatty acids

RL: TEM (Technical or engineered material use); USES (Uses) (coatings, containing fatty acid glycidyl esters and melamine resins, high-solids)

L80 ANSWER 21 OF 32 HCAPLUS COPYRIGHT 2004 ACS on STN

AN 1978:425984 HCAPLUS

DN 89:25984

TI Permanent topical antistats

IN Stevens, Violete L.; Sexton, Arthur R.; Lalk, James W.; Deibel, Ronald D.; Corson, Fred P.

PA Dow Chemical Co., USA

SO U.S., 4 pp.

CODEN: USXXAM

DT Patent

LA English

FAN.CNT 2

L'UIA.	TAN.CNI Z						
	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE		
ΡI	US 4080161	Α	19780321	US 1976-678574	19760420		
	NL 7704015	Α	19771024	NL 1977-4015	19770413		
	FR 2348959	A1	19771118	FR 1977-11715	19770419		
	FR 2348959	B1	19800912				
	GB 1586520	Α	19810318	GB 1977-42150	19771010		
PRAI	US 1975-553713		19750227				
	US 1976-678574		19760420				
	US 1977-781826		19770328				

AB Antistatic compns. for polyamide and polyester fiber comprise a mixture of a polymer consisting of glycidol, ethylene oxide, and glycidyl stearate units with a mol. weight of <20,000, a formaldehyde-melamine copolymer (I) [9003-08-1], and an acid catalyst which promotes the condensation between the polymer and I. A typical polymer was prepared by polymerizing a mixture of ethylene oxide and tert-Bu glycidyl ether

(9:1 molar ratio) using ethylene glycol as the initiator and KOH or catalyst to give a polymer of mol. weight 1300 and the polymer was simultaneously dealkylated and esterified with stearic acid. When applied in conjunction with I and ZnBF4 to a polyamide carpet and cured the sample develop 6100 V compared to 12,300 for an untreated sample.

IC C08G023-22

NCL 008115600

CC 39-10 (Textiles)

ST antistatic compn polyamide polyester fiber; glycidyl stearate polymer antistatic; melamine formaldehyde polymer antistatic

IT Carpets

(antistatic agents for polyamide-based, ethylene oxide-glycidol-glyciyl stearate polymer-melamine-formaldehyde polymer mixts. as)

IT Polyamide fibers, uses and miscellaneous
Polyester fibers, uses and miscellaneous
RL: USES (Uses)

(antistatic agents for, ethylene oxide-glycidol-glyciyl stearate polymer-melamine-formaldehyde polymer mixts. as)

IT Antistatic agents

(ethylene oxide-glycidol-glycidyl stearate polymers and **melamine**-formaldehyde polymers, for polyamides and polyester fibers)

IT 57-11-4D, ester with dealkylated tert-Bu glycidyl ether-ethylene
 oxide copolymer 57301-20-9D, dealkylated, ester with stearic
 acid

RL: USES (Uses)

(antistatic agents, containing formaldehyde-melamine polymer, for polyester and polyamide fibers)

IT 57-11-4D, ester with dealkylated tert-Bu glycidyl ether-ethylene
 oxide copolymer

RL: USES (Uses)

(antistatic agents, containing formaldehyde-melamine polymer, for polyester and polyamide fibers)

RN 57-11-4 HCAPLUS

CN Octadecanoic acid (9CI) (CA INDEX NAME)

 $HO_2C^-$  (CH<sub>2</sub>)<sub>16</sub>-Me

L80 ANSWER 22 OF 32 HCAPLUS COPYRIGHT 2004 ACS on STN

AN 1978:407554 HCAPLUS

DN 89:7554

TI Antistatic treatment for textile materials

PA Dow Chemical Co., USA

SO Neth. Appl., 15 pp.

CODEN: NAXXAN

DT Patent

LA Dutch

FAN.CNT 2

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	NL 7704015	A	19771024	NL 1977-4015	19770413
	US 4080161 US 4125369	A A	19780321 19781114	US 1976-678574 US 1977-781826	19760420 19770328
PRAI	US 1976-678574 US 1977-781826		19760420 19770328		
	US 1975-553713		19750227		

- AB Antistatic compns. which can be applied to textile products such as rugs contain partially esterified hydroxylated polyethers of mol. weight <20,000, a primary OH-reactive hardener, and a catalyst. Thus, a random 9:1 ethylene oxide-tert-Bu glycidyl ether copolymer of mol. weight 5000 was prepared, dealkylated at 130° in the presence of p-toluenesulfonic acid [104-15-4], and esterified with stearic acid to give a product (I) with an average of 1 stearate group per mol. A nylon shag-plush carpet of weight 1020 g/m2 was treated with aqueous I containing 1.6%
  - p-toluenesulfonic acid and 17.5% melamine-formaldehyde polymer [9003-08-1] hardener at I add-on 6.8 g/m2 and cured 7 min at 150°, giving a carpet sample with charge at 20% relative humidity 1120 and soil resistance equal to that of an untreated control.
- IC C08L071-02
- CC 39-10 (Textiles)
- IT Carpets

Polyamide fibers, uses and miscellaneous RL: USES (Uses) (antistatic agents for, ethylene oxide-glycidol copolymer stearates as) IT Antistatic agents (ethylene oxide-glycidol copolymer stearates, for nylon carpet) 57-11-4D, esters with dealkylated butylglycidyl ether-ethylene ΙT oxide copolymer 57301-20-9D, dealkylated, stearate RL: USES (Uses) (antistatic agents, for nylon carpeting) 57-11-4D, esters with dealkylated butylglycidyl ether-ethylene IT oxide copolymer RL: USES (Uses) (antistatic agents, for nylon carpeting) 57-11-4 HCAPLUS RN Octadecanoic acid (9CI) (CA INDEX NAME) CN  $HO_2C^-$  (CH<sub>2</sub>)<sub>16</sub>-Me L80 ANSWER 23 OF 32 HCAPLUS COPYRIGHT 2004 ACS on STN 1977:537564 HCAPLUS AN 87:137564 DN ΤI Mass sizing of paper Taubitz, Chritof; Reichel, Fritz IN BASF A.-G., Fed. Rep. Ger. PA SO Ger. Offen., 19 pp. CODEN: GWXXBX DΤ Patent LA German FAN.CNT 1 KIND DATE APPLICATION NO. PATENT NO. DATE DE 2604547 A1 19770811 DE 1976-2604547
AU 7721497 A1 19780727 AU 1977-21497
SE 7701060 A 19770807 SE 1977-1060
FR 2340401 A1 19770902 FR 1977-3154
FR 2340401 B3 19791005
JP 52096209 A2 19770812 JP 1977-11710
DE 1976-2604547 19760206 PΤ 19770120 19770201 19770204 19770207 PRAI DE 1976-2604547 19760206 Mixts. of melamine resins and styrene (I) copolymer dispersions, obtained by polymerization of I with acrylates in the presence of K2S208 and sulfonates, were used for paper sizing. Thus, addition of 0.7%, butyl acrylate-2-ethylhexyl acrylate-styrene-vinyltoluene copolymer [64078-54-2] dispersion and 0.3% formaldehydemelamine-sodium o-aminobenzoate copolymer [ 40883-90-7] to spruce sulfite pulp gave paper (surface weight 80 g/mm2) with 19.3 Cobb value (DIN 53 132) and 12 min 50% ink penetration (DIN 53 126). IC D21H003-36 CC 43-7 (Cellulose, Lignin, Paper, and Other Wood Products) Section cross-reference(s): 36 ST styrene copolymer paper sizing; melamine copolymer paper sizing IT Sizing (of paper, with styrene copolymer dispersions containing

```
melamine resins)
IT
     Paper
        (sizing of, with styrene copolymer dispersions containing
        melamine resins)
IT
     Castor oil
     RL: USES (Uses)
        (sulfated emulsifier, for manufacture of styrene copolymer
        dispersions)
IT
     Emulsifying agents
        (sulfonates, for manufacture of styrene copolymer dispersions)
     112-80-1D, sulfonated
                              151-21-3, uses and miscellaneous
ΙT
     2386-53-0
                25155-30-0
                               27136-81-8
     RL: USES (Uses)
        (emulsifier, for manufacture of styrene copolymer dispersions)
ΙT
     51793-47-6
     RL: USES (Uses)
        (emulsifiers, containing sodium sulfate, for manufacture of styrene
        copolymer dispersions)
ΙT
     25153-46-2
                25767-47-9
                               26636-08-8
                                             64054-79-1
                                                           64078-54-2
     RL: USES (Uses)
        (paper sizing with, containing melamine resin)
IT
     38566-14-2
     RL: USES (Uses)
        (paper sizing with, containing stryrene copolymer dispersions)
TΤ
     40883-90-7
     RL: USES (Uses)
        (paper sizing with, containing styrene copolymer)
IT
     40883-89-4
     RL: USES (Uses)
        (paper sizing with, containing styrene copolymer dispersions)
     112-80-1D, sulfonated
ΙT
     RL: USES (Uses)
        (emulsifier, for manufacture of styrene copolymer dispersions)
RN
     112-80-1 HCAPLUS
     9-Octadecenoic acid (9Z)- (9CI) (CA INDEX NAME)
CN
Double bond geometry as shown.
HO<sub>2</sub>C (CH<sub>2</sub>) 7 _Z
                   /(CH2)7
     38566-14-2
TΥ
     RL: USES (Uses)
        (paper sizing with, containing stryrene copolymer dispersions)
RN
     38566-14-2 HCAPLUS
     Glycine, monosodium salt, polymer with formaldehyde and
CN
     1,3,5-triazine-2,4,6-triamine (9CI) (CA INDEX NAME)
     CM
          1
     CRN
         6000-44-8
     CMF C2 H5 N O2 . Na
```

Na

CM 2

CRN 108-78-1 CMF C3 H6 N6

CM 3

CRN 50-00-0 CMF C H2 O

H2C=0

IT 40883-90-7

RL: USES (Uses)

(paper sizing with, containing styrene copolymer)

RN 40883-90-7 HCAPLUS

CN Benzoic acid, 2-amino-, monosodium salt, polymer with formaldehyde and 1,3,5-triazine-2,4,6-triamine (9CI) (CA INDEX NAME)

CM 1

CRN 552-37-4

CMF C7 H7 N O2 . Na

Na

CM 2

CRN 108-78-1 CMF C3 H6 N6

CM 3

CRN 50-00-0 CMF C H2 O

н2С=0

IT 40883-89-4

RL: USES (Uses)

(paper sizing with, containing styrene copolymer dispersions)

RN 40883-89-4 HCAPLUS

CN Hexanoic acid, 6-amino-, monopotassium salt, polymer with formaldehyde and 1,3,5-triazine-2,4,6-triamine (9CI) (CA INDEX NAME)

CM 1

CRN 48047-10-5 CMF C6 H13 N O2 . K

 $H_2N-(CH_2)_5-CO_2H$ 

K

CM 2

CRN 108-78-1 CMF C3 H6 N6

CM

3

CRN 50-00-0 CMF C H2 O  $H_2C = 0$ L80 ANSWER 24 OF 32 HCAPLUS COPYRIGHT 2004 ACS on STN 1977:518643 HCAPLUS ΑN DN 87:118643 TI Surface treatment of melamine molding pieces ΑU Wirpsza, Zygmunt; Aleksandrowicz, Maria; Piechna, Jan CS Inst. Tworzyw Sztucznych Skory, Politech. Swietokrzyska, Radom, Pol. Polimery (Warsaw, Poland) (1977), 22(2), 47-9 SO CODEN: POLIA4; ISSN: 0032-2725 DTJournal LA Polish AB The treatment of melamine-formaldehyde copolymer (I) [ 9003-08-1] dishes with 4% melamine (II) [108-78-1] solution in 0.1% aqueous H3PO4 solution for 15 min at 100° decreased their content of free HCHO [50-00-0], which could be estimated with boiling water, by 1/3and decreased their discoloration by coffee and ink .apprx.9 times. addnl. treatment of I with 1% stearamide [124-26-5] solution in EtOH completely eliminated the staining with coffee. The concentration of free HCHO in I could be also reduced by treatments with o- or p-phenylene diamines, but they colored I; H2SO3 and (NH4)2SO4 were not effective. The discoloration with coffee could also be prevented by treating I with maleic anhydride [108-31-6], stearic acid [57-11-4] or Siltex 24 (an organic Si compound) solns.; oxalic acid was not effective. CC 36-6 (Plastics Manufacture and Processing) ST melamine resin coffee discoloration; surface treatment melamine resin; formaldehyde extractable melamine resin; staining prevention melamine dishes IT Dishes (from melamine resins, discoloration prevention and free formaldehyde extraction from) Siloxanes and Silicones IT RL: USES (Uses) (melamine resins treated with, discoloration resistance of) ΙT Discoloration prevention (of melamine resins, by coffee and ink) 108-31-6 IT 57-11-4 124-26-5 RL: USES (Uses) (melamine resins treated with, discoloration resistance of) ΙT 108-78-1 RL: USES (Uses) (melamine resins treated with, for removal of free formaldehyde and staining prevention) IT 50-00-0 RL: USES (Uses) (residues of, in melamine resins, discoloration in relation to)

L80 ANSWER 25 OF 32 HCAPLUS COPYRIGHT 2004 ACS on STN

```
1977:56896 HCAPLUS
ΑN
DN
      86:56896
TI
      Aqueous lacquer formulation
      Massy, Derek J. R.; Winterbottom, Kenneth
IN
PΑ
      Ciba-Geigy A.-G., Switz.
      Ger. Offen.
SO
      CODEN: GWXXBX
DT
      Patent
LA
      German
FAN.CNT 1
PATENT NO. KIND DATE APPLICATION NO. DATE

PI DE 2621078 A1 19761125 DE 1976-2621078 19760512

GB 1489109 A 19771019 GB 1975-20552 19750515

CA 1077638 A1 19800513 CA 1976-252410 19760513

FR 2311071 A1 19761210 FR 1976-14573 19760514

JP 51149323 A2 19761222 JP 1976-55983 19760515

PRAI GB 1975-20552 19750515
      Polymerizing dicarboxylic acids with glycols, fatty acids
      and mercaptocarboxylic acids gave polyesterthiols for use in coating of
      glass and steel plates. Thus, a mixture of 80% polyesterthiol (acid and SH
      number 0.73 and 0.78 val/kg, resp.) NH4 salt solution, from coconut oil
      fatty acid, glycerol [56-81-5], mercaptoacetic acid
      [68-11-1], pentaerythritol [115-77-5] and phthalic anhydride [85-44-9],
      and 75% formaldehyde-melamine copolymer [
      9003-08-1] solution in 100:25 ratio based on dry solids was diluted to
      65% solid content, applied on glass and galvanized steel plate at 75 µm
      film thickness, hardened for 20 min at 130°C, and conditioned
      overnight at 20°C and 65% relative humidity to give a specimen with
      277 s hardness and 12.5 mm flexure.
IC
      C09D003-66
CC
      42-9 (Coatings, Inks, and Related Products)
ST
      polyesterthiol aminoplast coating compn; melamine resin coating
      compn; glass polyesterthiol aminoplast coating; steel polyesterthiol
      aminoplast coating
L80 ANSWER 26 OF 32 HCAPLUS COPYRIGHT 2004 ACS on STN
AN
      1976:510211 HCAPLUS
DN
      85:110211
ΤI
     Coreaction products
      Yurcheshen, Michael; Levine, Morris; Brane, Ralph M.
IN
PΑ
      PPG Industries, Inc., USA
SO
      U.S., 8 pp. Division of U.S. 3,502,557.
      CODEN: USXXAM
DТ
      Patent
LA
     English
FAN.CNT 3
                       KIND DATE APPLICATION NO.
                                                                          DATE
      PATENT NO.
                                                  _____
     US 3965058 A 19760622 US 1969-854262 19690829
US 3502557 A 19700324 US 1965-444793 19650401
BE 678907 A 19661003 BE 1966-678907 19660401
US 1965-444793 19650401
PΙ
PRAI US 1965-444793
     Electrophoretic coatings with good throwing power contained a reaction
     product of an amine-aldehyde condensate with a polyfunctional material
      containing both hydroxyl and carboxyl groups, preferably
```

bis(hydroxyaryl)alkylidenecarboxylic acids, and contained optionally other resins. Thus, a mixture containing 1000 parts hexakis(methoxymethyl)

melamine and 400 parts ricinoleic acid was heated 8 hr at

```
290°F to give a product with acid number .apprx.91, which was heated 3
     hr at 200°F with 600 parts bisphenol A and 8 parts
     p-toluenesulfonic acid to give a copolymer [60279-89-2
     ] with acid number 24, that provided a hard, thermoset coating on a
     substrate.
TC
     C09D003-50
NCL
     260021000
     42-7 (Coatings, Inks, and Related Products)
CC
ST
     methoxymethylmelamine copolymer coating; ricinoleic
     acid copolymer coating; bisphenol A copolymer coating
ΙT
     Coating materials
        (electrophoretic thermosetting, from amine-aldehyde
        condensate-hydroxycarboxylic acid copolymers)
     1,3,5-Triazine-2,4,6-triamine, N,N,N',N',N'',N''-hexakis(methoxymethyl)-,
ΙT
        methylated, polymers with bis(hydroxyphenyl)pentanoic acid
     9-Octadecenoic acid, 12-hydroxy-, polymers with butylated
        formaldehyde-urea resin or hexakis (methoxymethyl) melamine
        -bisphenol A mixture, [R-(Z)]-
     Benzenebutanoic acid, 4-hydroxy-\gamma-(4-hydroxyphenyl)-\gamma-methyl-,
        polymers with methylated hexamethylolmelamine or butylated
        formaldehyde-urea resin
     Urea, polymer with formaldehyde, butylated, polymers with
        bis(hydroxyphenyl)pentanoic acid or ricinoleic acid
     RL: TEM (Technical or engineered material use); USES (Uses)
        (coatings, electrophoretic thermosetting)
ΙT
     31423-75-3
                  31423-76-4 31423-77-5 31533-93-4
                  60279-88-1 60279-89-2
     33395-69-6
     RL: TEM (Technical or engineered material use); USES (Uses)
        (coatings, electrophoretic thermosetting)
ΙT
     31423-77-5 31533-93-4 60279-89-2
     RL: TEM (Technical or engineered material use); USES (Uses)
        (coatings, electrophoretic thermosetting)
RN
     31423-77-5 HCAPLUS
     9-Octadecenoic acid, 12-hydroxy-, [R-(Z)]-, polymer with
CN
     N, N, N', N', N'', N''-hexakis (methoxymethyl)-1,3,5-triazine-2,4,6-triamine
           (CA INDEX NAME)
     CM
          1
     CRN
         3089-11-0
     CMF C15 H30 N6 O6
                 CH2-OMe
                     -СH2-ОМе
                    CH2-OMe
    MeO-CH2
```

KATHLEEN FULLER EIC 1700 REMSEN 4B28 571/272-2505

CM

2

CRN 141-22-0

TRUONG 10/642873 8/10/04 Page 118

CMF C18 H34 O3

Absolute stereochemistry. Rotation (-). Double bond geometry as shown.

RN 31533-93-4 HCAPLUS

CN Benzenebutanoic acid, 4-hydroxy- $\gamma$ -(4-hydroxyphenyl)- $\gamma$ -methyl-, polymer with N,N,N',N',N'',h''-hexakis(methoxymethyl)-1,3,5-triazine-2,4,6-triamine and [R-(Z)]-12-hydroxy-9-octadecenoic acid (9CI) (CA INDEX NAME)

CM 1

CRN 3089-11-0 CMF C15 H30 N6 O6

$$\begin{array}{c|c} \text{MeO-CH}_2 \\ \text{N-CH}_2\text{-OMe} \\ \\ \text{N-CH}_2\text{-OMe} \\ \\ \text{MeO-CH}_2\text{-N-N-CH}_2\text{-OMe} \\ \\ \text{MeO-CH}_2\text{-CH}_2\text{-OMe} \\ \end{array}$$

CM 2

CRN 141-22-0 CMF C18 H34 O3

Absolute stereochemistry. Rotation (-). Double bond geometry as shown.

CM 3

CRN 126-00-1 CMF C17 H18 O4

RN 60279-89-2 HCAPLUS

CN 9-Octadecenoic acid, 12-hydroxy-, [R-(Z)]-, polymer with N,N,N',N',N'',hexakis(methoxymethyl)-1,3,5-triazine-2,4,6-triamine and 4,4'-(1-methylethylidene)bis[phenol] (9CI) (CA INDEX NAME)

CM 1

CRN 3089-11-0 CMF C15 H30 N6 O6

$$\begin{array}{c|c} \text{MeO-CH}_2 \\ & \text{N-CH}_2\text{-OMe} \\ \\ & \text{N-CH}_2\text{-OMe} \\ \\ & \text{MeO-CH}_2\text{-N-N-CH}_2\text{-OMe} \\ \\ & \text{MeO-CH}_2 & \text{CH}_2\text{-OMe} \\ \end{array}$$

CM 2

CRN 141-22-0 CMF C18 H34 O3

Absolute stereochemistry. Rotation (-). Double bond geometry as shown.

$$HO_2C$$
 (CH<sub>2</sub>) 7 Z R (CH<sub>2</sub>) 5 Me

CM 3

CRN 80-05-7 CMF C15 H16 O2

```
L80 ANSWER 27 OF 32 HCAPLUS COPYRIGHT 2004 ACS on STN
     1976:75022 HCAPLUS
AN
     84:75022
DN
ΤI
     Alkali-soluble resins with improved heat resistance
IN
     Kato, Sadanobu
     Denki Kagaku Kogyo K. K., Japan
PA
     Jpn. Kokai Tokkyo Koho, 5 pp.
SO
     CODEN: JKXXAF
DT
     Patent
LA
     Japanese
     JP 50121240
FAN.CNT 1
                                         APPLICATION NO.
                                                                 DATE
                   A2 19750925
B4 19811006
19740128
                                           ______
                               -----
                               19750923 JP 1974-10979 19740128
     JP 50121349
JP 56042609
PΙ
PRAI JP 1974-10979
     An alkali-soluble resin with improved heat resistance is prepared by
     heat-treating a copolymer obtained by polymerizing a 80-97:3-20 vinyl
     fatty acid ester-unsatd. carboxylic acid or carboxylate
     ester mixture containing 0.1-5 parts (based on 100 parts mixture) aromatic
vinyl
     compound or diallyl compds. Thus, a mixture of H2O [containing 0.3 part poly(vinyl alc.)] 180, vinyl acetate 95, crotonic acid (I) 5, lauroyl
     peroxide 3.5, and N, N-diallylmelamine 0.5 part was polymerized at
     60-3° for 5 hr, filtered, and dried to give pearly
     copolymer [58152-81-1] which was heat-treated at
     85°, giving an alkali-soluble polymer with softening point 81°
     and m.p. 162°.
     CO8LJF; CO9D
IC
CC
     36-3 (Plastics Manufacture and Processing)
     alkali sol vinyl copolymer; heat resistant vinyl
ST
     copolymer; fatty acid ester copolymer; carboxylic acid
     unsatd copolymer; carboxylate ester unsatd copolymer
L80 ANSWER 28 OF 32 HCAPLUS COPYRIGHT 2004 ACS on STN
     1975:481424 HCAPLUS
AN
DN
     83:81424
     Acrylate coating compositions for cement boards
TΤ
     Yano, Naomichi; Koshizuka, Satoru; Tanaka, Satoshi; Takeuchi, Takao;
IN
     Yamada, Shunichi
    Kubota, Ltd., Japan
PΑ
     Jpn. Kokai Tokkyo Koho, 5 pp.
SO
     CODEN: JKXXAF
DT
     Patent
LA
     Japanese
FAN.CNT 3
                      KIND DATE APPLICATION NO.
     PATENT NO.
                                                                DATE
                               -----
                       ____
                                           _____
                        A2 19750327
A 19750120
                                          JP 1973-82219
SE 1974-9399
PΤ
     JP 50030934
                      A2
A
B
C
A
                                                                 19730719
    SE 7409399
                                                                 19740718
                              19810824
     SE 419748
                              19811203
    SE 419748
    SE 7808052
                              19780721
                                          SE 1978-8052
                                                                19780721
PRAI JP 1973-82218
                               19730719
    JP 1973-82219
                               19730719
     JP 1973-82220
                               19730719
     Coating compns. giving an alkali- and blocking-resistant film on cement
AB
```

building materials, such as asbestos cement boards, are prepared from a mixture of a copolymer (comprising C1-8 alkyl (meth)acrylate, a monomer containing carboxyl group, one or more of styrene, α-methylstyrene, vinyltoluene, vinyl acetate, vinyl propionate, vinyl fatty acid ester, one or more of monomers containing N-methylol, methylol ether, OH, or epoxy group) and a water-soluble amino resin. Thus, a mixture of water 60, nonionic surfactant (Emulgen) 3, anionic surfactant 0.2, and Na2S2O8 was mixed at 65° with a 0.1 part Na2S2O3, mixed dropwise with a monomer mixture of Me methacrylate 64, 2-ethylhexyl acrylate 32, and methacrylic acid 4 parts, mixed with 0.2 part Na2S2O8 and 0.2 part Na2S2O3, heated 2 hr at 65° to give a copolymer [25133-98-6] emulsion (42% solids), which (80 parts) was mixed 10% of a water-soluble melamine resin (Sumitex M3 [ 39429-81-7]), 85 parts water, 20 parts of an aqueous 2% poly(Na acrylate) solution, 0.1 part of a silicone defoaming agent, and 0.1 part formalin to give a coating composition The composition was coated on an asbestos cement board, and cured 10 hr at 180° and with 7 kg/cm2 steam to give a surface smooth and blocking-resistant coating film. 24(3)B825.1; 24(3)C9 42-7 (Coatings, Inks, and Related Products) Section cross-reference(s): 58 ST coating compn cement board; acrylate coating compn emulsion; melamine acrylate resin coating IT (asbestos blocks, methacrylate copolymer-melamine resin coatings on, alkali- and blocking-resistant) ΙT Building materials (asbestos-cement blocks, methacrylate copolymermelamine resin coatings on, alkali- and blocking-resistant) ΙT Asbestos RL: USES (Uses) (cement blocks, methacrylate copolymer-melamine resin coatings on, alkali- and blocking-resistant) IT Coating materials (methacrylate copolymers-melamine resins, on asbestos cement boards, alkali- and blocking-resistant) ΙT 9003-08-1 RL: TEM (Technical or engineered material use); USES (Uses) (coatings, containing methacrylate copolymer, on cement boards, alkali- and blocking-resistant) L80 ANSWER 29 OF 32 HCAPLUS COPYRIGHT 2004 ACS on STN 1974:97445 HCAPLUS ΑN DN 80:97445 ΤI Coating by electrophoresis IN Maeda, Hiroshi; Sunada, Masuyuki; Kanai, Yuji PΑ Dainippon Ink and Chemicals, Inc. Jpn. Tokkyo Koho, 3 pp. SO CODEN: JAXXAD DT Patent LAJapanese FAN.CNT 1 PATENT NO. KIND DATE APPLICATION NO. DATE \_\_\_\_\_ \_\_\_\_\_ \_\_\_\_ JP 48022831 PΙ В4 19730709 JP 1964-61865 19641102 PRAI JP 1964-61865 19641102 Coatings having good resistance to alkali, gasoline, impact, hot H2O, and corrosion are prepared by electrodeposition from an amine-substituted

condensate of an alkyd resin with a hydroxymethylated or alkoxymethylated aminoplast. Thus, an alkyd resin prepared from soybean oil fatty acids 350, glycerol [56-81-5] 220, phthalic anhydride [85-44-9] 420, and maleic anhydride [108-31-6] 20 parts is heated 8 hr at 50.deg. with 250 parts methoxymethylated melamine-formaldehyde copolymer [9003-08-1] and 500 parts BuOCH2CH2OH, diluted with H2O to 50% solids, neutralized to pH 8.5 with triethylamine [121-44-8], coated as a 10% solution on steel for 20 sec at 50 V, and baked 30 min at 150.deg. to give a hard, chemical resistant film. IC C23B; C09D 42-7 (Coatings, Inks, and Related Products) CC electrophoretic coating alkyd; melamine resin coating; ST triethylamine salt resin; chem resistance coating ΙT Chemically resistant materials (alkyd resin-melamine resin reaction product electrophoretic coatings) TΤ Coating materials (alkyd resin-melamine resin reaction products, chemical resistant) ΙT Coating process (electrophoretic, of alkyd resin-melamine resin reaction 1,2,3-Propanetriol, polymer with maleic anhydride, phthalic anhydride and ΙT soybean oil fatty acids, reaction products with melamine 1,3,5-Triazine-2,4,6-triamine, polymer with formaldehyde, methoxymethylated, reaction products with alkyd resins 1,3-Isobenzofurandione, polymer with glycerol, maleic anhydride and soybean oil fatty acids, reaction products with melamine resins 2,5-Furandione, polymer with glycerol, phthalic anhydride and soybean oil fatty acids, reaction products with melamine resins Ethanamine, N, N-diethyl-, with alkyd resin-melamine resin reaction products RL: USES (Uses) (electrophoretic coatings, chemical resistant) L80 ANSWER 30 OF 32 HCAPLUS COPYRIGHT 2004 ACS on STN 1972:516203 HCAPLUS AN 77:116203 DN TI Aminoplast resins particularly for electrocoating ΙN Sekmakas, Kazys De Soto, Inc. PΑ SO U.S., 5 pp. CODEN: USXXAM DT Patent LA English FAN.CNT 1 PATENT NO. KIND DATE APPLICATION NO. DATE \_\_\_\_\_ \_\_\_\_ -----US 3676312 Α 19720711 US 1970-19457 19700313 PRAI US 1970-19457 19700313 AB A unidirectional current flowing through an aqueous electrocoating bath containing polyhydric polyether or esterified styrene-allyl alc. copolymer (acid value 25.1-34.5) and benzoguanamine-paraformaldehyde copolymer [26160-89-4]-containing aminoplast resin (acid value 9.2-29.2), prepared in the presence of butanol [71-36-3]) caused codeposition of the resins on an elec. conductive base to give a

continuous non-blistering coating with good substrate adhesion, surface properties, and solvent and detergent resistance. Thus, benzoguanamine, (HCHO)n, BuOH, dimethylolpropionic acid, and C6H6 were heated with H2O and some BuOH distillation and removal. 2-Hydroxyethoxyethanol was added as a thinner to give a 66.8% solids product(I) with Gardner viscosity Z6-Z7 (200 P), and a nonvolatile acid value 29.2. A 78.2% nonvolatile esterified polyether resin (II) with Gardner viscosity X-Y, and nonvolatile acid value 25.1, was prepared from UCAR POLYOL DCP-200 (III) [36445-75-7], soya fatty acids, xylol, trimellitic anhydride, MeCOEt, Et3N, and butylene oxide. A I-II blend neutralized with Et3N and deposited from an aqueous coating bath, was cured adhesion glossy continuous film with 2H-3H hardness and good EtCOMe resistance and fair resistance to aqueous detergent, while a XM-1116 (methylated-ethylated melamine resin)-I coating blistered completely and lost substrate adhesion after immersion in aqueous detergent. B01K; C23K

IC

NCL 204181000

CC 42-10 (Coatings, Inks, and Related Products)

L80 ANSWER 31 OF 32 HCAPLUS COPYRIGHT 2004 ACS on STN

AN 1972:502786 HCAPLUS

DN 77:102786

ΤI Composition for obtaining cellular plastic material

IN White, Woodrow W.; Reynolds, James A.

PΑ Uniroyal, Inc.

SO Fr. Demande, 17 pp.

CODEN: FRXXBL

DT Patent

LA French

FAN.CNT 2

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
ΡI	FR 2081878	<b>A</b> 5	19711210	FR 1971-8325	19710310
	ZA 7101250	Α	19711124	ZA 1971-1250	19710226
	NL 7103027	A	19710914	NL 1971-3027	19710308
	BE 763991	A1	19710909	BE 1971-100676	19710309
	ES 389176	A1	19730601	ES 1971-389176	19710312
PRAI	US 1970-19090		19700312		

Aging resistant foams, useful as carpet pads and floor coverings, were AB prepared from a latex, e.g. acrylonitrile-butadiene copolymer (I) [9003-18-3], containing 3-15% reactive resin, e.g. a melamine resin. Thus, I latex modified with rosin and a fatty acid, was mixed with a phenolic antioxidant, (NH4)2HPO4,

dimethylolmelamine-trimethoxymethylolmelamine-

hexamethoxymethylolmelamine resin [36355-45-0], Al(OH)3,

silicone surfactant, Na lauryl sulfate, and water, the mixture having a pH 9.1 was pulverized on a heated fabric surface, and vulcanized at 149.deg. to give a detergent resistant foam. Addition of the melamine resin to the latex increased the tensile strength of the product and reduced its elongation.

IC B29D

CC 36-6 (Plastics Manufacture and Processing)

ST carpet pad foam; melamine resin latex foam; butadiene latex

foam; styrene latex foam; acrylonitrile latex foam

IT Rubber, butadiene-styrene, uses and miscellaneous Rubber, nitrile, uses and miscellaneous

(cellular, containing melamine-formaldehyde resins)

IT 9003-56-9

RL: USES (Uses)

(cellular, containing melamine-formaldehyde resins)

L80 ANSWER 32 OF 32 HCAPLUS COPYRIGHT 2004 ACS on STN

AN 1970:122865 HCAPLUS

DN 72:122865

TI Condensation products for dry impregnation of textile materials

IN Enders, Heinz; Singer, Heinrich; Stenzinger, Theodor

PA Chemische Fabrik Pfersee G.m.b.H.

SO Brit., 4 pp.

CODEN: BRXXAA

DT Patent

LA English

FAN.CNT 1

AB Compns. were prepared for imparting H2O repellency to textile materials. Thus, tetramethylolmelamine 246, phthalic anhydride 148, octadecyl alc. 27 0, stearic acid 142 g, and MeOH 1320 ml were slowly heated 1 hr and t he MeOH distilled off. The mixture was stirred 30 min at 160°, cooled to 120°, and a 50% solution of the resulting condensate prepared by adding C2Cl4. This solution was stable and could be diluted with C2Cl4 to a slightly opalescent solution Textiles treated with this solution had good water-repellent properties after evaporation of the solvent

and drying. Dimethylolurea, propanol, C2Cl3H, paraffin wax, maleic anhydride, dodecanol, palmitic acid, penta-methylolmelamine, EtOH, succinic anhydride, hexadecyl alc., cerotic acid, and methylolethyleneurea were similarly used.

IC C08G

CC 39 (Textiles)

IT Textiles

(waterproofing of, hydroxymethyl nitrogenous compds. copolymers for)

IT 27555-52-8 27555-53-9 27555-54-0 **27989-51-1** 

28213-49-2

RL: USES (Uses)

(for waterproofing of textiles)

IT 27989-51-1 28213-49-2

RL: USES (Uses)

(for waterproofing of textiles)

RN 27989-51-1 HCAPLUS

CN Maleic anhydride, polymer with dodecyl alcohol, palmitic acid and (s-triazine-2,4,6-triyltriimino)trimethanol (8CI) (CA INDEX NAME)

CM 1

CRN 1017-56-7 CMF C6 H12 N6 O3

CM 2

CRN 112-53-8 CMF C12 H26 O

 ${\rm HO^-}$  (CH<sub>2</sub>)<sub>11</sub>-Me

CM 3

CRN 108-31-6 CMF C4 H2 O3

CM 4

CRN 57-10-3 CMF C16 H32 O2

 $HO_2C^-$  (CH<sub>2</sub>)<sub>14</sub>-Me

RN 28213-49-2 HCAPLUS

CN Phthalic anhydride, polymer with [[6-[bis(hydroxymethyl)amino]-s-triazine-2,4-diyl]diimino]dimethanol, 1-octadecanol and stearic acid (8CI) (CA INDEX NAME)

CM 1

CRN 17600-13-4 CMF C7 H14 N6 O4

HO- 
$$CH_2$$
-  $NH$ 

N-  $CH_2$ -  $OH$ 

N-  $CH_2$ -  $OH$ 

N-  $CH_2$ -  $OH$ 

NH-  $CH_2$ -  $OH$ 

CM 2

CRN 112-92-5 CMF C18 H38 O  $HO-(CH_2)_{17}-Me$ 

CM 3

CRN 85-44-9 CMF C8 H4 O3

CM 4

CRN 57-11-4 CMF C18 H36 O2

 ${\rm HO_2C^-}$  (CH<sub>2</sub>)<sub>16</sub>-Me

=>